



GLOBAL
MERCURY
PARTNERSHIP

Mercury from Non-Ferrous Metals Mining and Smelting: current knowledge and tools towards sustainable practices

Webinar, Thursday 20 March 2024



Opening remarks and scene setting

Rodges Ankrah

U.S. Environmental Protection Agency, Co-Chair of the UNEP Global Mercury Partnership Advisory Group

For the smooth running of the webinar, please :



Keep **microphones off** unless when making an intervention, cameras are optional.



Use the “**Chat**” to ask technical questions or share views.



The **webinar will be recorded**. Please indicate if you have any objection.

AGENDA

The event will be facilitated by [Kenneth Davis](#), Chemicals and Health Branch, UNEP, and [Misuzu Asari](#), Co-lead of the Waste Management Area

Opening remarks and scene setting, [Rodges Ankrah](#), U.S. Environmental Protection Agency, Co-Chair of the UNEP Global Mercury Partnership Advisory Group

- **Study report on mercury from non-ferrous metals mining and smelting**, [Peter Nelson](#), Lead author of the Technical Study report on Mercury from the Non-ferrous Metals Mining and Smelting, Macquarie University, Co-lead of the Partnership Area on Mercury Releases from Coal Combustion
- **Latest Developments in the context of the Minamata Convention**, [Eisaku Toda](#), Secretariat of the Minamata Convention
- **Mercury management in zinc smelting**, [Mic Gilles](#), International Zinc Association

Questions and Answers

- **Environmentally sound management of mercury from the mining and smelting of non-ferrous metals in Argentina**, [Marisol Diaz Rivera](#), National Coordinator of Hazardous Waste in the Undersecretary of Environmental, Ministry of Interior
- **Treatment of mercury wastes and stabilization of mercury from the non-ferrous metal industry**, [David Hunter](#), BATREC Industrie AG
- **Treating mercury waste from non-ferrous metals mining and smelting: best available treatment technologies and mercury conversion as the safest option for final disposal**, [Xavier Ibarz](#), Econ Industries

Questions and Answers

Next steps and closing remarks, [Rodges Ankrah](#), U.S. Environmental Protection Agency, Co-Chair of the UNEP Global Mercury Partnership Advisory Group



Part 1

facilitated by [Kenneth Davis](#), Chemicals and Health Branch, UNEP

STUDY REPORT ON MERCURY FROM NON-FERROUS METALS MINING AND SMELTING

**WEBINAR: Mercury from Non-Ferrous Metals Mining and Smelting: current knowledge
and tools towards sustainable practices**

Thursday 21 March

Peter Nelson
Professor Emeritus of Environmental Studies
Macquarie University

Co-lead UN Environment Mercury in Coal Combustion Partnership

peter.nelson@mq.edu.au

Work on cross-cutting themes

- Mercury from oil and gas and from non-ferrous metals
- Decision by Partnership Advisory Group in November 2019 (PAG-10)
- Expert consultations launched in April 2020
- Study report published online following comments from experts and relevant stakeholders



Report on mercury in non-ferrous metals mining and smelting

- Supported by UN Environment through a Small Scale Funding Agreement (SSFA) with Macquarie University
- Timeline:
 - Draft version of outline circulated for comment 30th November
 - Annotated outline discussed at PAG-11
 - First draft of report 15 January
 - Review GMP secretariat, co-chairs and interested Partnership Area leads 29 January
 - Second draft mid March
 - Expert Consultation of 2nd Draft 30 April 2021
 - Final draft
 - Circulation of final report with call for final comments November 2021
 - Final report appears online 2022

Study report on mercury from non-ferrous metals

Aim: to provide a better understanding of the mercury mass balance globally between

- supply,
 - storage, and
 - waste treatment related to non-ferrous metals mining and smelting operations
-
- Be concise
 - Benefit from global experience

Methodology of Report

- expert consultations,
- open access sources of information, including
 - published reports and toolkits,
 - websites and
 - the peer reviewed scientific research literature
- presents
 - a critical review of existing knowledge and information gaps concerning mercury from the non-ferrous sector;
 - a showcase of the different methods currently in use for reducing mercury emissions and releases and disposing of mercury from mining and smelting at different key stages of the processes; and
 - suggestions for further work including capacity development.

Methodology

- Lead author: Macquarie University (Australia)
- Consultations of experts and relevant stakeholders (online meetings, call for input and comments)
- Open access sources
- Emphasis on copper, lead, zinc and large-scale gold

The report presents:

- Review of existing knowledge and information gaps
- Showcase the different reduction methods
- Suggestions for further work including capacity development

Study report - overview

Executive summary – key highlights

1. Introduction

2. Existing activities related to mercury in the non-ferrous sector

2.1. National and regional regulations and guidelines

2.2. International agreements, partnerships and guidance

2.3. Industry sector activities

3. Life cycle of mercury in non-ferrous metals mining and smelting

3.1. Indicative processes

3.2. Quantitative studies and data

Study report – overview (*continued*)

4. Mercury emissions and releases estimates from the non-ferrous sector – existing knowledge and knowledge gaps

- 4.1. Background on previous work on estimation of mercury emissions and releases and mercury release mechanisms and speciation
- 4.2. Methodology for estimating emissions and releases from the non-ferrous sector – key inputs and knowledge gaps
- 4.3. The impact of global trade in concentrates on the location of mercury emissions and releases

5. Control of mercury emissions and releases

- 5.1. Process descriptions
- 5.2. Emission control techniques
- 5.3. Smelter/ roasting waste and by-product management and releases to land and water
- 5.4. Case studies prepared for BAT/BEP guidance

Study report – overview (*continued*)

6. Major conclusions and identified needs for further investigation, research and cooperation
7. Guidance on best practices
8. References and bibliography
9. Attachment: case studies

Draft study report – key findings

Large source of mercury releases to the environment:

- 3rd largest source of emissions to air – about 10 to 15 % of total emissions, estimated at more than 300 tonnes per year (but industry estimates significantly lower and this difference requires resolution)
- 2nd largest source of release to water – about 40% of total releases, estimated at more than 200 tonnes per year, mostly from large scale gold production
- Releases to land and waste production estimated to be much larger, but include secure impoundments such as controlled tailings piles or engineered landfills; however large uncertainties remain

Large source of global mercury supply, estimated at around 500 tonnes per year, about 15% of total – need to strengthen data

Sector is likely to grow considerably over the next 30 years

Draft study report – key findings

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Draft study report – key findings (*continued*)

Identified uncertainties and knowledge gaps :

- Hg content in ores and concentrates, at plant and country level
- Hg air emissions test data
- Hg concentrations in reject material
- Hg distributions between emissions and other releases
- Activity data (amounts of ores and concentrates processed)
- Effects of pollution control technologies, incl. on distribution of Hg between emissions to air, and capture in solid and liquid waste
- Additional quantitative information on how mercury deports to emissions and releases to air, land, water, waste and by-products

Draft study report – key findings (*continued*)

Identified needs :

- Effective and sustainable regional solutions for Hg secure and long-term storage
- Research and development on improvements to mineral processing
- Further investigations / information-sharing on Hg fate during mining and smelting of copper, lead and zinc and large-scale gold production
- Sharing of best practices and case studies, including Hg removal systems, off-gas cleaning systems in smelters, risks mitigation measures as well as options for environmentally sound interim storage and disposal of mercury.

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DISCUSSION

(ABC News: file photo)





MINAMATA
CONVENTION
ON MERCURY

Latest Developments in the context of the Minamata Convention

Global Mercury Partnership: Mercury from Non-Ferrous Metals Mining and Smelting:
current knowledge and tools towards sustainable practices

21 Mar 2024

Fifth meeting of the Conference of the Parties



Fifth meeting of the Conference of the Parties to the Minamata Convention on Mercury (COP-5)

Geneva, Switzerland, 30 Oct 2023 - 03 Nov 2023



More than 800 participants and 115 Parties represented

21 decisions adopted

- ❖ 5/1: The effects of mercury pollution on Indigenous Peoples and on local communities
- ❖ 5/2: Mercury supply sources and trade
- ❖ 5/3: Study of the global supply, trade and use of mercury compounds
- ❖ 5/4: Amendments to annexes A and B
- ❖ 5/5: Preparation of a report on cosmetics listed in part I of annex A to the Minamata Convention on Mercury
- ❖ 5/6: Information on the Economic and Technical Feasibility of Mercury-Free Catalysts in VCM Production
- ❖ 5/7: Artisanal and small-scale gold mining
- ❖ 5/8: Mercury emissions
- ❖ 5/9: Guidance on BAT/BEP to control releases
- ❖ 5/10: Mercury waste thresholds
- ❖ 5/11: Review of the financial mechanism
- ❖ 5/12: Capacity building, technical assistance and technology transfer
- ❖ 5/13: National reporting
- ❖ 5/14: First effectiveness evaluation of the Minamata Convention on Mercury
- ❖ 5/15: Gender action plan
- ❖ 5/16: Knowledge management
- ❖ 5/17: Contribution of the Minamata Convention to the Kunming-Montreal Global Biodiversity Framework
- ❖ 5/18: Enhanced international cooperation and coordination
- ❖ 5/19: Cooperation between the secretariat of the Minamata Convention on Mercury and the BRS secretariat
- ❖ 5/20: Programme of work and budget for 2024-2025
- ❖ 5/21: Dates and venue of COP-6

Compilation of decisions available from [Convention website](#).

[Follow-up letter](#) sent to Parties and stakeholders.



Minamata Convention Controls the whole life cycle of mercury

Life cycle of mercury

Article 3: Supply Trade

- By-product mercury from Non-Ferrous sector to be identified as supply source

Guidance on identifying mercury stocks (COP-1)
 Guidance and format for import consent (COP-1)



Article 4: Products Article 5: Processes Article 7: ASGM

NAP guidance (COP-1)



Article 8: Emissions Article 9: Releases

- Smelting and roasting processes in Pb, Zn, Cu and Au production is listed as a point source category.
- Requirement for control of emission to air and emission inventory.

BAT/BEP & inventory guidance (COP-1)
 Inventory guidance (COP-4)
 BAT/BEP guidance (COP-5)



Article 10: Storage

Interim storage guidelines (COP-2)

- By-product mercury needs to be stored properly

- Metals production may be identified as relevant point source of mercury releases to land and water
- Requirement for release control and inventory

Article 12: Contaminated sites

Guidance on management (COP-3)

- Contaminated sites related to metals production need to be addressed

- Mercury waste from metals production needs to be managed in an environmentally sound manner

Article 11: Waste

Thresholds (COP-3, 4 and 5)
 Basel Convention guidelines

Minamata Convention Article 11

2. For the purposes of this Convention, mercury wastes means substances or objects:

- (a) **Consisting of mercury or mercury compounds;**
- (b) **Containing mercury or mercury compounds;** or
- (c) **Contaminated with mercury or mercury compounds,**

in a quantity above **the relevant thresholds** defined by the COP, in collaboration with the relevant bodies of the Basel Convention in a harmonized manner, that are disposed of or are intended to be disposed of or are required to be disposed of by the provisions of national law or this Convention.

COP Decisions 3/5 and 5/10

No threshold needs to be established for waste consisting of mercury, and waste listed in Table 1 shall be regarded as such mercury waste.

No threshold needs to be established for waste containing mercury, and mercury-added products that are disposed of, are intended to be disposed of or are required to be disposed of, including those listed in Table 2, will be regarded as such mercury waste;

COP established **15 mg/kg total concentration of mercury as the threshold for waste contaminated with mercury,** allowing Parties to use alternative approaches to define such waste.

Table 1: List of mercury waste consisting of mercury or mercury compounds

- Recovered elemental mercury
- Elemental mercury
- Mercury (I) chloride and mercury (II) chloride
- Mercury (II) oxide (mercuric oxide)
- Mercury (II) sulfate (mercuric sulfate)
- Mercury (II) nitrate (mercuric nitrate)
- Cinnabar concentrate
- Mercury sulfide

Table 2: Non-exhaustive list of waste containing mercury or mercury compounds

- Non-electronic measuring devices containing mercury (barometers, hygrometers, manometers, thermometers, sphygmomanometers)
- Electrical and electronic switches, contacts, relays and rotating electrical connectors with mercury
- Fluorescent bulbs, high intensity discharge (HID) bulbs (mercury vapour bulbs, metal halide and high-pressure sodium bulbs), neon/argon lamps
- Batteries/accumulators containing mercury
- Biocides and pesticides containing mercury and their formulations and products
- Paints and varnishes containing mercury
- Pharmaceuticals containing mercury for human and veterinary uses, including vaccines
- Cosmetics and related products containing mercury
- Dental amalgam
- Scientific instrument used for the calibration of medical or scientific devices containing mercury

Minamata Convention Article 11

2. For the purposes of this Convention, mercury wastes means substances or objects....

This definition excludes overburden, waste rock and tailings from mining, except from primary mercury mining, unless they contain mercury or mercury compounds above thresholds defined by the Conference of the Parties



COP Decisions 3/5, 4/6 and 5/10

No need to develop thresholds for overburden and waste rock at the time of Decision MC-3/5.

