



ATLANTIC ENERGY
ASSOCIATES LLC

Inventory enhancement for coal using the iPOG

Workshop to enhance inventories and strategies under Article 8 of the Minamata Convention in South Africa

Eskom Megawatt Park, Sunninghill, Johannesburg

Thursday 31st May 2024

Wojciech Jozewicz, PhD

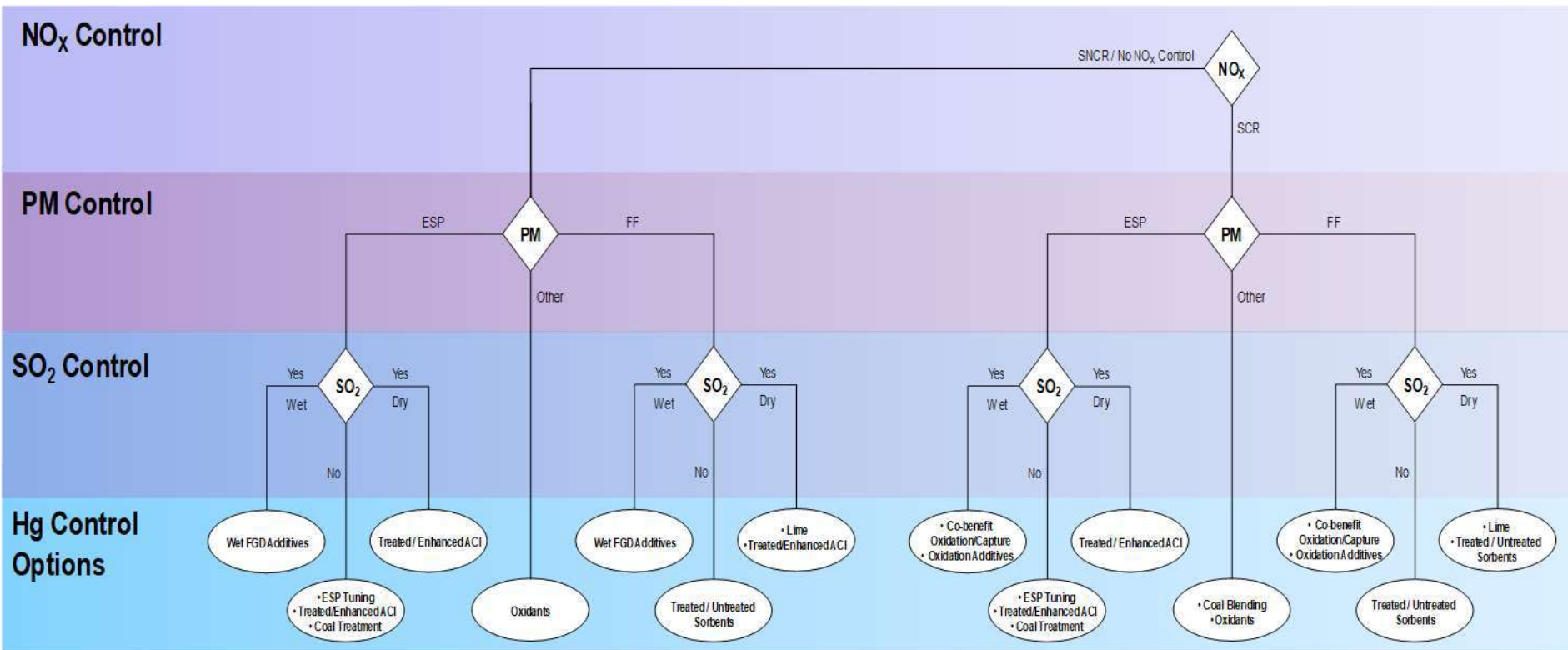


Background

- iPOG is an interactive application for UNEP's POG
- Developed for UNEP Coal Partnership by Niksa Associates
- Tool to help determine approaches to Hg emission control and rank them for individual coal-fired units
- Tradeoffs were made to only include basic inputs at the expense of quantitative accuracy
- Allows for addition of flue gas cleaning approaches and systems according to BAT/BEP
 - Improved fuel quality and blending
 - PM, SO₂, and NO_x control systems for co-benefit
 - Dedicated Hg control technology
- Follows “Decision Tree” logic from the POG

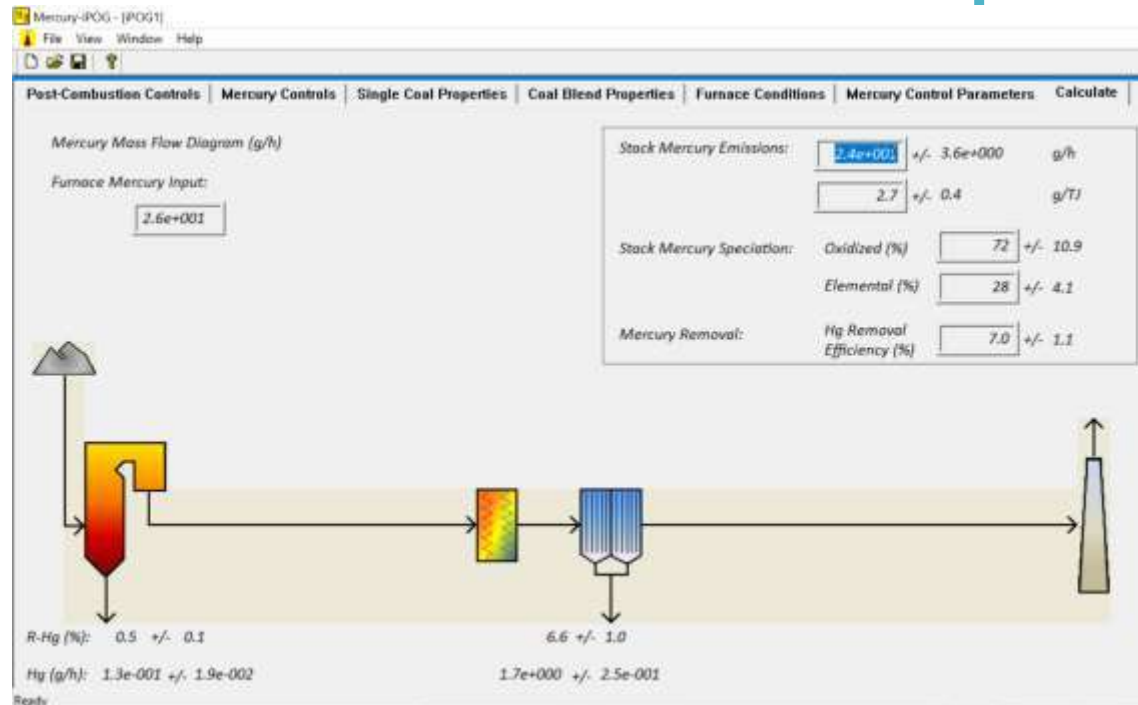


iPOG "Decision Tree" Structure





iPOG Calculations Tab - Example



- Final tab to initiate calculations sequence
- In this example: older but well-controlled 500 MW, wall-fired boiler, burning low-S coal, cold-side ESP
- Essentially no Hg removal predicted (<10%)
- Estimated Hg emissions of 24 g/h or up to about 0.2 ton Hg/year