COP established the following two-tier thresholds:

- (i) Tier-1 threshold to be applied first: 25 mg/kg total mercury content;
- (ii) Tier-2 threshold to be applied to tailings above the tier-1 threshold: 0.15 mg/L in the leachate, using an appropriate test method simulating the leaching of mercury at the site where the tailings are deposited.

COP invited Parties to use the guidance document [UNEP/MC/COP.5/INF/13](#) on the test methods for the tier-2 threshold.

UNEP/MC/COP.5/INF/13: Guidance document on test methods to be used for the tier-2 threshold for tailings from mining other than primary mercury

Key considerations in selecting or defining tier-2 test methods

Liquid/Solid ratio

As the L/S ratio affects liquid-solid equilibrium over the residence time of eluent in a leach test, the L/S ratio of the chosen test should not differ $\pm 20\%$ from a ratio of 10:1.

Test pH

As no in-situ pH control exists at sites where tailings are deposited, it is recommended to use an eluant at neutral pH (such as deionized water) without pH control (buffering).

Test duration

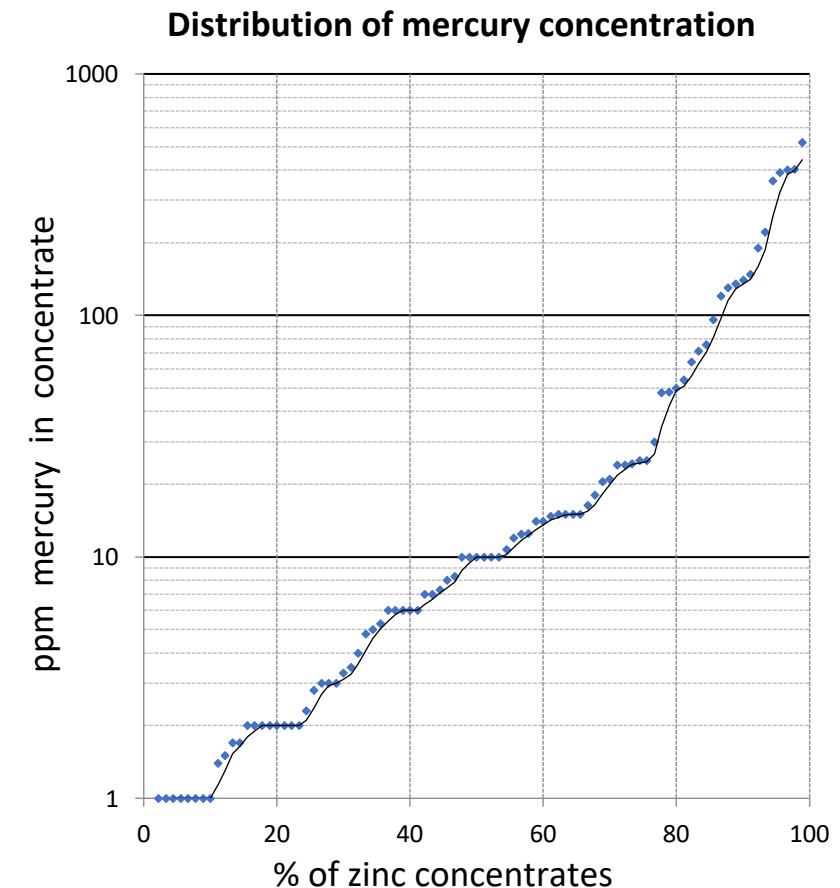
As oxidation, dissolution and interaction of "fresh" minerals in tailings are known to be slow enough to approximate initial near-equilibrium conditions, and weathering processes control release from tailings over the longer term, the duration of the chosen test is not to exceed 24-hours.

Mercury management in zinc smelting

- General considerations on emission control in zinc smelting
- Environmental performance of EU zinc smelters

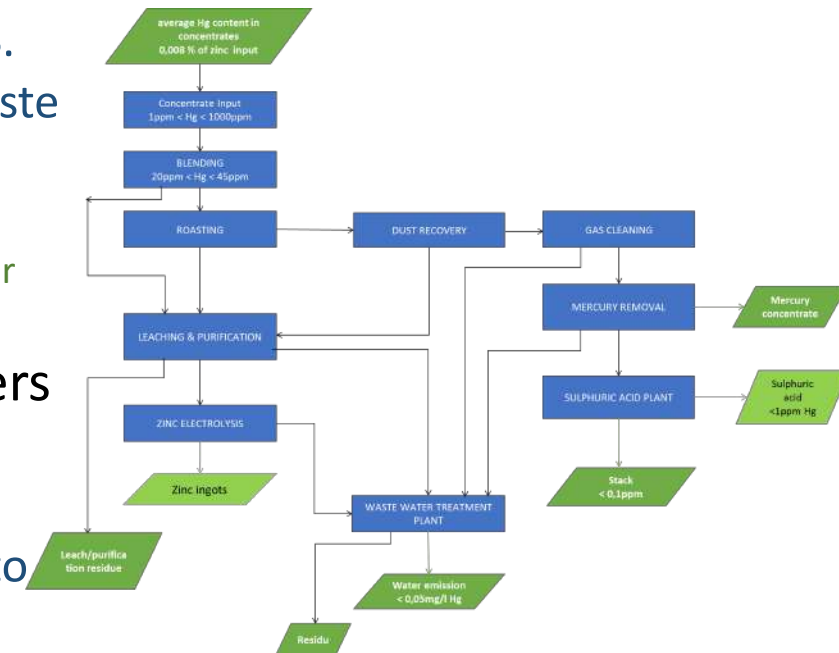
Zinc mineral concentrates in zinc refining:

- Mercury is found in small concentrations in zinc mineral deposits
 - Zn concentrates (+/-50% Zn) can contain a.o. up to 0,1% Hg
 - 50% of Zn concentrates have less than 10 ppm Hg
 - Average value 40 ppm Hg
- Zinc refineries use mixed concentrate input.
 - Most zinc mines produce too little concentrates to run a smelter
 - Composition of many zinc concentrates is not ideal to process in a zinc smelter
 - Smelters use a mix of mineral concentrates and secondaries to ensure a stable composition and controllable behaviour in the process.



Mercury removal in zinc smelting:

- Coolers and scrubbers remove +/- 50% of the mercury
 - Output is a water flow which is treated to precipitate Hg, typically as HgS.
 - Precipitated sludge from wastewater treatment is stored in approved waste storage facilities.
 - Final emissions to water are very low (<0,05% of input)
 - ✓ Precipitation of Hg as HgS allows removal down to very low concentrations of Hg in water (<1µg/l, often <0,1µg/l)
- Additionally, the Boliden-Norzink process is applied by most smelters which captures another 50% of the mercury
 - Output is a calomel sludge
 - Can be refined to metallic mercury or stabilized and disposed according to regulations.
- Other techniques to capture mercury exist but far less common.
- Stack emissions of mercury typically <1ppm (<0,25% of total Hg input)



=> UNEP toolbox was updated to reflect evolution in zinc smelting

Mercury: disposal or sales?

- Commercialized mercury originating from zinc smelters is steadily decreasing
- Several reasons
 - Legislative restriction to sell or export mercury (e.g. in EU)
 - Falling demand due to restrictions in use or development of alternatives for mercury
 - ✓ Hg gas discharge lights replaced by LED lights
 - ✓ Dental amalgam replaced by ceramics
 - ✓ Replacement in instrument like temperature and pressure measurement and switches
 - Big uncertainty about downstream sustainable use
 - ✓ Companies become more hesitant to commercialize mercury due to uncertainty on its final use.
 - Balancing between legal (un)certainty of acceptable mercury disposal conditions and control over downstream use when commercialized
 - Sustainability of operations has moved high on the agenda of zinc refiners.

Mercury disposal in the EU

- Most mercury from EU zinc smelters' BNZ filters is stabilized and environmentally safely stored according to EU waste regulation
 - Stabilized as very stable compound: sulphide or sulpho-selenide
 - Stabilisation often done by specialized third parties
 - Packed in sealed drums
 - Mostly stored in decommissioned potash mine caverns at K&S Germany but some in granite rock caverns.
- Precipitated mercury from waste water treatment is stored in sealed ponds.
 - Storage conditions aligned with regulation, ensuring appropriate containment and avoiding releases to the environment (groundwater, air)

Mercury emissions and releases in the EU

- Releases to water (E-PRTR): 6,35kg/y
 - Releases of most plants below the reporting threshold of 1kg/y.
 - Total release by all EU zinc smelter: <15kg/y (including from smelters below reporting threshold)
- Releases to air (E-PRTR): E-PRTR: 58kg/year
 - Releases of most plants below the reporting threshold of 10kg/y.
 - Total EU emissions to air from zinc smelting <100 kg/year (including from smelters below reporting threshold)

Note: Refined zinc output from EU + Norway represents 1/6 of world zinc output.

Global outlook

- The Minamata Convention has brought the environmental hazard of mercury to the attention of regulators worldwide.
 - Increasing number of zinc smelters outside EU have installed mercury filters
 - Implementation of sulphide precipitation in waste water treatment has become more standard practice
- The status in the EU demonstrates what is achievable:
 - When average environmental performance relative the mercury of all zinc smelters worldwide reaches the same performance as EU zinc smelters, the worldwide mercury releases to the environment would be limited to:
 - < 90 kg/y to water
 - < 600 kg/y to air

Thank you for your attention

Mik Gilles
International Zinc Association
mgilles@zinc.org



Questions and Answers

facilitated by [Kenneth Davis](#), Chemicals and Health Branch, UNEP



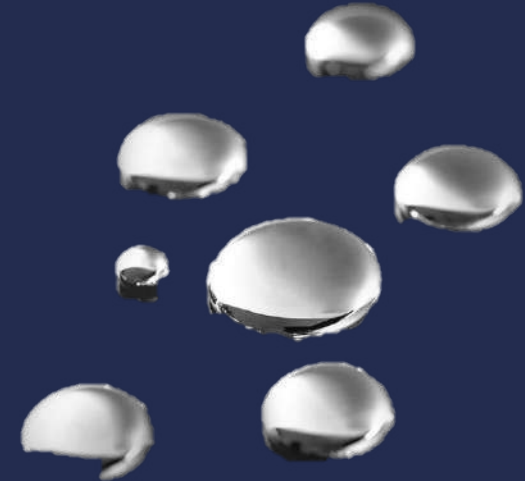
Part 2

facilitated by [Misuzu Asari](#), Co-lead of the Waste Management Area

Environmentally sound management of mercury from the mining and smelting of non-ferrous metals in Argentina

Webinar on Mercury from Non-ferrous Metals Mining and Smelting

Thursday 21 March 2024



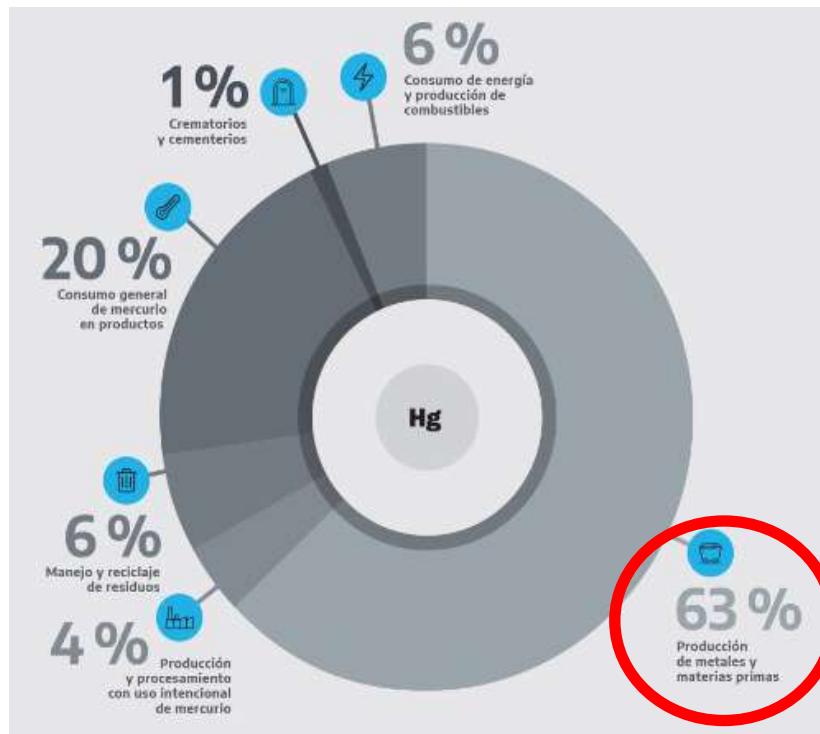
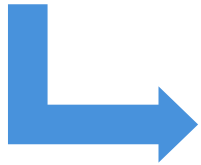
Secretaría de Turismo,
Ambiente y Deportes
Ministerio del Interior

Subsecretaría
de Ambiente

*Coordinación de Residuos
Peligrosos*

*Coordination of Hazardous
waste*

Initial Assessments of the Minamata Convention



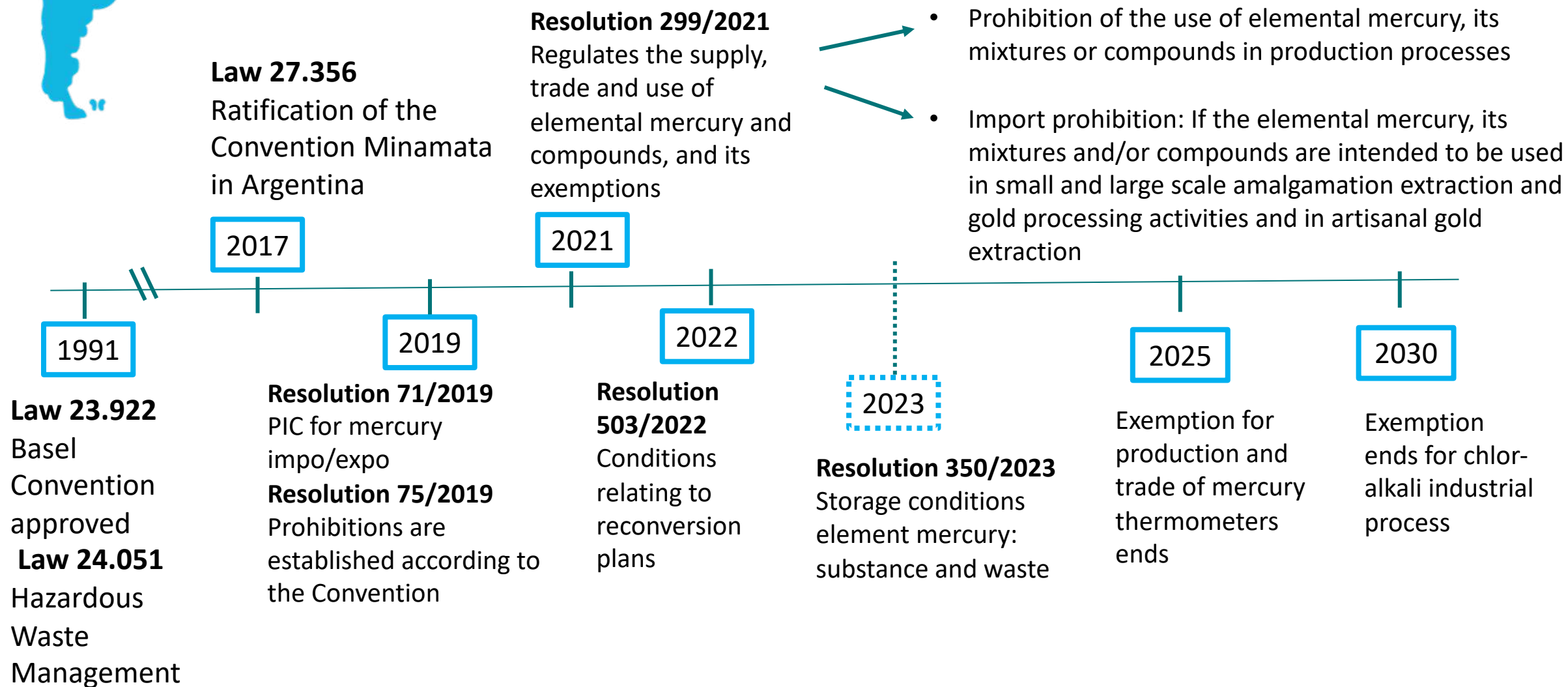
- 60% to Byproducts and impurities (mainly from gold extraction by methods other than amalgamation)
- 15% to waste that will receive treatment that includes the sector, but is not the only one.



- Argentina does not register primary production of mercury.
- Large-scale gold production is distributed in 12 mining sites and the presence of mercury was detected in 2 of them.

Environmentally sound management of mercury

Argentina regulations





Resolution 350/2023:

- The minimum technical requirements for the storage of mercury are established, which must be met by those subjects who have said material in their possession.

- Some requirements:
- IDENTIFICATION OF MERCURY

- 1.1. Identification of the substance:

- 1.2.4. Labeling or Label:

Mercurio N° CAS 7439-97-6 Masa molar 200,59 g/mol		PELIGRO H330 – Mortal en caso de inhalación. H360 – Puede dañar al feto. H372 – Provoca daños en los órganos tras exposiciones prolongadas o repetidas (sistema nervioso). H410 – Muy tóxico para los organismos acuáticos con efectos nocivos duraderos.
Código de Identificación: Lote:		P203 – Procurarse, leer y aplicar todas las instrucciones de seguridad antes de uso. P260 – No respirar polvos/humos/gases/neblinas/vapores/aerosoles. P271 – Utilizar solo al aire libre o en lugar bien ventilado. P280 – Usar guantes / ropa de protección / equipo de protección para los ojos / cara. P284 – En caso de ventilación insuficiente llevar equipo de protección respiratoria.
Peso neto: Fecha de envasado:		P270 – No comer, beber o fumar mientras se manipula este producto. P264 – Lavarse la piel cuidadosamente después de la manipulación. P304+P340 – EN CASO DE INHALACIÓN: Transportar a la persona al aire libre y mantenerla en una posición que le facilite la respiración. P319 – Buscar ayuda médica si la persona no se encuentra bien. P318 – En caso de exposición demostrada o supuesta consultar a un médico. P403+P405+P233 – Almacenar en un lugar bien ventilado, bajo llave. Mantener el recipiente herméticamente cerrado. P273 – No dispersar en el medio ambiente. P391 – Recoger los vertidos. P501 – Eliminar el contenido / recipiente conforme a la reglamentación vigente.
Nombre de la compañía: Domicilio: Teléfono: Página web:		

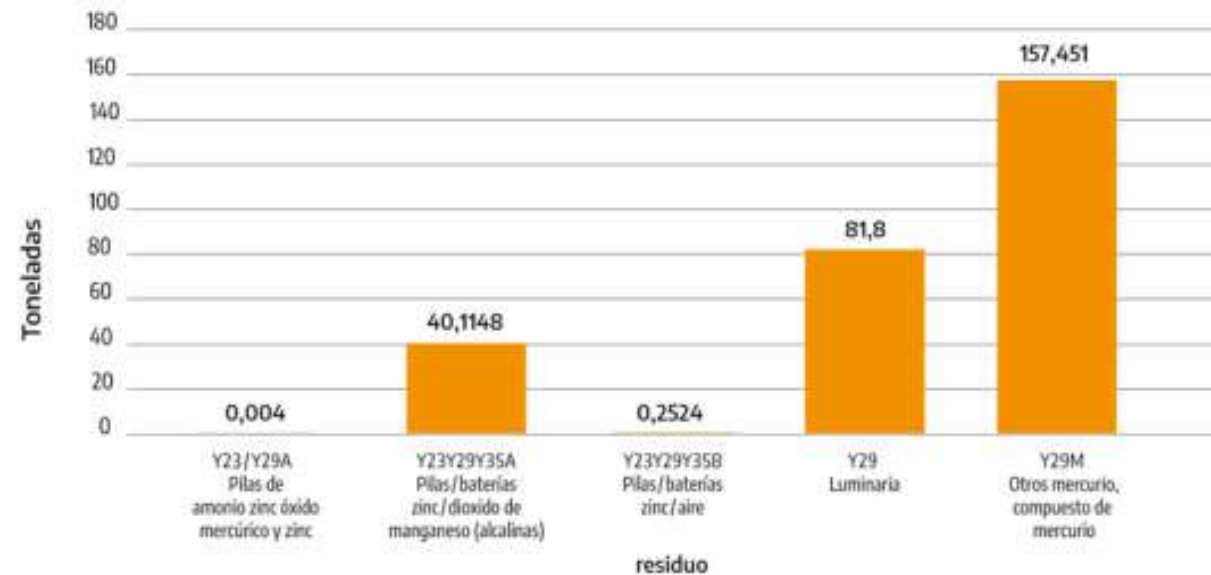




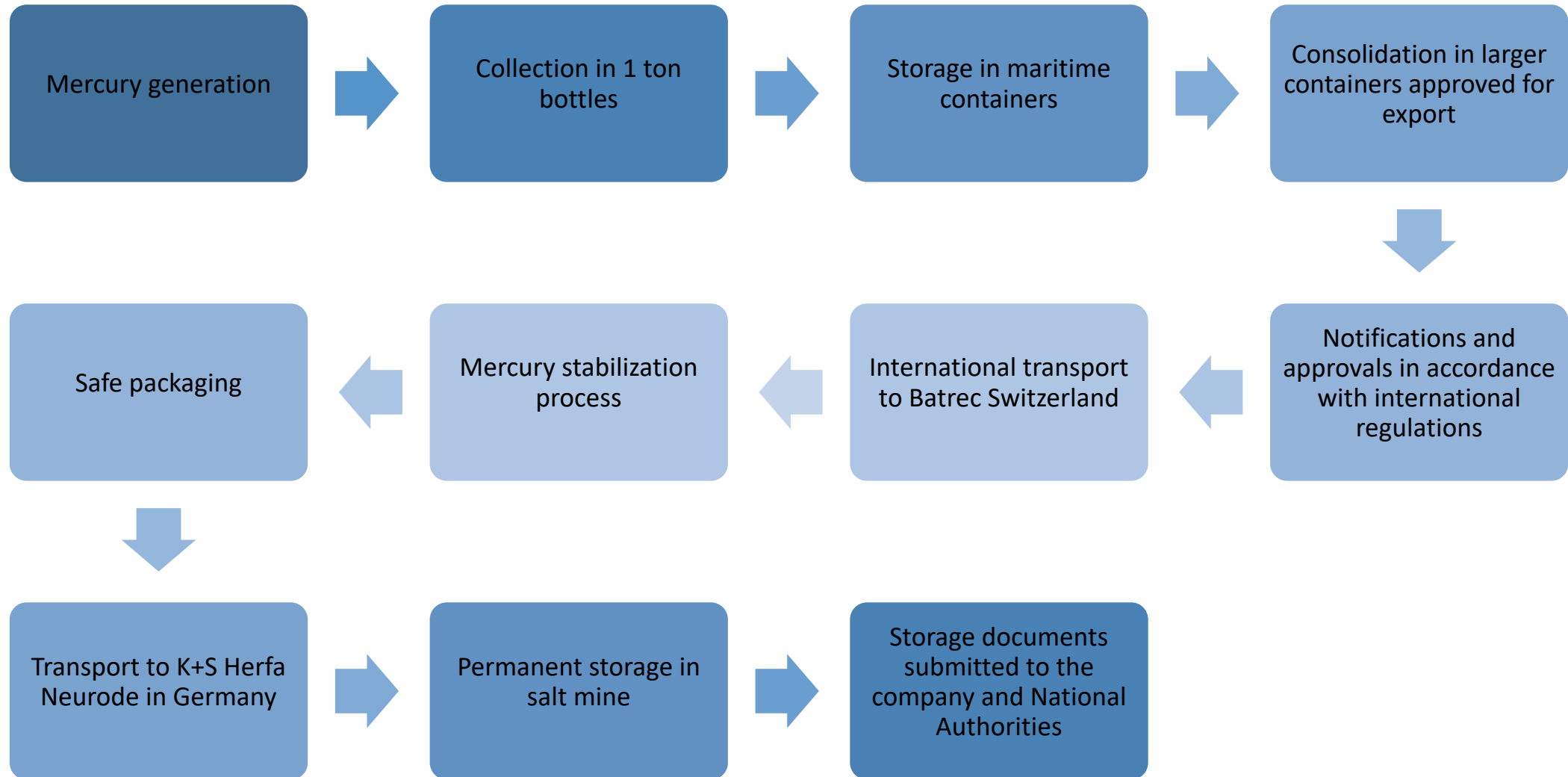
Environmentally sound management of mercury waste



- Within the framework of Law 24051 there is only one mercury compound waste operator, located in the province of Buenos Aires.
- There are currently no operators authorized to treat waste elemental mercury and other mercury compounds.
- For its management, it is exported to countries with said treatment and final disposal capacity.
- This process of cross-border movement is regulated by the Basel Convention, hence the synergy process with the Minamata Convention.
- Regarding waste generated in 2022, around 279.62 tons have been treated.