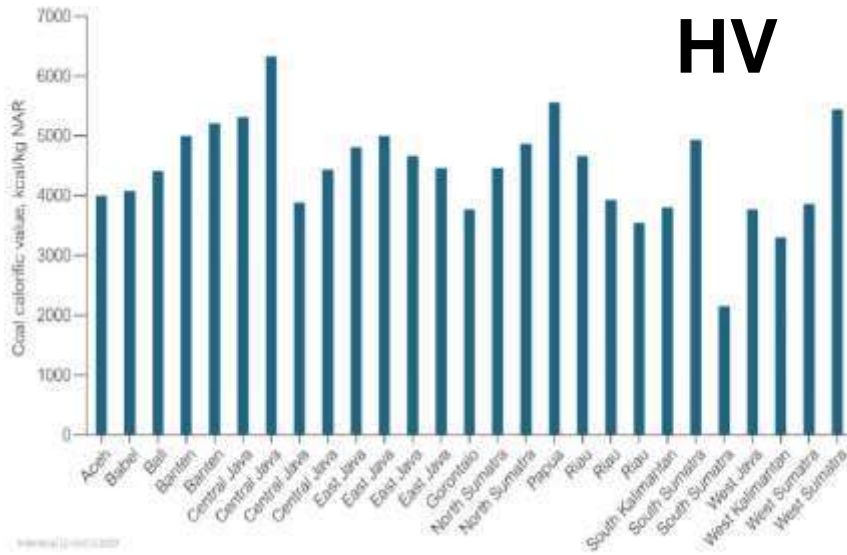
Data Quality Very Important

- Stakeholders should ensure that any missing data are obtained directly from the plant considered for the project rather than by the proxy calculations
- Unit details: generating capacity, commissioning date, planned retirement
- Unit performance: operational load, utilization, gross efficiency, coal consumption, LOI
- Coal quality: calorific value, ash-S-Hg-Cl content
- Emissions controls: PM, FGD, Hg controls
- Quality data in – Quality results out!