Export process from generation: example of gold mining





Agua



Suelo



Biodiversidad



Bosques



Áreas protegidas



Educación y Participación



Residuos



Riesgo



Cambio climático

agroquímicos obsoletos (Y4) tratados

Cantidad de residuos de mercurio tratados

Cantidad de residuos de pilas y baterías tratados

Cantidad de residuos hidrocarburos, emulsiones y aceites vegetales tratados

Cantidad de residuos peligrosos tratados en interjurisdicción, por operación de eliminación

Eliminación de residuos contaminados con PCBs

Exportación de Residuos Peligrosos

Indicador 12.4.1 Nivel de cumplimiento de los compromisos y obligaciones ante los Acuerdos Multilaterales de productos químicos y desechos de los cuales Argentina es Parte

Operaciones de acuerdo al Anexo III de la Ley

Residuos / Residuos peligrosos / Cantidad de residuos de mercurio tratados

TABLA

GRÁFICO

METADATOS








tipo de residuo tratado	descripción de residuo	toneladas
Y23/Y29A	Pilas de amonio zinc óxido mercúrico y zinc.	0.00
Y23Y29Y35A	Pilas/baterías zinc/dióxido de manganeso (alcalinas)	40.11
Y23Y29Y35B	Pilas/baterías zinc/aire	0.25
Y29	Luminaria	81.8
Y29M	Otros mercurio, compuesto de mercurio	157.45





Strengthening analytical capacity

a - Espectrofotómetro de absorción atómica	b - Digestor para metales DEENA II	c - UPS 10KVA para equipo espectrofotómetro de absorción atómica	d - Lámparas de cátodo hueco para medir Hg y AS
			
e - Instalación de mesa base y sistema de aspiración para equipo para equipo espectrofotómetro de absorción atómica	f - Adecuación de cañería de gases y reguladores para Argón	g - Estándares de Hg y As	a - \$ 3.087.549,00 UNSAM b - \$ 2.474.800,00 UNDAV c - 292.560,00 UNSAM d - 220.548,00 UNDAV e - \$ 245.530,00 UNSAM f - \$ 130.000,00 UNSAM g - \$ 59.764,30 UNSAM Y UNDAV
			

Two laboratories from public national universities were equipped, which collaborate with the Ministry. This investment makes it possible to improve the quality of environmental data produced in the academic field and put it at the service of management and decision-making for public policies. The total amount invested was more than 63 thousand dollars. The objective is to carry out environmental monitoring, including pilot projects developed and national monitoring plans. The first monitoring was carried out in a gold extraction mine in Argentina.

Registry of Emissions and Transfer of Pollutants (RETC)

It is a key tool for pollution prevention and access to environmental information.

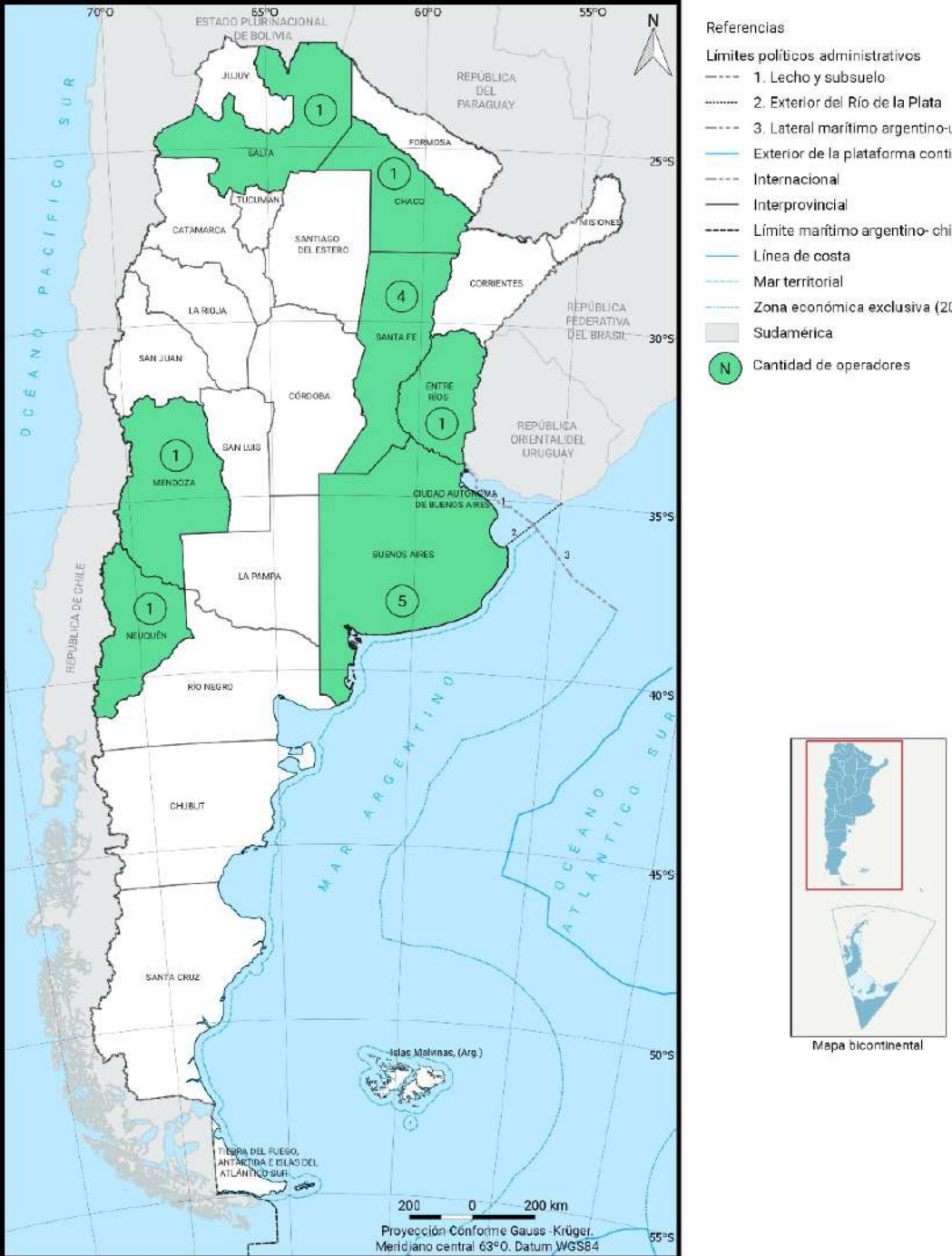




Challenge and needs in managing mercury

In particular from this Sector:

1. Mercury waste management
 - Treatment of elemental mercury waste and compounds
 - Treatment technologies / technology transfer
2. Reduction of emissions and releases and waste generation.
3. Data collection on emissions and releases
4. Articulation and coordination with local authorities.



¡Thank you!

Marisol Diaz Rivera
National Coordinator of Hazardous Waste
Argentina



Official website: chemicals and
hazardous waste in Argentina



**Secretaría de Turismo,
Ambiente y Deportes**
Ministerio del Interior

**Subsecretaría
de Ambiente**

*Coordinación de Residuos
Peligrosos*

*Coordination of Hazardous
waste*

Global Mercury Partnership Waste Management Area

Batrec Industrie AG

Treatment of mercury wastes and stabilisation of mercury
from the Non-Ferrous Metal industry

David Hunter 21st March 2024 david.hunter@batrec.ch +41 7 93 83 53 12

SARPI in brief – European hazardous waste network of Veolia

OUR MISSION

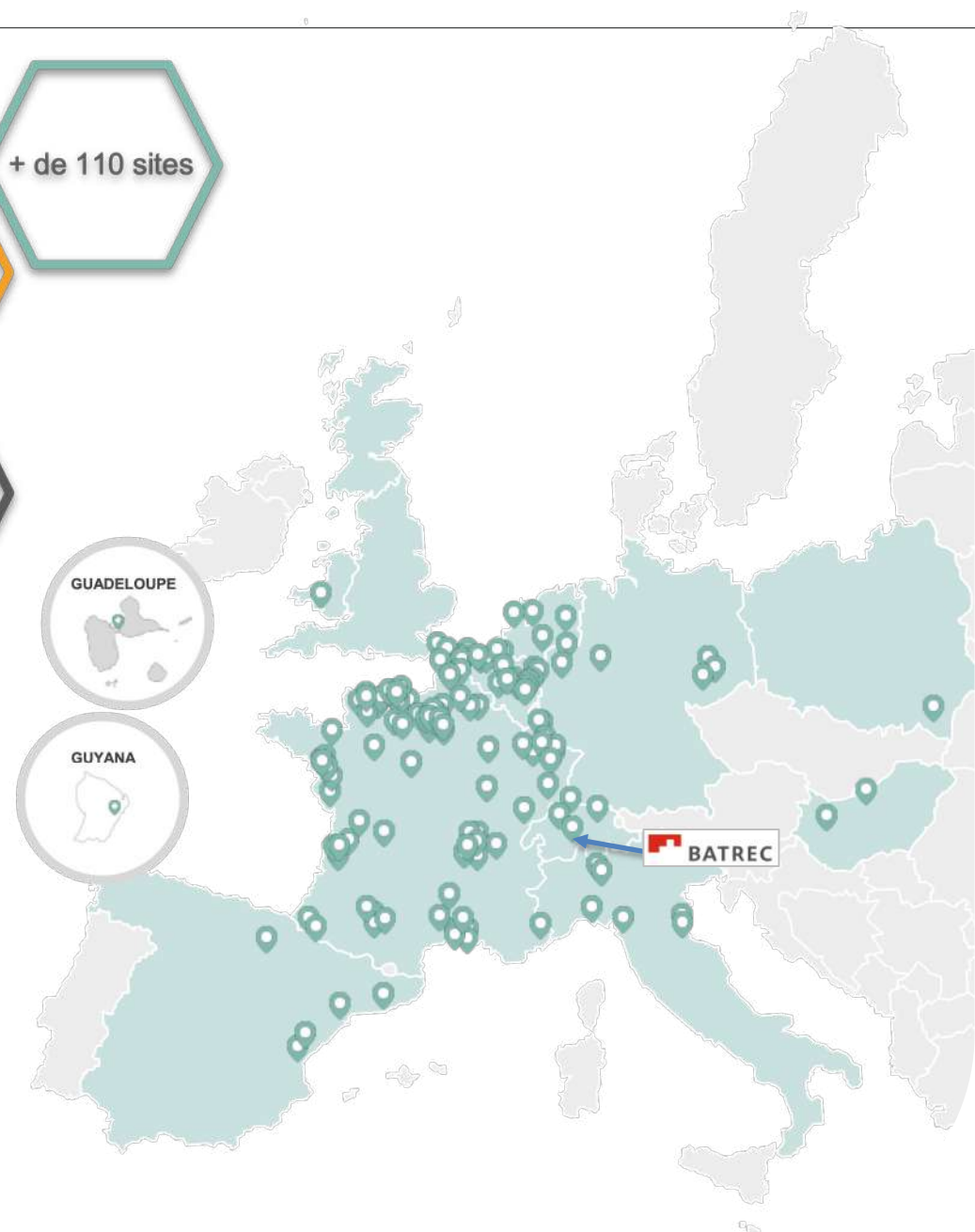
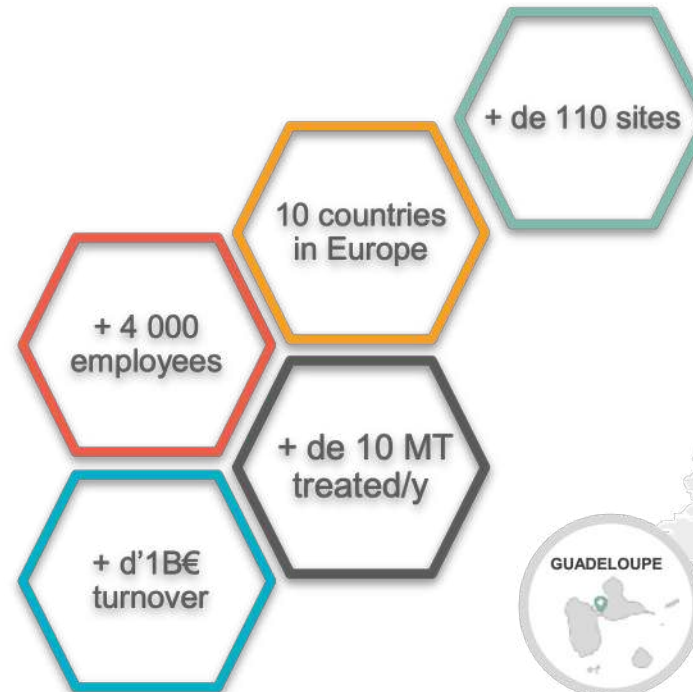
“ To support industrial and territorial development through sustainable risk management of waste.”