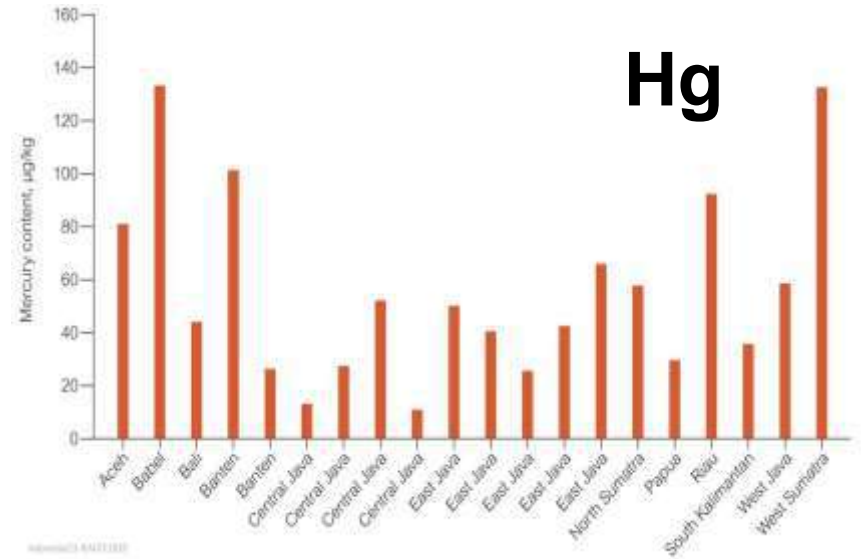


Variability Examples

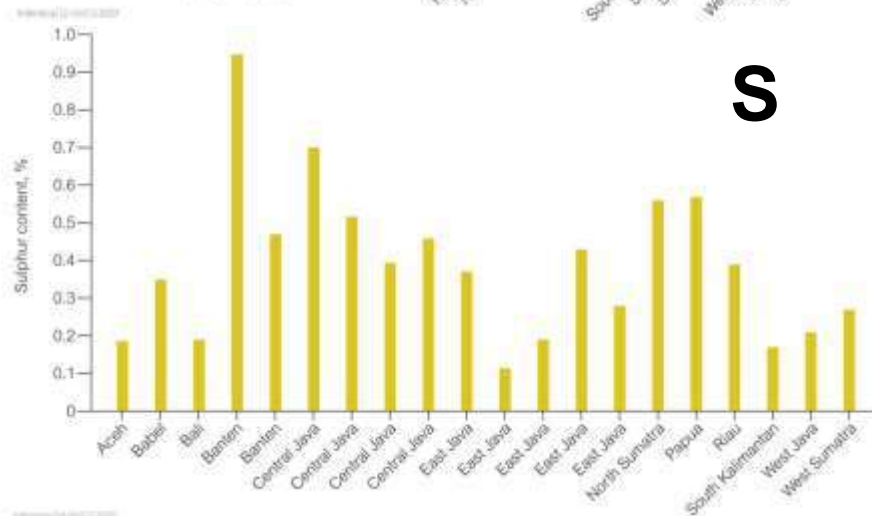
HV



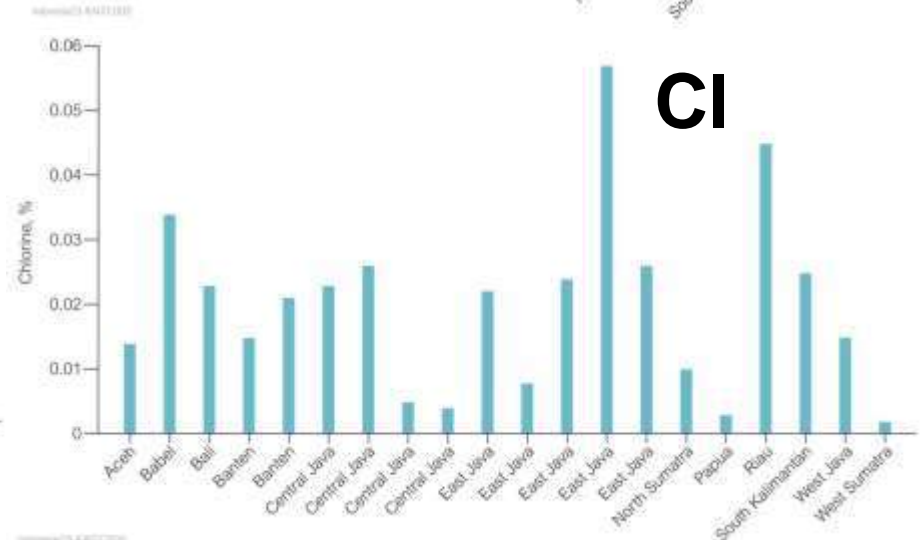
Hg



S

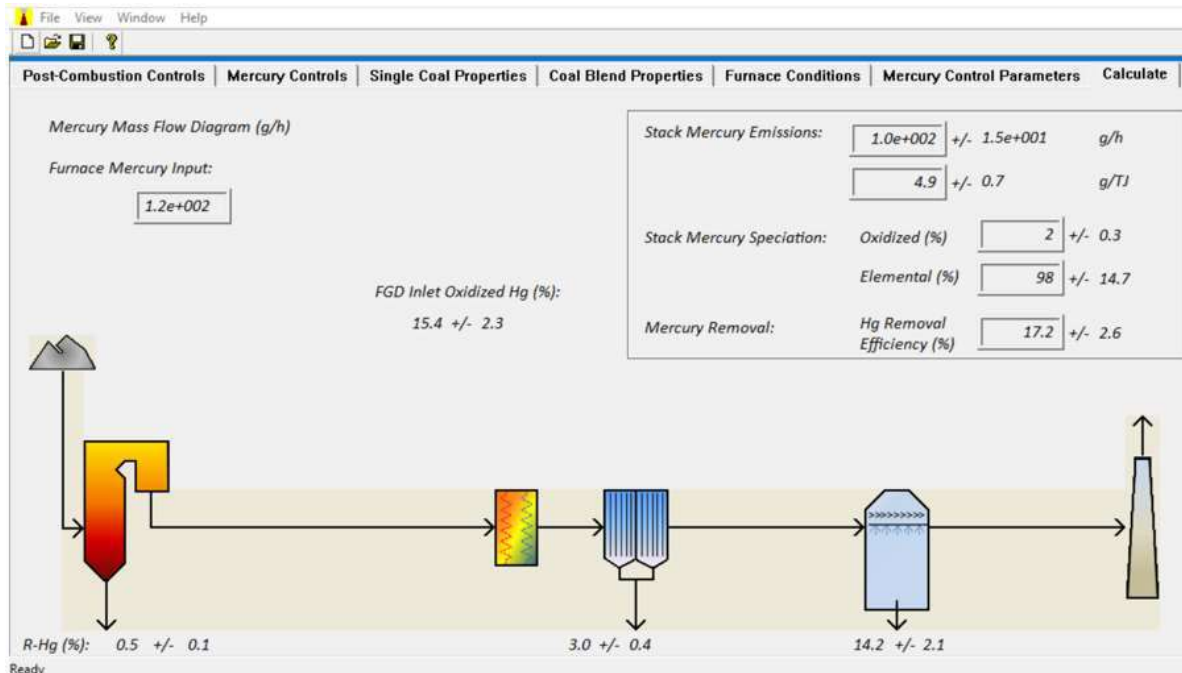


Cl





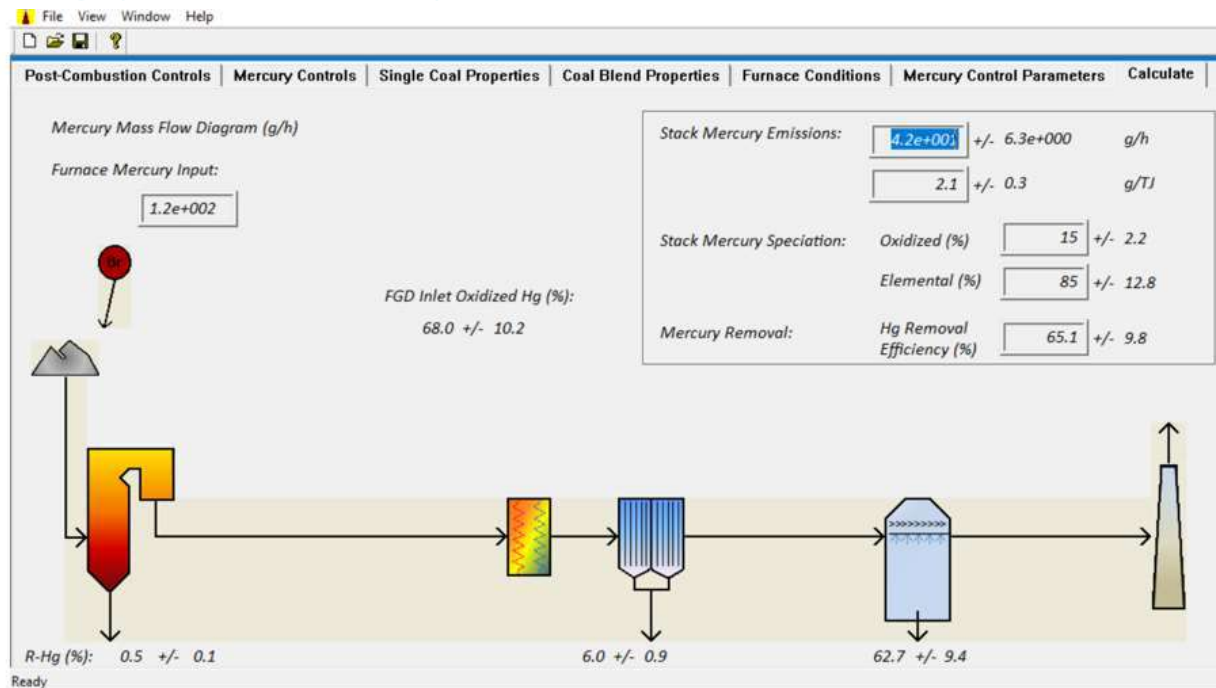
State-of-the-art Unit



- 800 MW unit with ESP and wet FGD
- Only about 17% Hg removal; emissions 98% of Hg₀ and 2% of Hg₊₊
- More mercury removal could be accomplished with more efficient Hg₀ oxidation



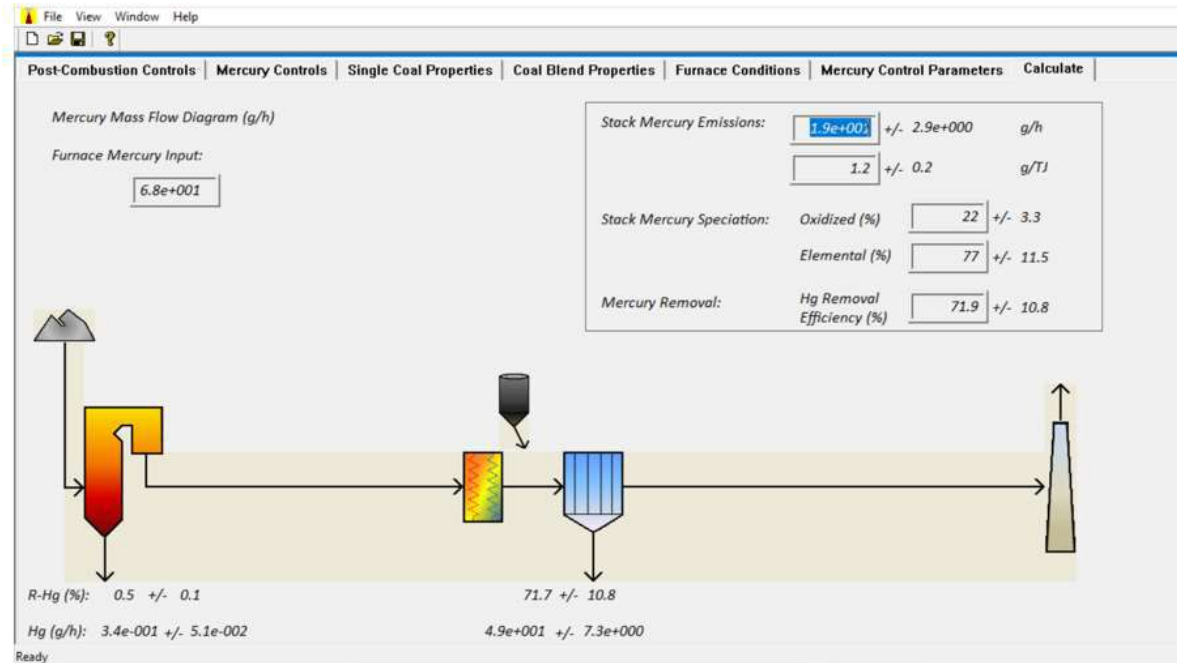
Strategy for Improvement



- Over 65% removal with addition of 250 ppm of Br to coal
- Some other options
 - Activated carbon injection upstream of the ESP
 - SCR- expensive



Another Strategy



- Addition of 0.02 g/m³ of activated carbon upstream of the baghouse increases Hg removal to 72%



Summary

- Only limited application of FGD throughout the country
- Data quality very important for accurate predictions
- Compliance and improvement strategies for units of varying size and age
- However,
 - Growth projected for power demand
 - Ambitious renewable energy goals