OUR JOB

“ Treatment and recovery of hazardous waste and polluted sites.”

OUR MODEL

“ A european network of facilities and services integrating the entire treatment and recovery chain including final containment of residual waste.”





Wimmis



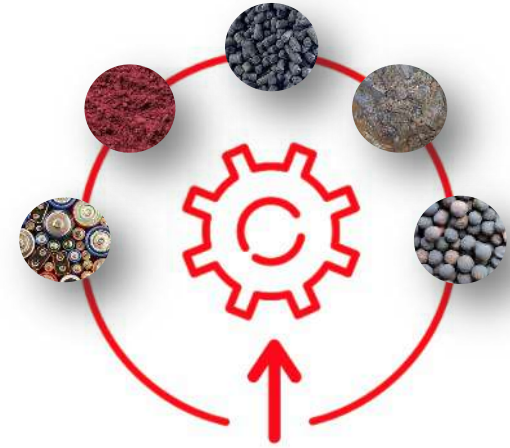
1991

FUNDATION



ISO 9001
ISO 14001
OSHAS 18001

CERTIFICATIONS



Liquid Mercury Stabilisation
Activated Carbon reactivation
Mercury wastes treatment
Mercury adsorbents recycling
Battery recycling

5 CORE SPECIALITIES



Wimmis



Mercury Waste Treatment – Batrecs Capabilities

able to treat any type of waste containing Mercury

Distil Hg-containing waste → Transform highly toxic Hg to HgS → Permanent Storage of HgS



Mercury Waste Treatment – Batrecs Capabilities

able to treat any type of waste containing Mercury

Distil Hg-containing waste → Transform highly toxic Hg to HgS → Permanent Storage of HgS



Wastes from Non-Ferrous Metal Mining/Smelting Industry



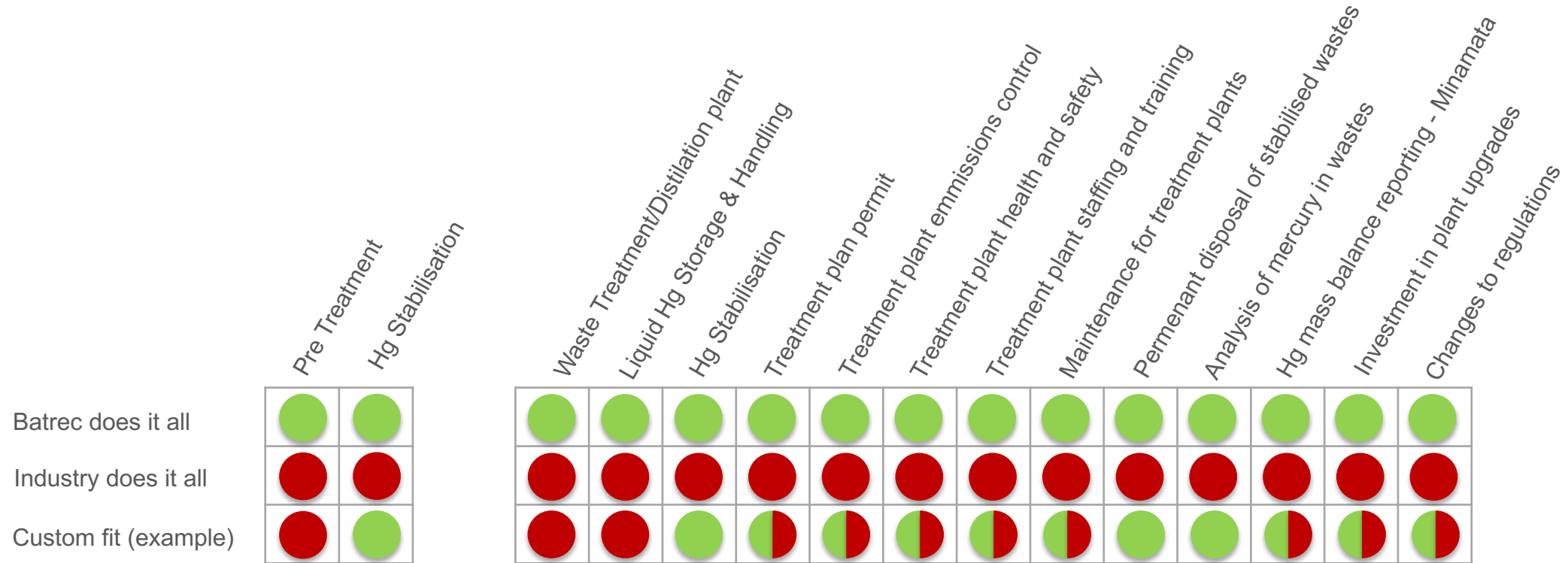
How can Batrec remain so versatile?



Options – Cost and liability of treating mercury waste

Who does what:

Who retains Costs & Liability for:



● = Non-Ferrous Metal Industry

● = Batrec, dedicated mercury waste treatment plant

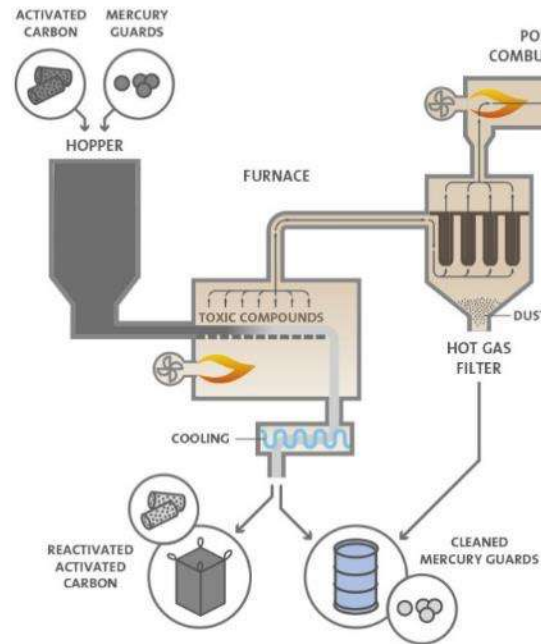
Treatment of Mercury Wastes



1

THERMAL TREATMENT

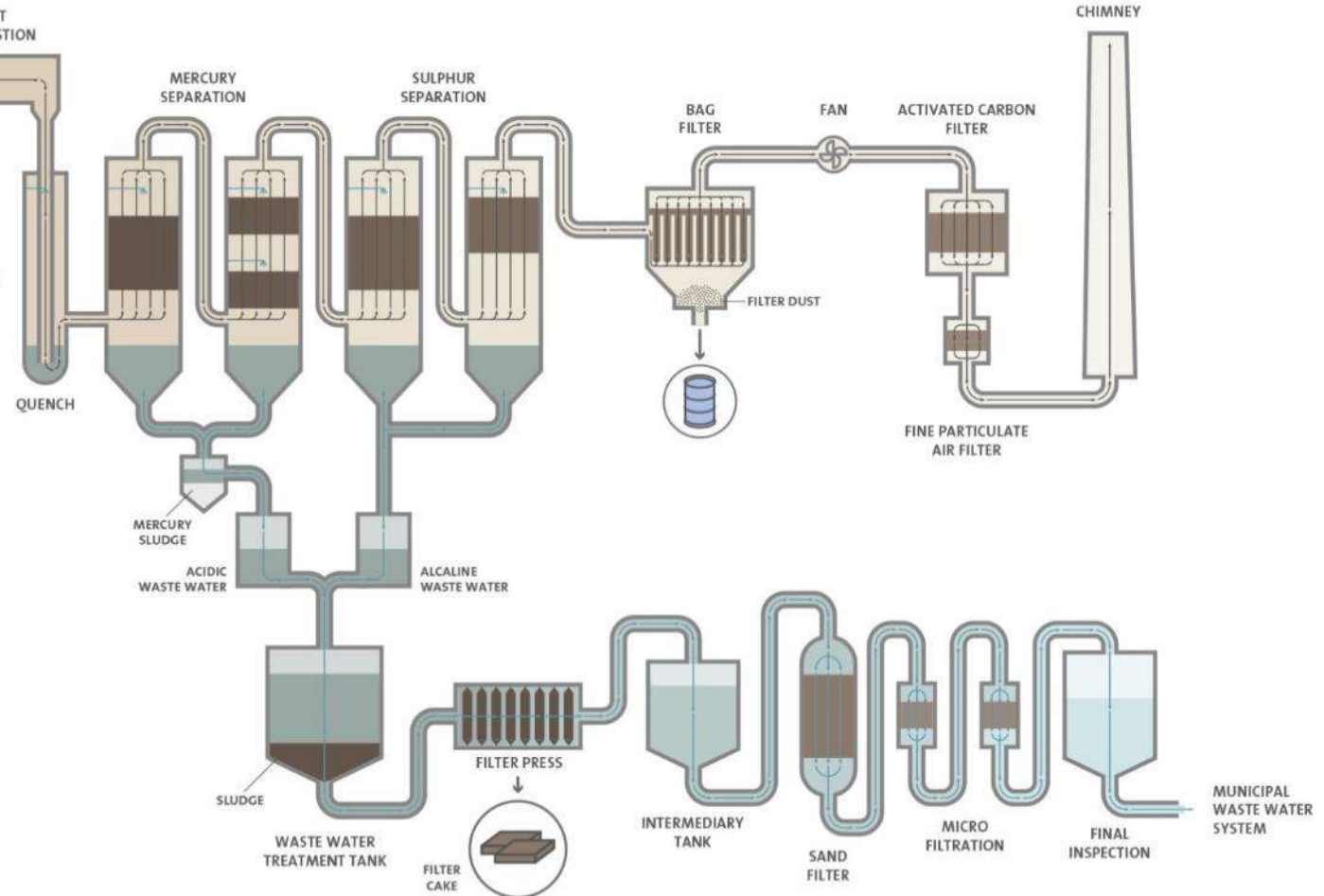
- desorption of the pollutants at 750 – 850°C
- destruction of the organic pollutants in the post-combustion chamber



2

WASTE GAS WET CLEANING

- condensation of Mercury
- removal of Sulfur



3

WASTE GAS DRY CLEANING

- removal of trace level Mercury
- removal of other pollutants and fine dust
- removal HEPA Filter