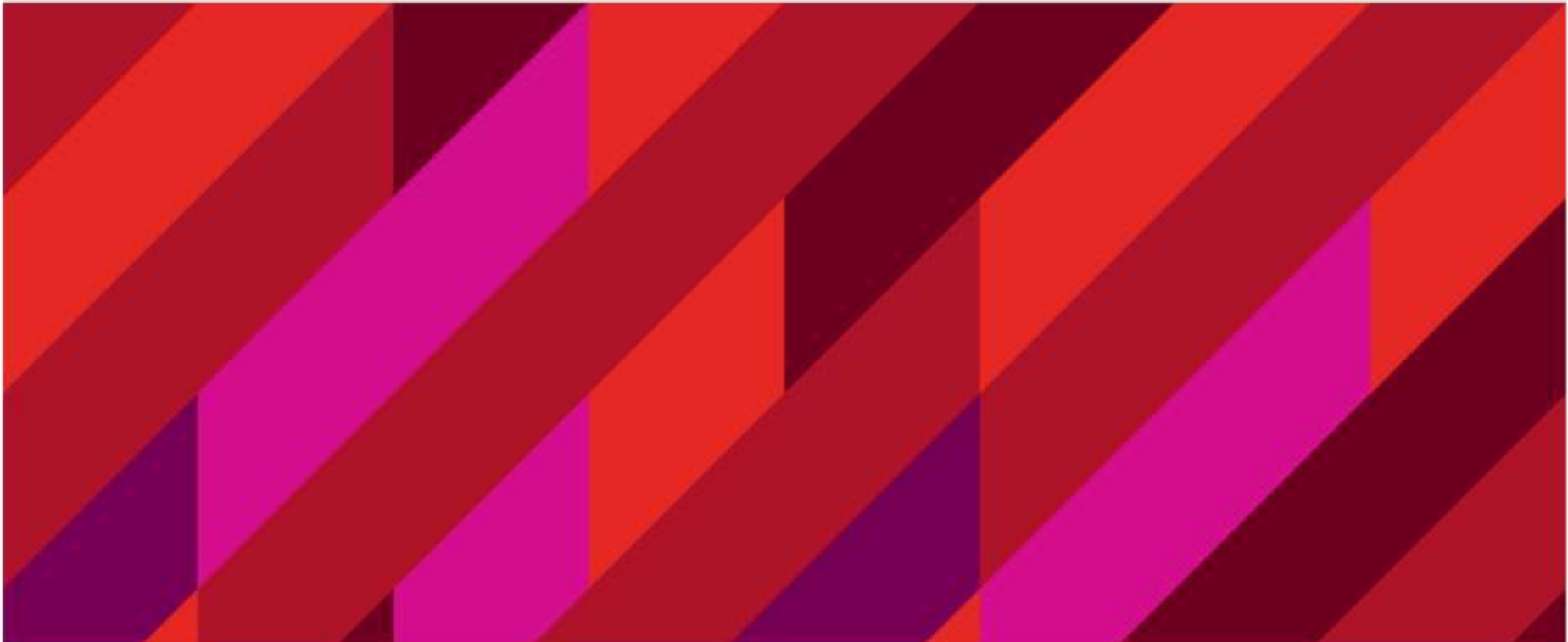
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Thank you!

Mercury emission inventory enhancement for the coal sector

PROF LESLEY SLOSS

June 2024





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Full report freely available from [www. sustainable-carbon.org](http://www.sustainable-carbon.org)

Improving data quality and applicability in the coal sector



Creating enhanced emission factors






Improving activity data

Focussing on the important differences

Ranking the results

Emission factors for coal

$$\text{EMISSION} = \text{EF} \times \text{RF} \times \text{AV}$$

Approach	Emission Factor, EF Relates to the mercury content of the coal	Retention factor, RF Subtracts mercury that ends up in ash etc	EF x RF Estimates the amount of mercury released per unit of coal fired	Activity value Multiplies to cover all coal used in each source	Comments
UNEP Toolkit*	Generic – 0.05 g/kg	Generic - minus 10%	0.045 g/kg	Coal burn, t	Assumes all plants and coals are identical. Targets busier units, often unfairly
2017 UNEP Project	Coal analyses Results averaged across the fleet	iPOG# model of generic national plant	Convert to g/TJ Applies to all plants and takes average plant efficiency into account	Coal burn, t	EF and RF are now more accurate for the national coal fleet BUT still assumes all plants and coals are identical
Advanced projects (eg Indonesia)	Coal analysis on a unit-by-unit basis	iPOG analysis on a unit-by-unit basis	Unit-specific emission factor	Unit-specific plant activity	Produces a unit-specific emission estimate
					

* <https://web.unep.org/globalmercurypartnership/mercury-emissions-coal-fired-power-plants-indonesia>

<https://web.unep.org/globalmercurypartnership/interactive-process-optimization-guidance-ipog%E2%84%A2>

Plant sampling for EF and RF

SAMPLES TAKEN AT PLANTS IN INDONESIA



Coal



Ash



Stack

- Sampling of coal as delivered and as fed into the boiler
- Coal samples from numerous mines were analysed and results collated
- Monitoring and mass balances are challenging but are still more useful than generic emission factors

Emission factor in g/GJ vs g/kg

A SLIGHT MODIFICATION TO THE EF UNITS CAN INCREASE VALUE



EF in g/kg

Average mercury contents in coal give an EF of around 0.05 g/kg but mercury contents of coal can vary significantly, even from seam to seam

When we multiply the EF by the amount of coal burned, we get a total emission based on coal consumption.

BUT

This assumes that all coals have the same mercury content AND that all coal burns the same



EF in g/GJ

If we know the amount of energy (gigajoules) of energy produced by each tonne of coal, then we can estimate mercury emissions by power output – g/GJ





This allows us to determine when plants are either firing poor-quality coal or running inefficiently

This allows us to see which plants are “cleaner” – that is, which plants produce more power whilst burning less coal. This information is not available with a g/kg emission factor

Creating the dataset

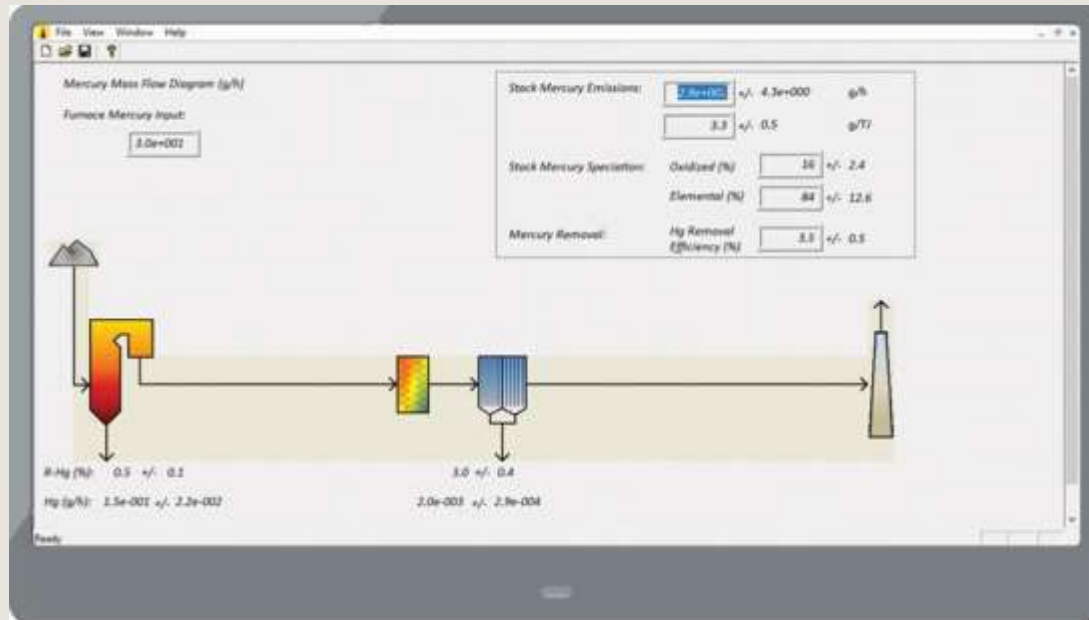
REAL DATA FROM PLANTS

MISSING DATA ESTIMATED THROUGH PROXY CALCULATIONS

 Unit/plant details	 Unit performance	 Emission controls	 Fuel quality
Unit and plant name Location Generating capacity Certified operating and commissioning date	Operational load Utilisation/capacity factor Specific energy consumption Annual coal consumption	Flue gas desulphurisation In boiler additives NOx burners or SCR PM controls	Calorific value Mercury content Sulphur content Chlorine content