WASTE WATER TREATMENT

Treatment Process

Decontamination Furnace

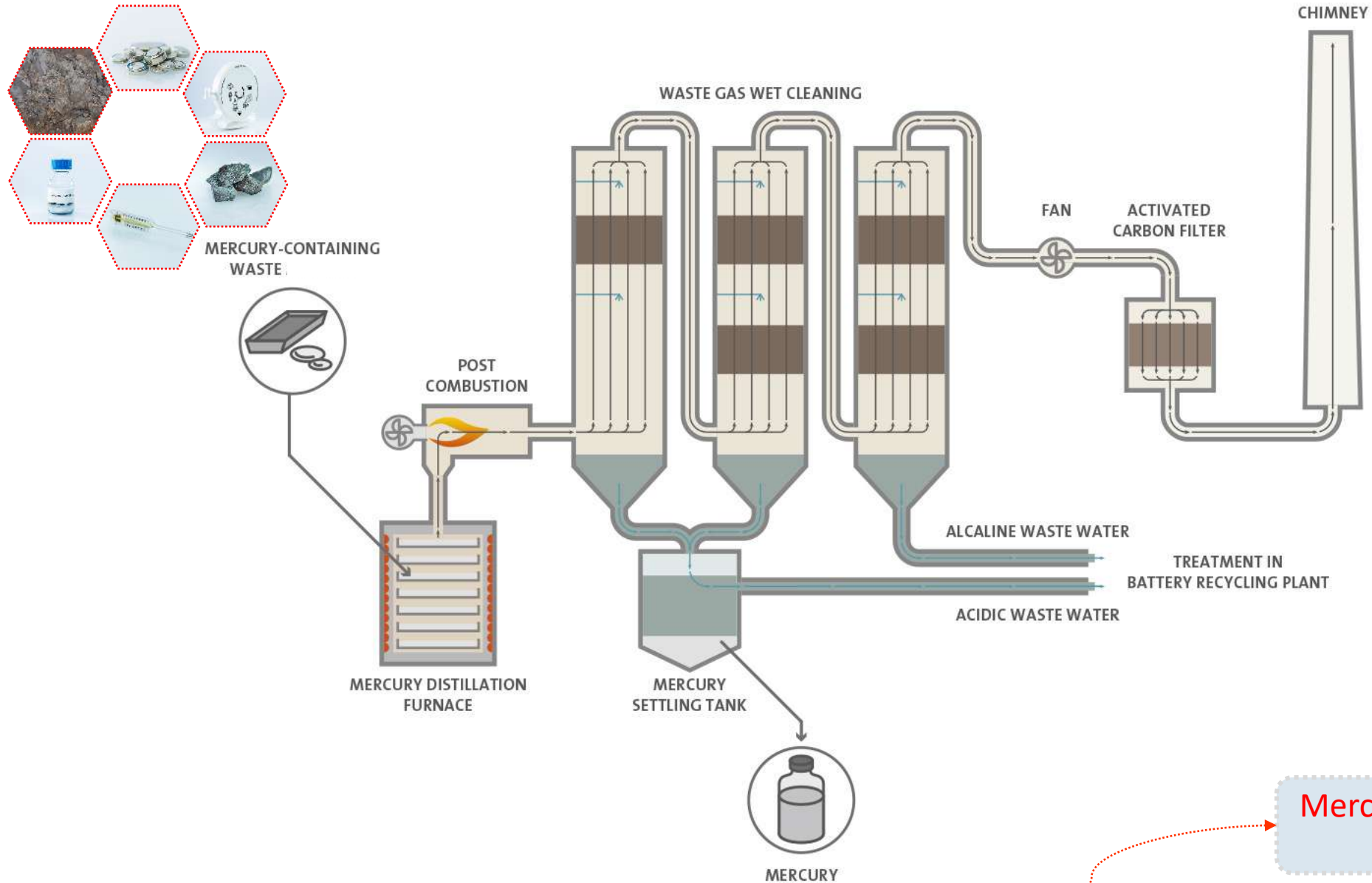


Treatment Process

Post-combustion chamber



Mercury Distillation – Process



Mercury Stabilisation
 $\text{Hg} \rightarrow \text{HgS}$

Treatment of Mercury Wastes

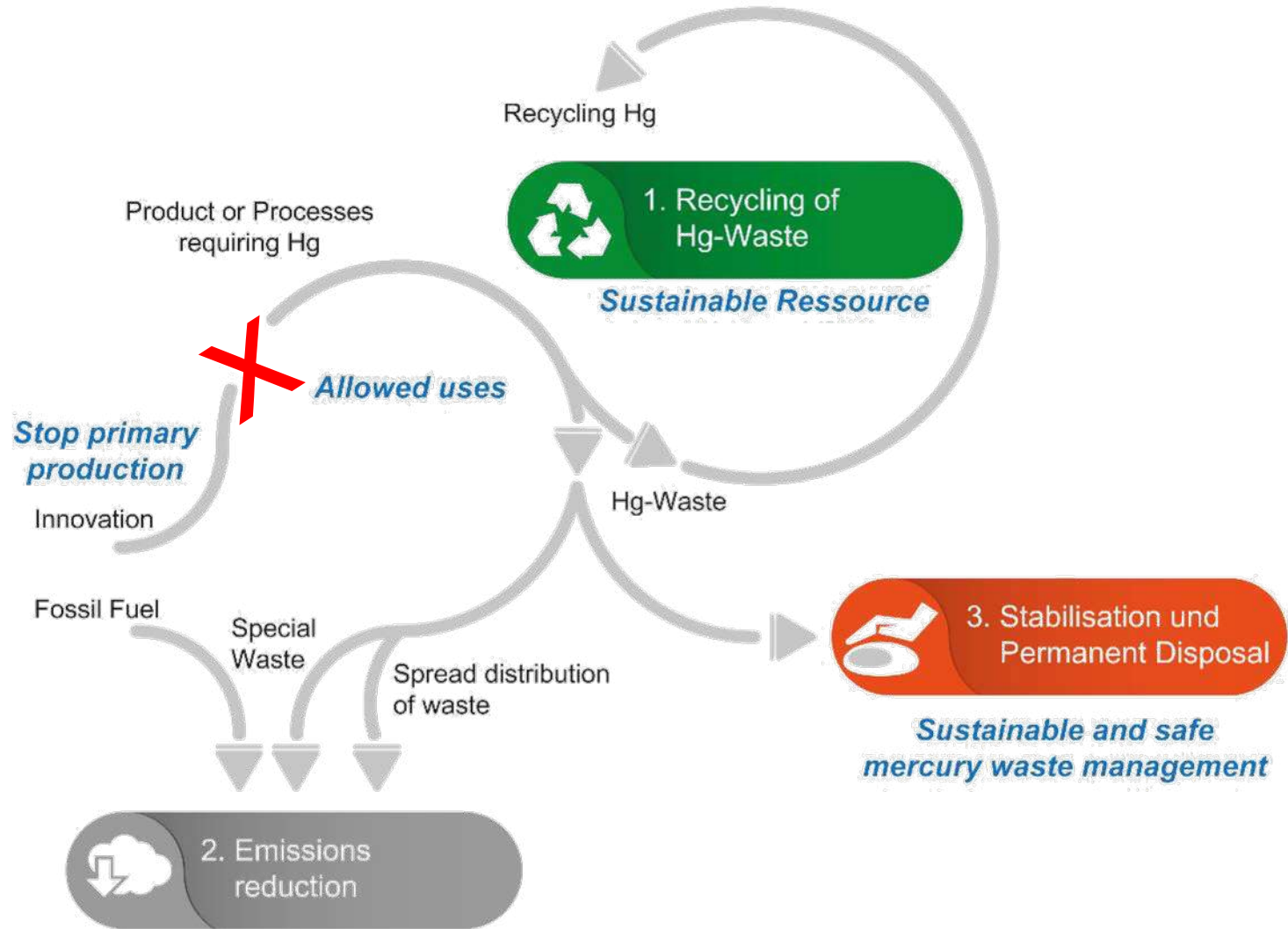


Mercury Operations



Why Stabilise Mercury? - The Minamata Convention

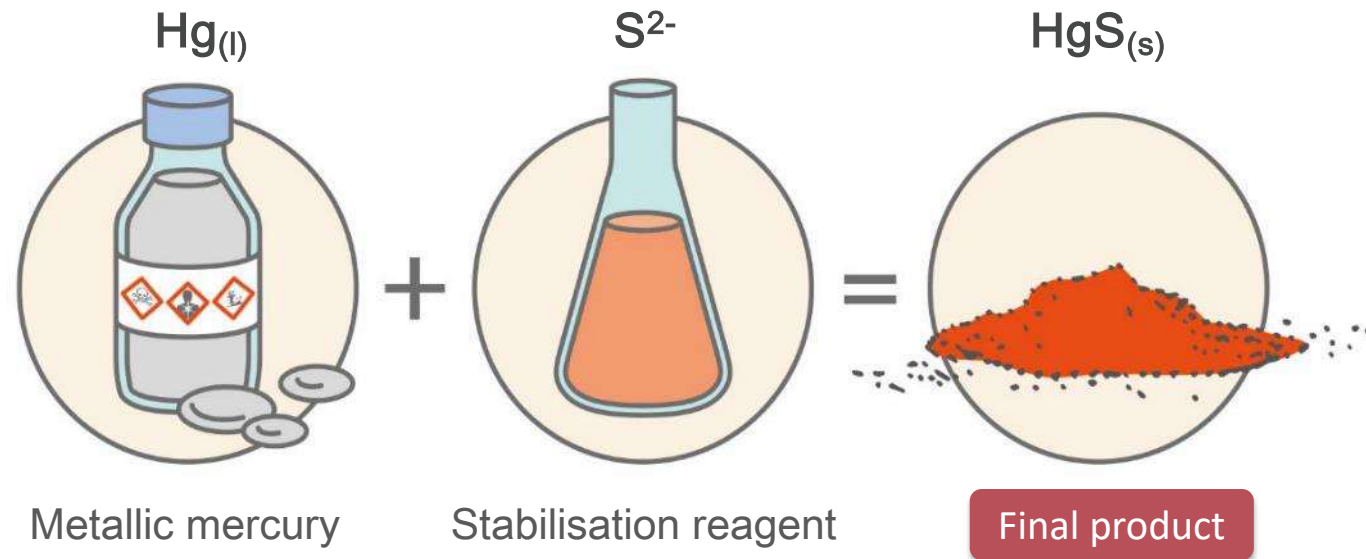
2018
Multilateral environmental agreement that addresses specific human activities which are contributing to widespread mercury pollution.



BATREC's approach

Transform highly toxic Hg into non-toxic HgS

Controlled reaction at ambient temperature and pressure → low risk, high conversion and consistent product