Using the iPOG

INTERACTIVE PROCESS OPTIMISATION GUIDANCE TOOL



- Input unit-specific data
- RF estimated from plant configuration, coal chemistry and control technologies in place
- Results based on extrapolation and modelling of data from thousands of real data sets
- Used to focus on RELATIVE emission rates, not “actual”
- Image

Demonstration to follow

Creation of the dataset

LIVING DOCUMENT TO BE UPDATED REGULARLY
PROVENANCE OF DATA TO BE RECORDED

No	Power unit	2020	Capacity (MW)	2020	2020	Total electricity production/gross (MWh)*	Installed	2020 Purnomo		Annual operating hours (hour/year)		2020 Purnomo + Baruya	2020 Purnomo + Baruya	2020 Purnomo + Baruya	Stack Mercury Emission					
	Power unit	WEPP UNIT NAME	Installed	Commissioning Date (WEPP)	Remaining life as of 2020 (40yr life)	Total electricity production/gross (MWh)*	Annual utilisation	Operational load	Fuel Consumption (ton/year)	Specific Fuel Consumption (ton/MWh)	hours/y	coal mercury content	coal sulphur content	coal chlorine content	Check if POG?	Annual Hg Emission, coal input, kg	Hg Emission intensity, g/MWh	Annual Hg Emission, POG prediction, kg	Remaining Plant Life Hg Emission, kg	
No	Power unit	WEPP UNIT NAME	Installed	Commissioning Date (WEPP)	Remaining life as of 2020 (40yr life)	Total electricity production/gross (MWh)*	Annual utilisation	%	Fuel Consumption (ton/year)	Specific Fuel Consumption (ton/MWh)	hours/y	ug/kg	%	%	Result (g/h)					
	PLTU Celukan Bawang	CELUKAN BAWANG	142	2015	35	274,827.00	22.09	89.20	150,000.00	0.546	7,970	SWFGD	44.6	0.19	0.023	2.4	6,890	24	360	12,600
	PLTU Celukan Bawang	CELUKAN BAWANG	142	2015	35	274,827.00	22.09	89.20	150,000.00	0.546	7,970	SWFGD	44.6	0.19	0.023	2.4	6,890	24	360	12,600
	PLTU Celukan Bawang	CELUKAN BAWANG	142	2015	35	274,827.00	22.09	89.20	150,000.00	0.546	7,970	SWFGD	44.6	0.19	0.023	2.4	6,890	24	360	12,600
	PLTU Paton Unit 5	PAITON-J NO 1	610	2000	20	3,549,546.25	66.42	100.00	1,618,332	0.456	6,416	SWFGD	26	0.19	0.024	5.5	42,077	12	8,901	178,017
	PLTU Paton Unit 6	PAITON-J NO 2	610	2000	20	3,549,546.25	66.42	100.00	1,618,332	0.456	6,899	SWFGD	26	0.19	0.024	5.5	42,077	12	8,901	178,017
	PLTU Paton Unit 7	PAITON-J NO 1	615	1999	19	3,584,327.53	66.53	106.52	1,947,206	0.543	6,146	SWFGD	40.88	0.118	0.0078	12	78,621	22	23,368	443,963
	PLTU Paton Unit 8	PAITON-J NO 2	615	1999	19	3,584,327.53	66.53	106.03	2,218,145	0.619	7,025	SWFGD	40.88	0.118	0.0078	14	80,700	25	31,054	580,027
	PLTU TJB Unit 1	TANJUNG JATI-B NC	710	2006	26	4,469,025.09	71.85	93.07	1,626,044	0.409	7,619	WLST	13.3	0.7	0.023	2.4	24,286	5	4,383	113,045
	PLTU TJB Unit 2	TANJUNG JATI-B NC	710	2006	26	4,879,123.00	78.45	93.07	1,944,636	0.399	8,215	WLST	13.3	0.7	0.023	2.3	25,864	5	4,473	116,289
	PLTU TJB Unit 3	TANJUNG JATI-B NC	721.8	2011	31	4,563,981.90	72.18	91.59	1,780,111	0.390	8,411	WLST	27.5	0.52	0.026	3.2	48,953	11	5,696	176,587
	PLTU TJB Unit 4	TANJUNG JATI-B NC	721.8	2012	32	4,201,836.40	66.45	91.59	1,679,439	0.400	7,807	WLST	27.5	0.52	0.026	3.4	46,185	11	5,710	182,723



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Thank you

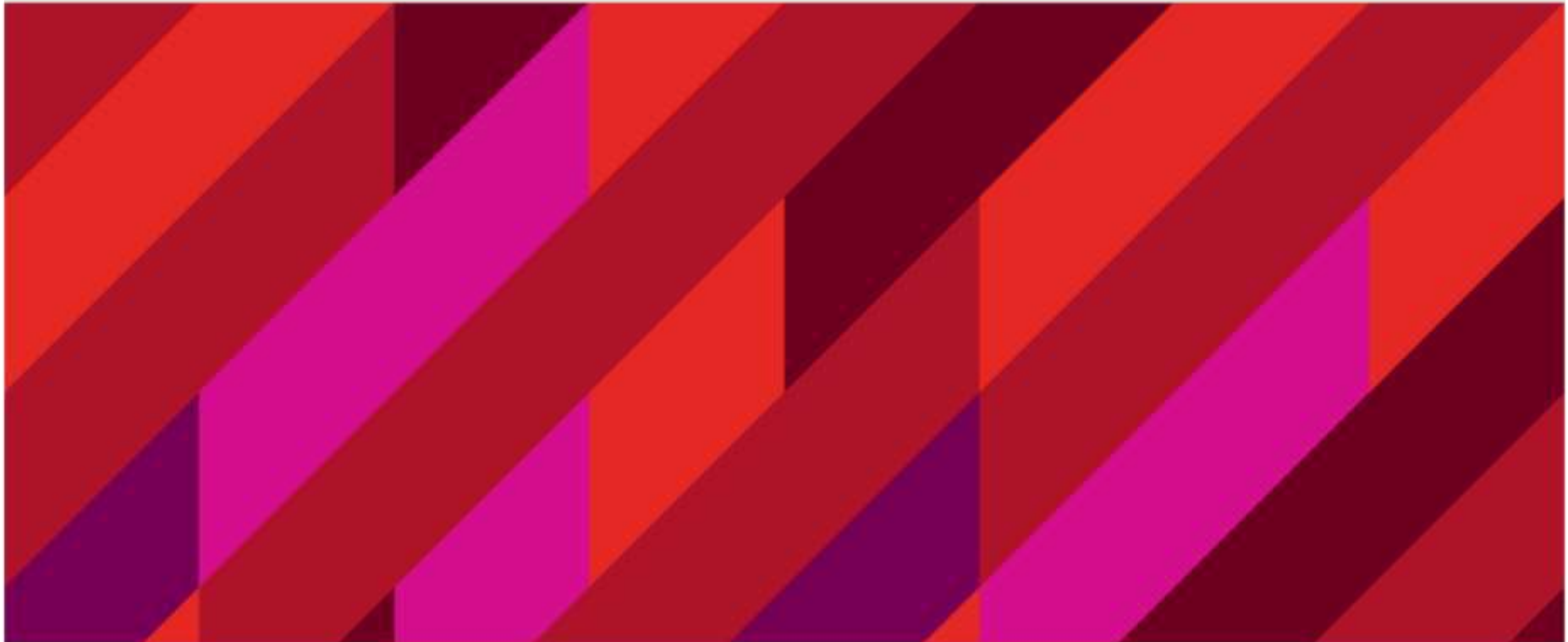
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Using enhanced data to rank sources and create a cost-effective targeting strategy

PROF LESLEY SLOSS

June 2024



Informed ranking of data



Selecting appropriate ranking criteria

Examples of ranking results

Informing a strategic approach to
emission reduction

Changing the input

MOVING FROM ASSUMPTIONS TO REAL DATA

All plants are assumed equal

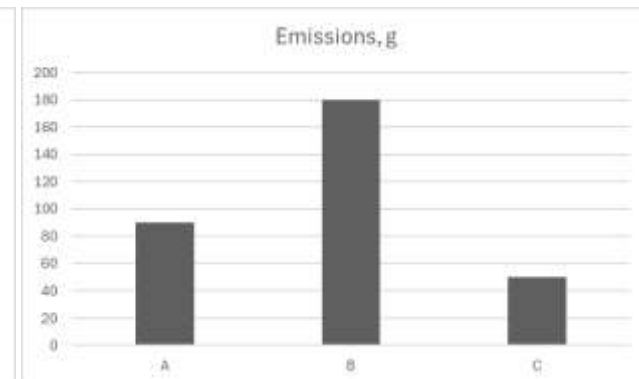
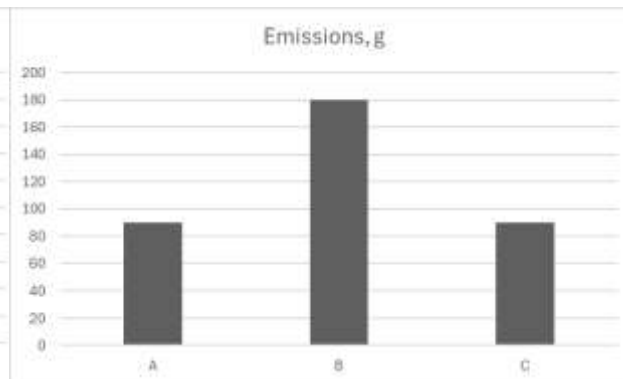
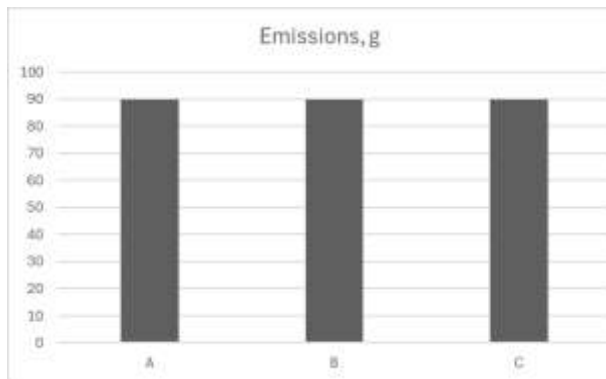
Plant	Emissions, g	EF, g/kg	RF, %	AV, t
A	90	1	10	100
B	90	1	10	100
C	90	1	10	100

Plant B has higher mercury coal

Plant	Emissions, g	EF, g/kg	RF, %	AV, t
A	90	1	10	100
B	180	2	10	100
C	90	1	10	100

Plant C has higher ash retention

Plant	Emissions, g	EF, g/kg	RF, %	AV, t
A	90	1	10	100
B	180	2	10	100
C	50	1	50	100



The total coal burned is the same in all assumptions
All plants are NOT the same

Emission factor in g/GJ vs g/kg

A SLIGHT MODIFICATION TO THE EF UNITS CAN INCREASE VALUE



EF in g/kg

Average mercury contents in coal give an EF of around 0.05 g/kg but mercury contents of coal can vary significantly, even from seam to seam

When we multiply the EF by the amount of coal burned, we get a total emission based on coal consumption.

BUT

This assumes that all coals have the same mercury content AND that all coal burns the same



EF in g/GJ

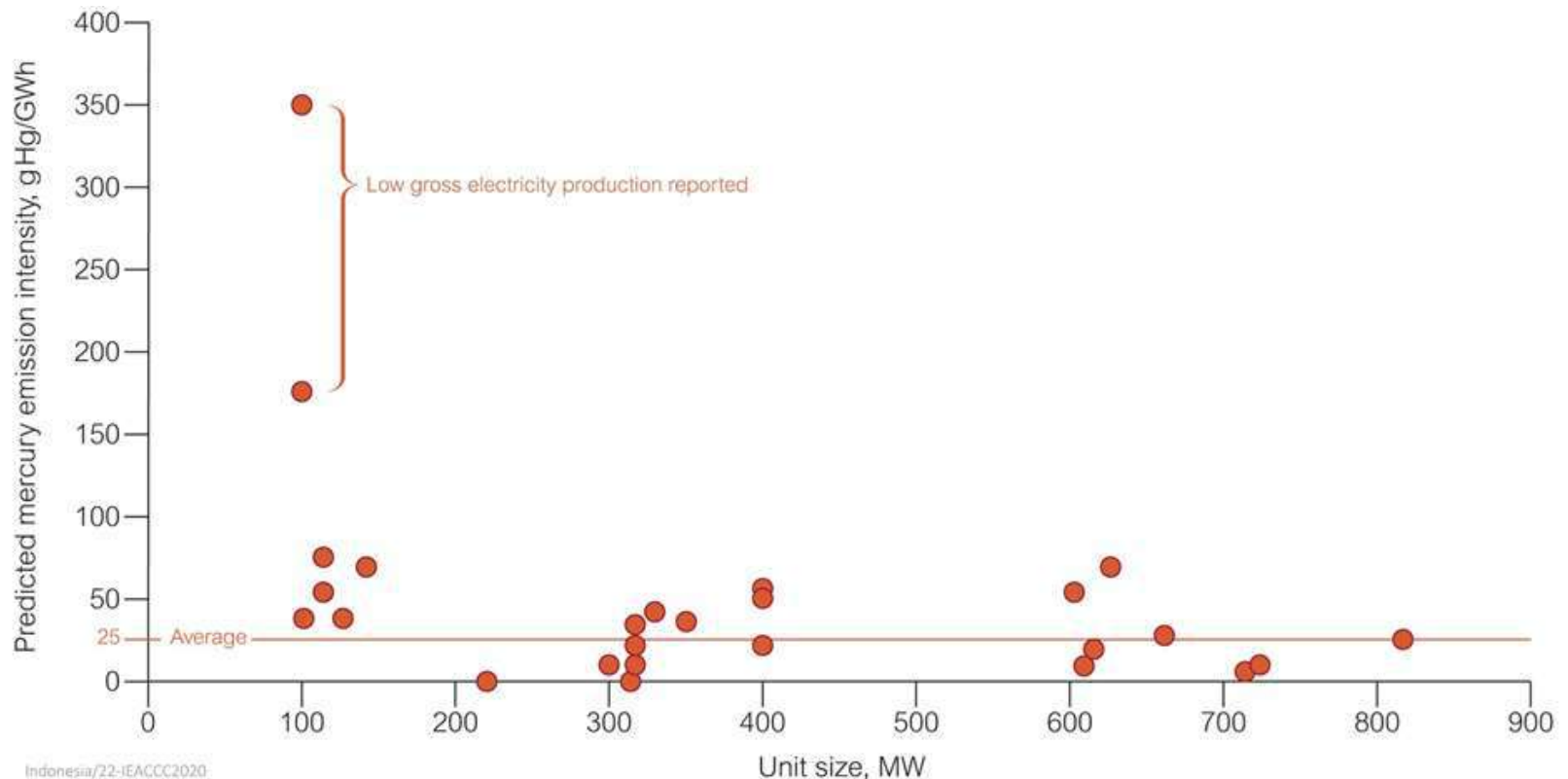
If we know the amount of energy (gigajoules) of energy produced by each tonne of coal, then we can estimate mercury emissions by power output – g/GJ