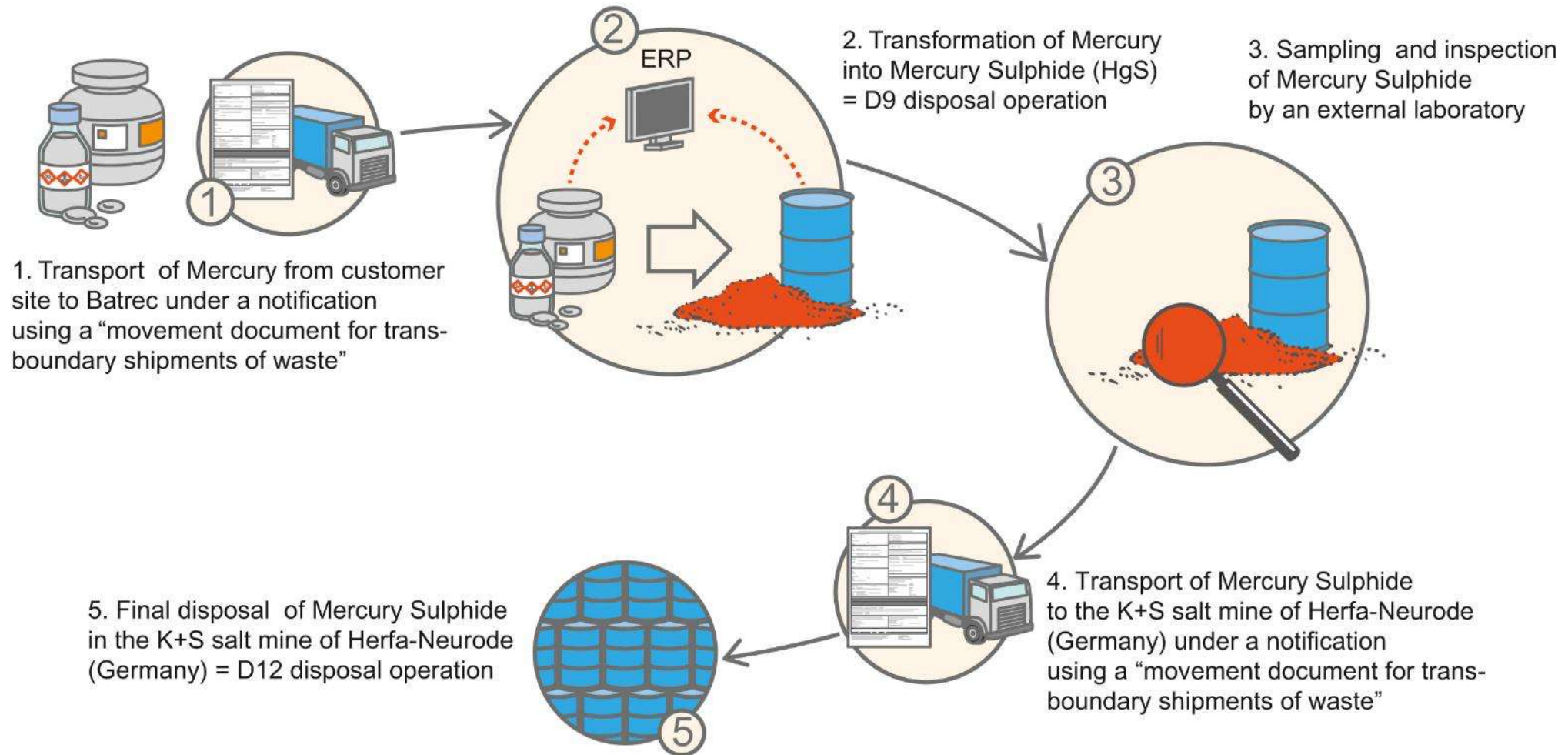
HgS is non-toxic

HgS is the most stable mercury compound

HgS is the most insoluble mercury compound

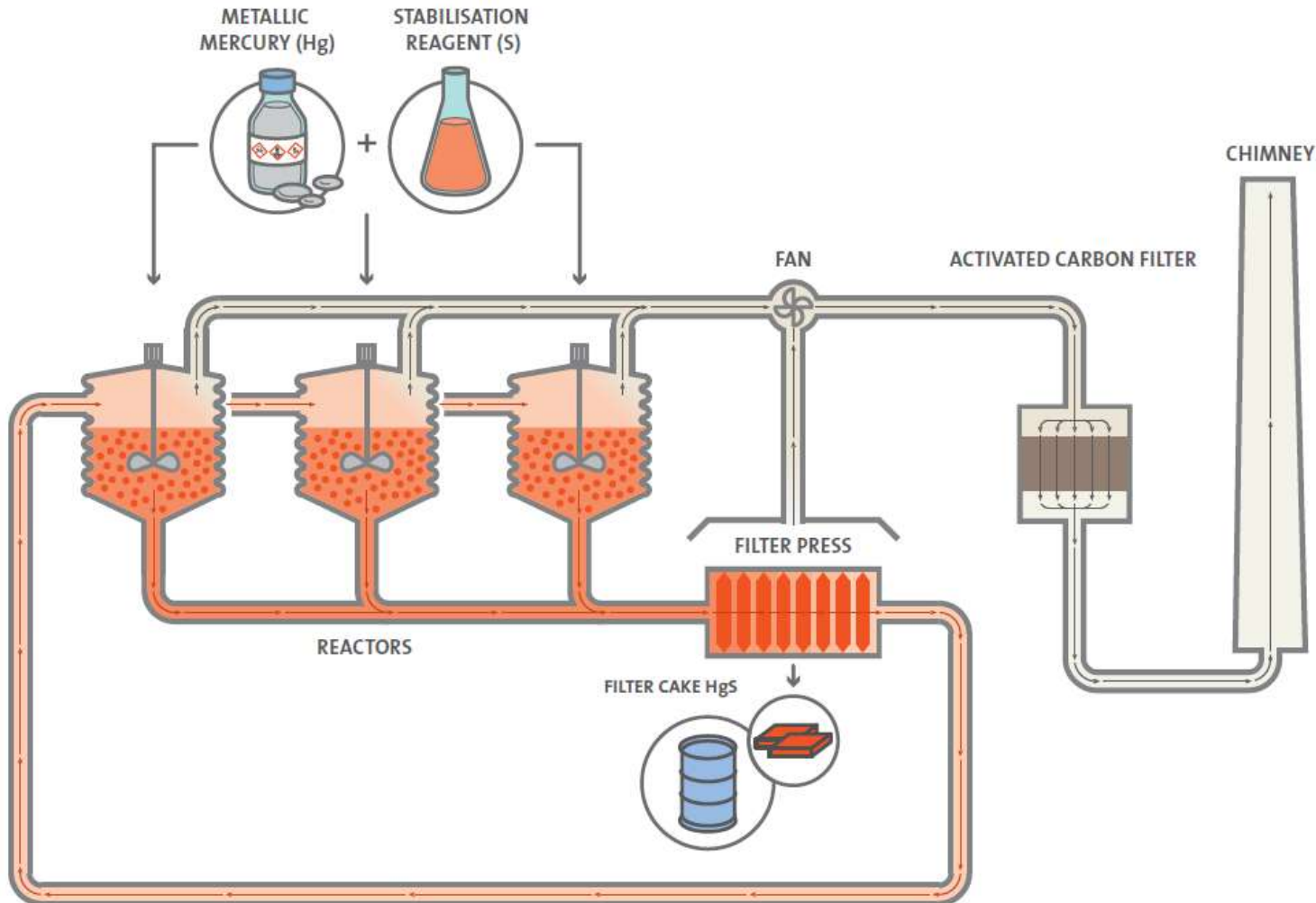
HgS is the natural mineral form [cinnabar] of Hg

Traceability is key



Mercury Stabilisation – Process

Capacity: 1.200 t/year



Process characteristics

- ✓ batch process
- ✓ wet process at low temperatures in a closed circuit limits the risk of Hg emissions
- ✓ no gaseous Hg in the process
- ✓ stabilisation solution is regenerated
→ *zero effluents produced*
- ✓ simple reactants

Safe permanent storage of HgS in a salt mine (Germany)
Long-term-safe removal of hazardous wastes from the biosphere



Acceptance criteria



Restricted acceptance criteria

e.g.

- Not biodegradable
- Not releasing gases
- Non-liquid
- Not radioactive
- No insufficient stability of geomechanical conditions



Shaft transport



Underground transport



Storage chambers



Sealing of by walls



Natural barriers

- Salt
- Clay
- Bunter stone

Artificial barriers

- Waste packaging
- Brick walls...



OUR COMMITMENTS

“ Risk management to ensure the safety of all our stakeholders.
Traceability, non-dilution, and decontamination of the
waste life cycle to protect the environment from pollution.”

Thank you for your attention

On-site Mercury Stabilization: The safest solution for Hg disposal

A photograph of industrial machinery, likely a large-scale vacuum drying or mercury treatment system. It features a large horizontal cylindrical tank, various pipes, valves, and orange structural supports.

Agenda

- 1 Market Position
- 2 Technologies for Mercury Treatment
- 3 Mobile Mercury Conversion unit
- 4 References

Market position



Who we are

- **econ** is a family owned business & was founded in 2003 as machinery supplier
- 30 hazardous waste treatment projects (20 **VacuDry**[®] units) worldwide, on 3 continents, in 13 different countries
- **econ** provides the most efficient and cleanest solutions to recover resources from industrial hazardous wastes

2007

World's first mercury recovery from NORM waste



2016

Mobile Mercury Conversion Unit



2018

1st mercury recovery facility of the southern hemisphere (Australia)



2021

VacuDry[®] for mercury clean up in India



2023

NORM waste treatment center



Currently commissioning two **VacuDry**[®] units for Greece and manufacturing Australia

one MMCU unit for

A photograph of industrial machinery, including a large horizontal cylindrical tank and various pipes and valves, set against a background of orange structural beams.

Agenda

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Mercury Treatment Technologies



3 Technologies for recovery & disposal

Recovery of mercury

VacuDry®



VacuDry® - vacuum distillation

- Max. 400 °C
- Low vacuum < 50 mbar(abs)
- For soils & sludges cont. with elemental mercury

HTTU



High Temperature Treatment Unit

- Max. 1000 °C
- Atmospheric pressure
- For spent activated carbon & catalyst cont. with mercury compounds

Conversion of mercury

MMCU



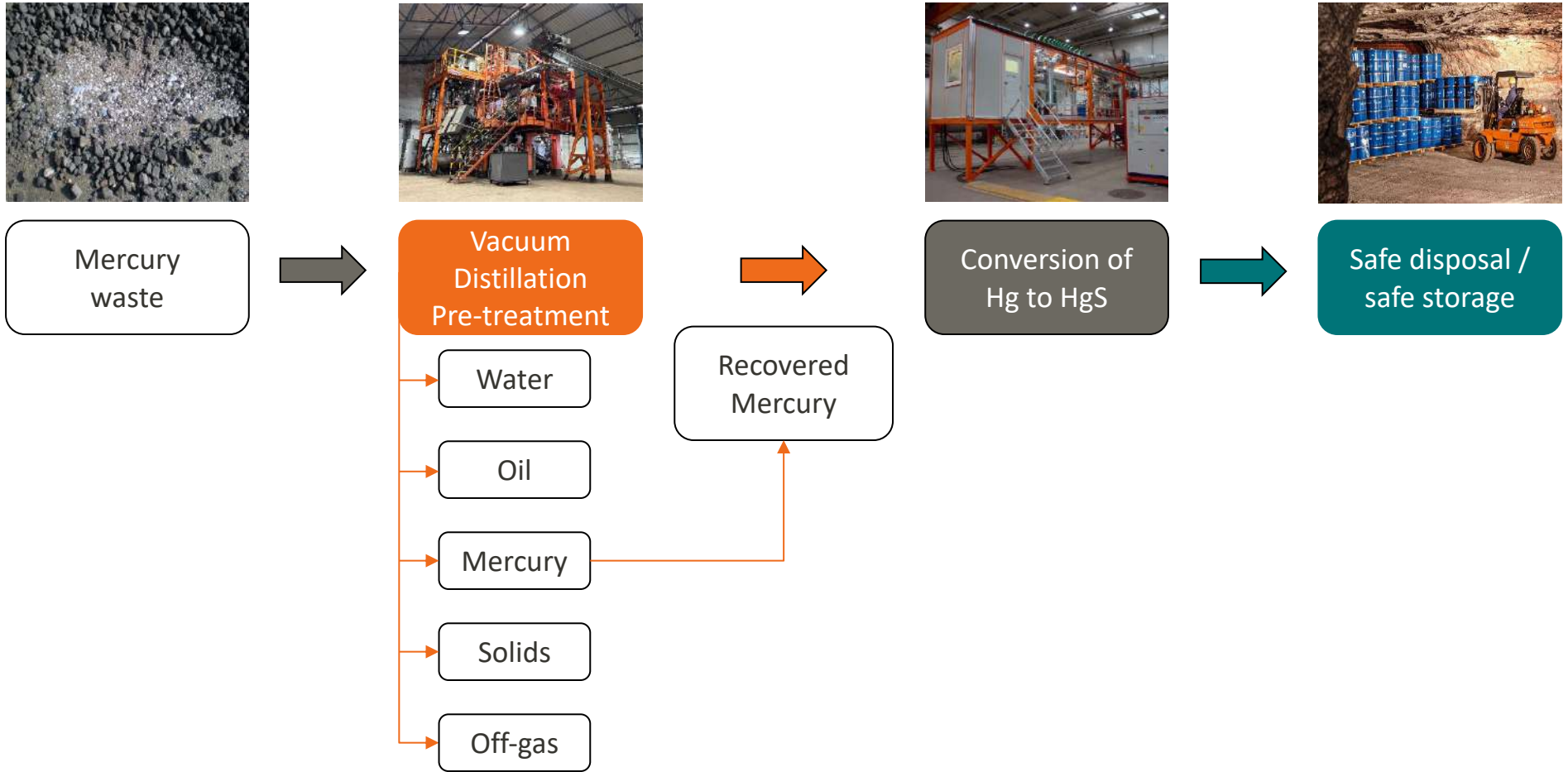
Mobile Mercury Conversion Unit

- max. 200 °C
- Atmospheric pressure
- Conversion of pure elemental Hg to HgS (cinnabar)

Mercury Treatment Technologies

Concept

Recovery of mercury from hazardous wastes and stabilization for final disposal



A photograph of industrial machinery, including large stainless steel tanks and pipes, with orange safety railings. A white box with the word "Agenda" is overlaid on the image.

Agenda

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Mobile Mercury Conversion Unit



Conversion process / Working principle

Elemental Mercury



Elemental Sulphur



Mercury sulphide (HgS)



MMCU - Mobile Mercury Conversion Unit

Conversion of pure elemental mercury (Hg) to mercury sulphide (HgS / cinnabar)

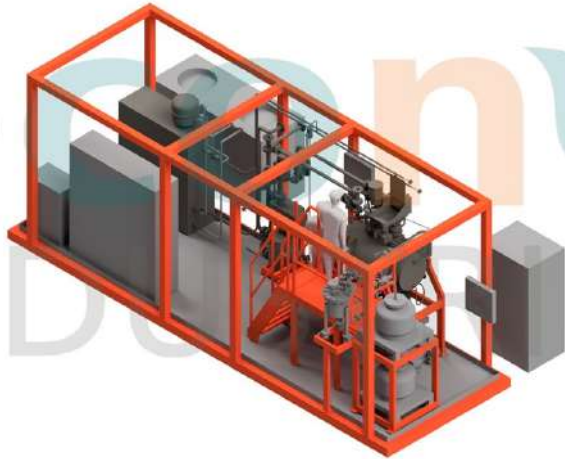


Mobile Mercury Conversion Unit



Technical specifications

MMCU 150



- **Input material:** metallic mercury + sulphur
- **Throughput capacity*:** 200 kg per batch (MMCU 150)
1,500 kg per batch (MMCU 1 500)
* 3 – 4 batches per day
- **Product treatment parameters:** liquid conversion process
max. 200 °C material temperature
at atmospheric pressure
nitrogen atmosphere