This allows us to determine when plants are either firing poor-quality coal or running inefficiently

This allows us to see which plants are “cleaner” – that is, which plants produce more power whilst burning less coal. This information is not available with a g/kg emission factor

Fleet emission intensity

Changing the emission factor from g/kg to g/GJ



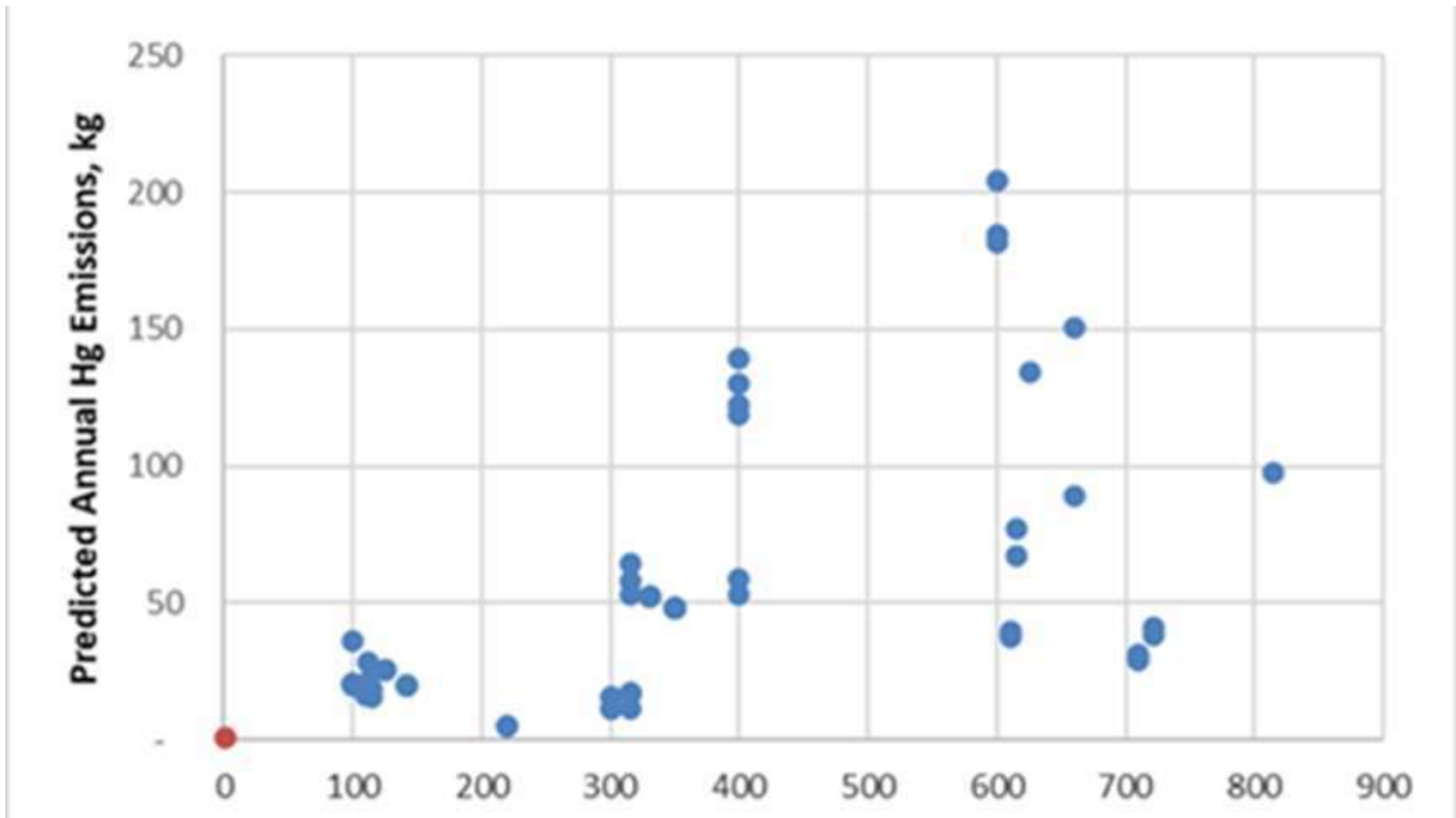
Amount of Hg (g) emitted per GWh of electricity produced – indication of “cleaner” burning plants.

No indication of size-related intensity of Hg emissions for units >100 MW

SOME UNITS EMIT AN ORDER OF MAGNITUDE MORE MERCURY PER GWh OF POWER PRODUCED THAN OTHERS



Predicted annual emissions from Indonesian coal plants



Bringing in plant lifetime

IMMEDIATELY BRINGS IN COST-EFFECTIVENESS



Current method

Identifies plants which emitted the most mercury in the last operating year

BUT assumes all plants are the same age



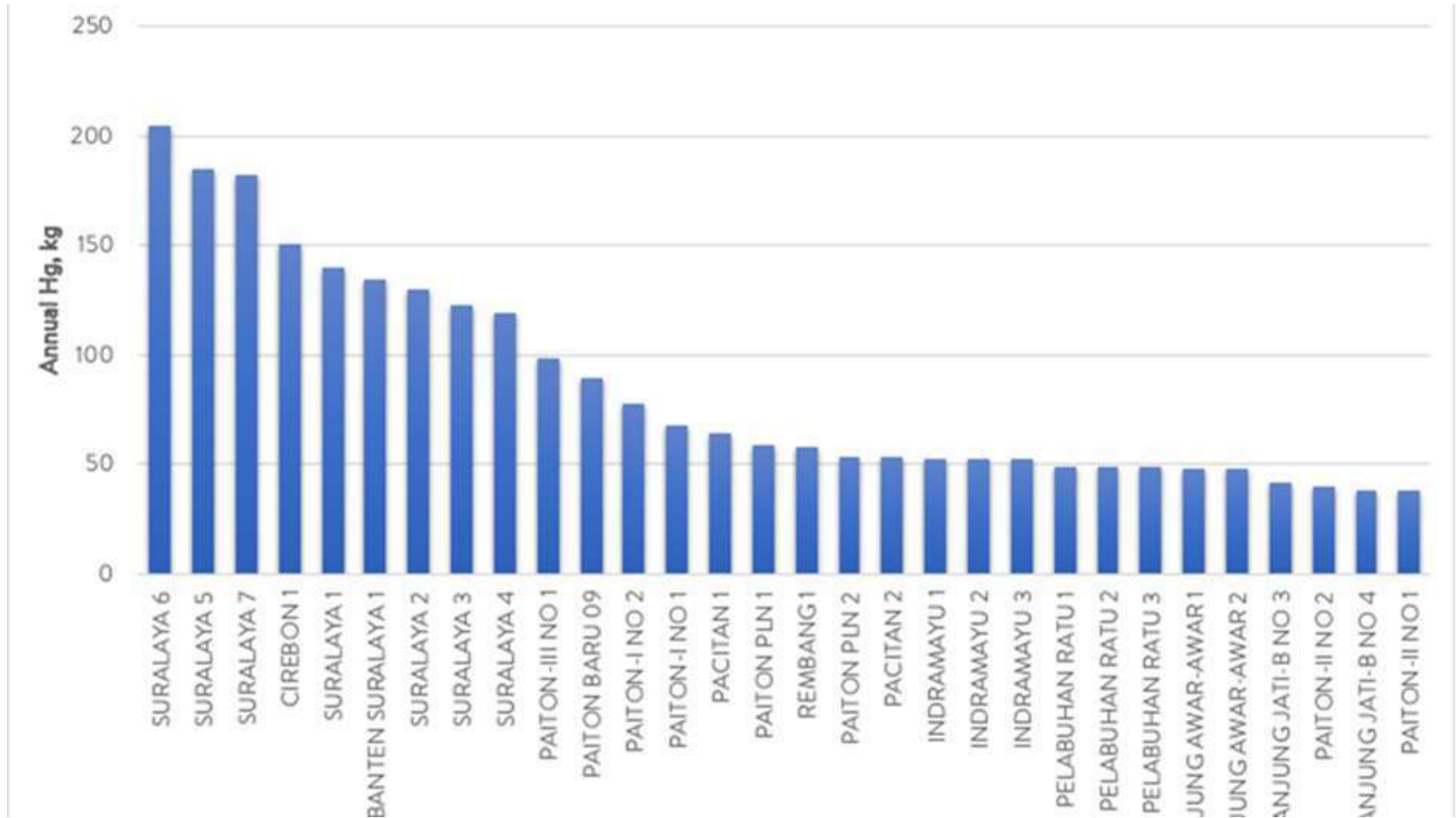
Add in capacity factor/remaining lifetime