MMCU 1 500



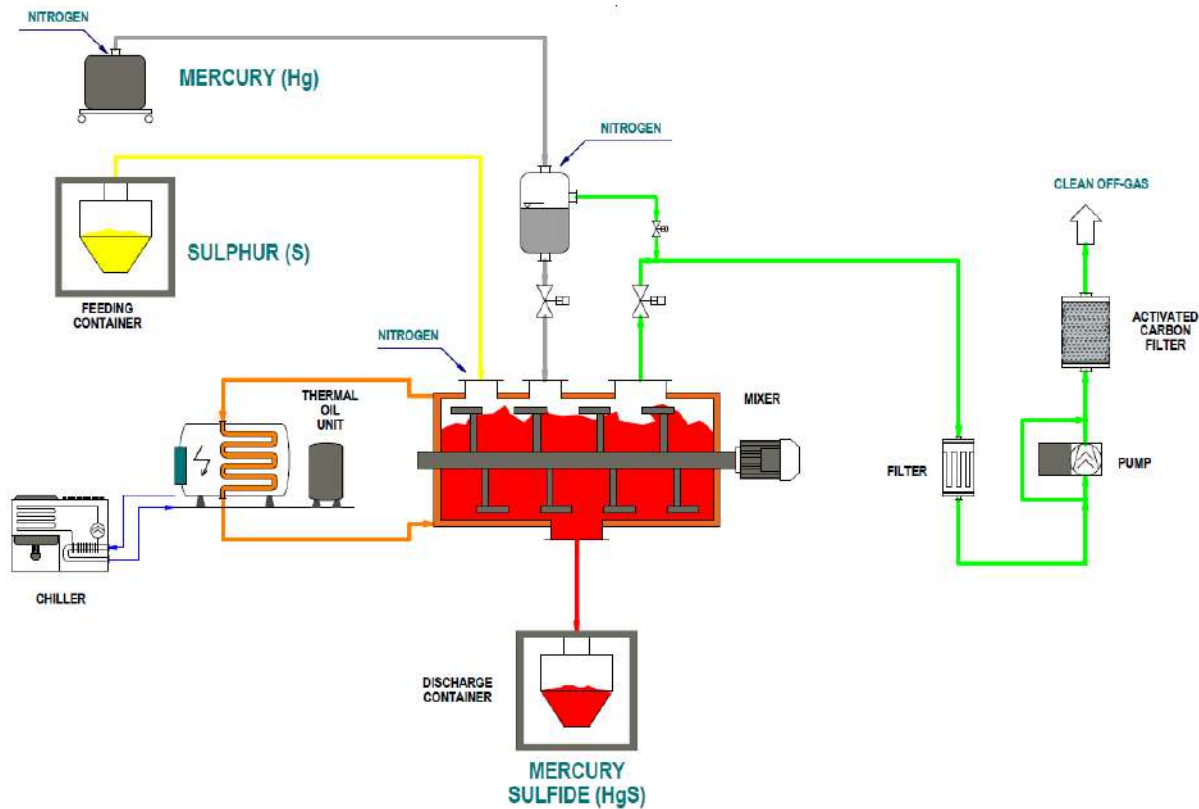
- **Solid output material:** 99.999 % conversion to HgS with
Hg emissions < 10 µg/m³
discharge temperature < 80 °C
- **Staff required:** 1 staff for MMCU 150
2 staff for MMCU 1 500
- **Delivery time:** 8 – 9 months for manufacturing

Mobile Mercury Conversion



For final disposal of pure elemental mercury

MMCU Flowchart



Specifications

- Nitrogen atmosphere, safe and hermetically closed mixer
- Operation under ambient pressure
- Mercury and sulphur are fed to the mixer
- Increased temperatures inside the mixer by the thermal oil unit
- active cooling of the mixer
- Liquid phase chemical reaction
- Low off-gas flow during stabilization process

An inert atmosphere and a liquid phase chemical reaction ensure safe operations

Mobile Mercury Conversion Unit



Disposal procedure

- Once stabilised into mercury sulphide (cinnabar), the mercury can be disposed of underground in salt mines (Germany - Europe) or in secure or chemical landfills (other continents).



Packaging (requirement)

- Plastic inliner
- Metal drum



The resulting HgS successfully passes the landfill acceptance criteria.

- Mercury conversion – 99.999%
- Hg emissions “air” < 10 $\mu\text{g}/\text{m}^3$ (20 $\mu\text{g}/\text{m}^3$ – Salt mine threshold)
- Leachate test < 0.025 mg/L (Method TCLP1311 - US EPA) – not required in Europe

A photograph of industrial machinery, likely a mercury treatment unit, with large stainless steel tanks and orange structural frames.

Agenda

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References / Update



econ stabilized more than 500 tons of Hg on-site



UK – chlor-alkali plant:

400 tons Hg converted to HgS on-site

Poland – chlor-alkali plant:

130 tons Hg converted to HgS on-site

HgS was sent to salt mine (K+S) for final disposal

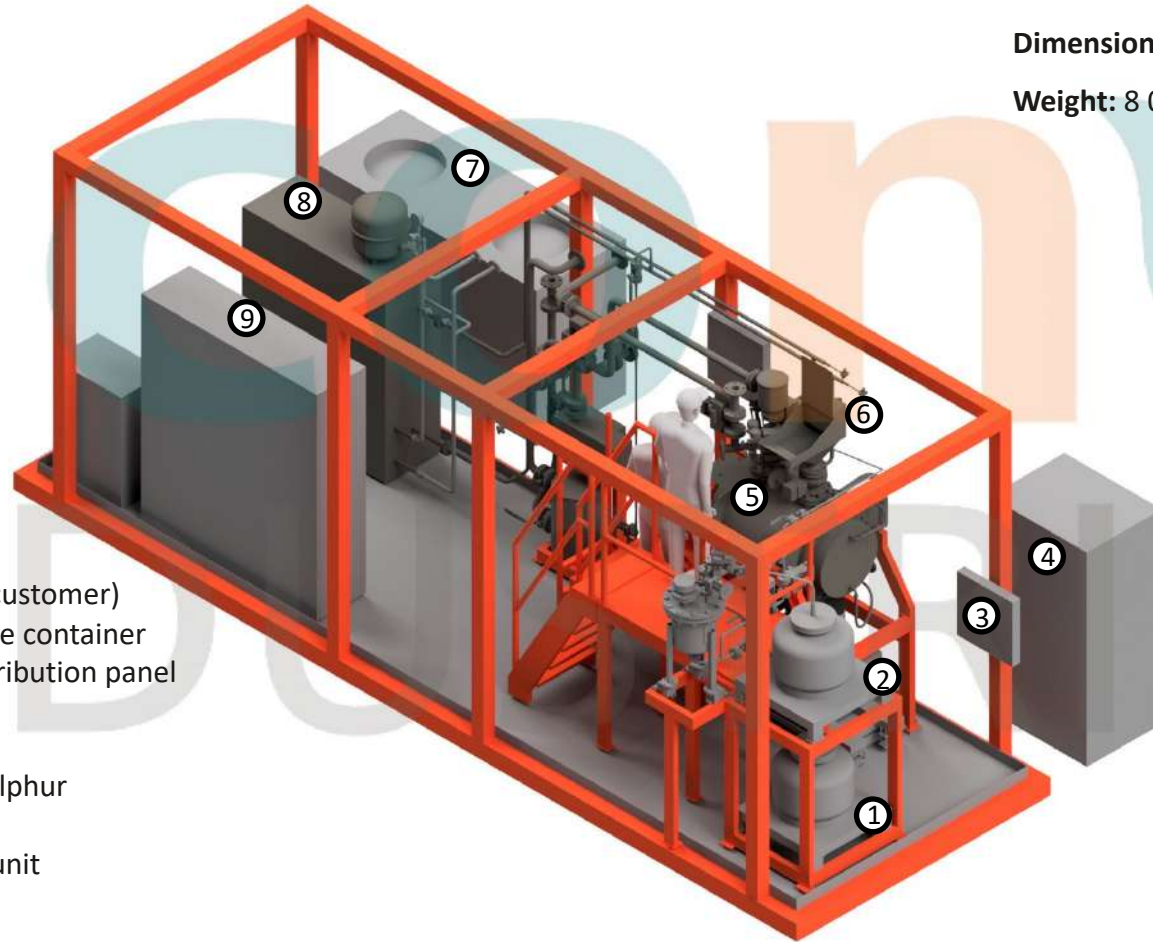
References / Update



MMCU 150 for Australia

Dimensions: 8m x 2.6m x 3.2m (L x W x H)

Weight: 8 000kg (approx.)

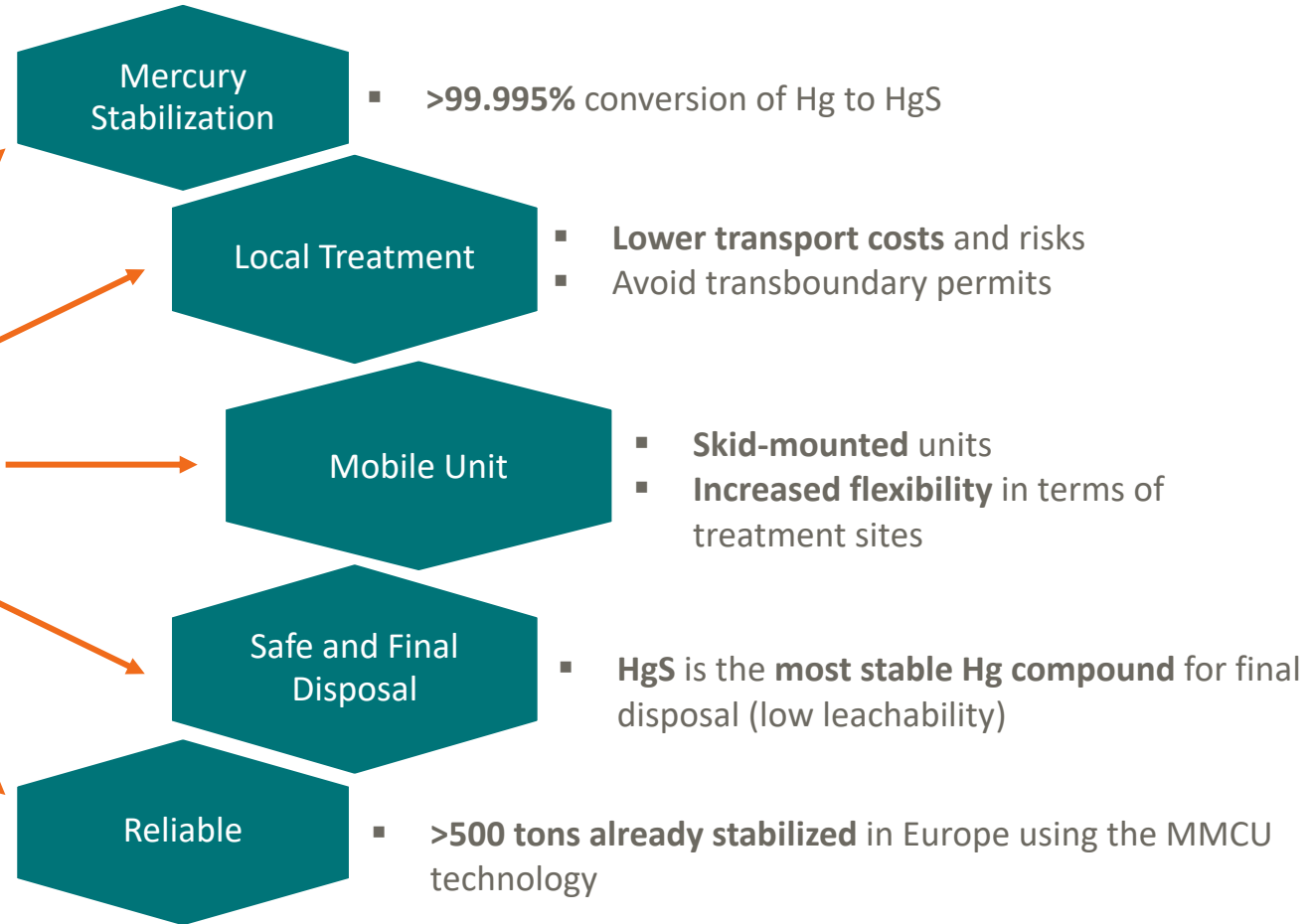


1. Mercury container (customer)
2. Mercury intermediate container
3. Air and nitrogen distribution panel
4. Nitrogen supply
5. MMCU – Mixer
6. Feeding Hopper – Sulphur
7. Cooling system
8. Thermal oil heating unit
9. Electrical cabinet

Australia will be 100 % independent in terms of sustainable Mercury Waste Treatment

Conclusion

Outcomes and Main Benefits



**econ industries provides
the most efficient and cleanest
solutions ...**



**... to recover resources
from special wastes worldwide.**

econ 
INDUSTRIES

Zero industrial waste ...!

Q&A

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Questions and Answers

facilitated by [Misuzu Asari](#), Co-lead of the Waste Management Area



Next steps and closing remarks

Rodges Ankrah

*U.S. Environmental Protection Agency, Co-Chair of the UNEP Global Mercury
Partnership Advisory Group*