Removes older plants which will slow down or close soon.

Allows focus for intervention on plants where control technologies may be effective in the long-term

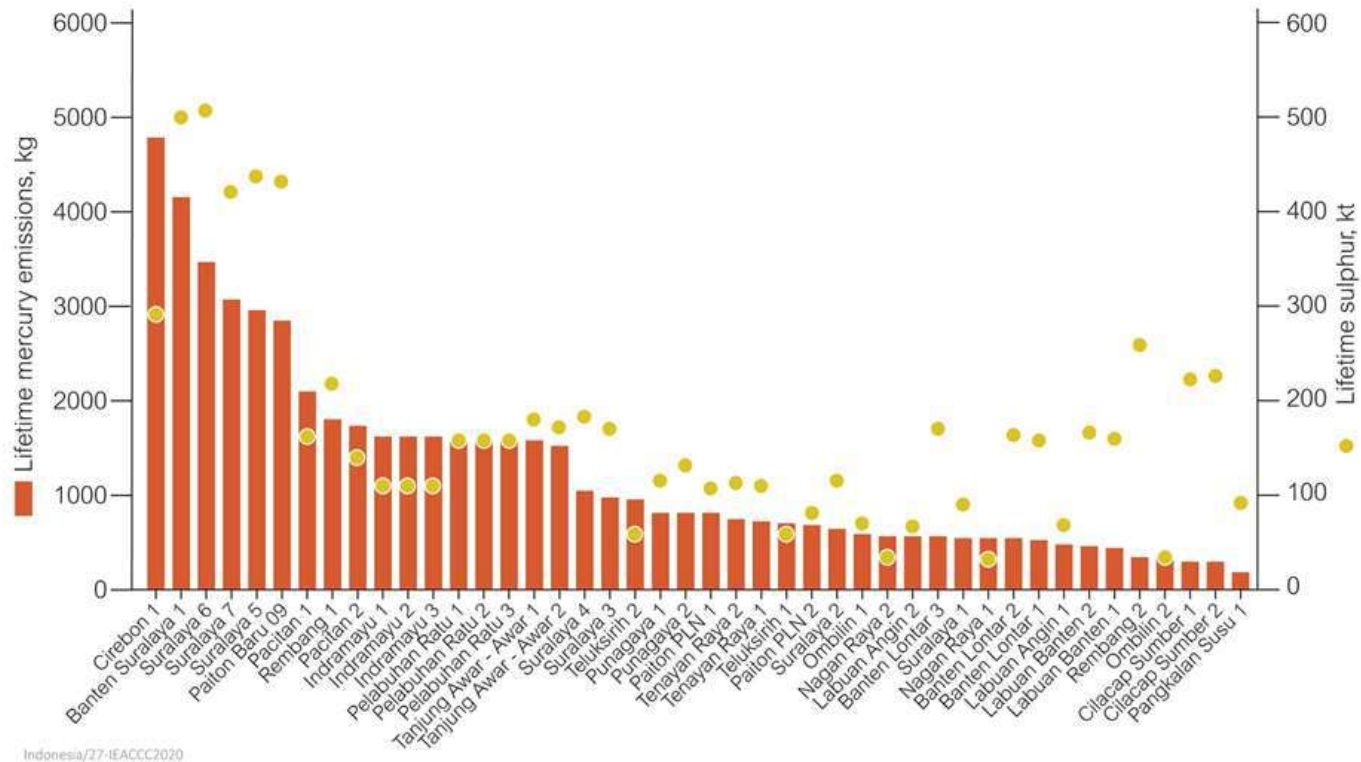
Units which will emit >1t Hg over their remaining lifetime (Indonesia)

ASSUMING PLANTS RUN UNTIL THEY ARE 40 YEARS OLD



The top 10 units in Indonesia (out of 111 units) emit around 50% of the total emissions from the entire fleet

Mercury emissions over remaining fleet lifetime



- Over 110 units analysed
- The top 10 units emit around 50% of the total emissions from the entire fleet

This provides valid science for an informed and strategic emission reduction strategy

Simple method to rank data

TOP 10 UNITS >15 years old														
UNIT NAME	CAPACITY MW	Remaining life as of 2020 (40yr life)	Operational load	Fuel Consumption (calculated)	Gross unit efficiency	SOx control (WEPP)	Coal Hg content	Coal S content	Coal Cl content	Annual Hg Emissions, coal input, kg	Hg Emissions Intensity, g/MWh	Annual Hg Emissions, iPOG prediction, kg	Remaining Plant Life Hg Emissions, kg	Total Score
PLSU1	4.0	2.0	5.0	3.0	4.0	5.0	4.0	5.0	3.0	5	4.0	5	4	53.0
PLSU2	4.0	2.0	5.0	3.0	4.0	5.0	4.0	5.0	3.0	5	4.0	5	4	53.0
PLSU3	4.0	2.0	5.0	3.0	4.0	5.0	4.0	5.0	3.0	5	4.0	5	4	53.0
PLBA	4.0	3.0	5.0	3.0	4.0	5.0	4.0	5.0	3.0	4	3.0	4	5	52.0
PLCI	4.0	3.0	5.0	3.0	4.0	5.0	3.0	2.0	3.0	5	3.0	5	5	50.0
PLTE1	1.0	4.0	5.0	1.0	5.0	5.0	5.0	2.0	5.0	3	5.0	3	2	46.0
PLTE2	1.0	4.0	5.0	1.0	5.0	5.0	5.0	2.0	5.0	3	5.0	3	2	46.0
PLPA1	2.0	3.0	5.0	3.0	4.0	5.0	4.0	3.0	2.0	4	2.0	4	4	45.0
PLPA2	2.0	3.0	5.0	3.0	4.0	5.0	4.0	3.0	2.0	4	2.0	4	4	45.0
PLRE	2.0	3.0	5.0	3.0	2.0	5.0	3.0	3.0	5.0	4	2.0	4	4	45.0

Creating a reduction strategy for coal



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BAT/BEP options

Maximising co-benefit

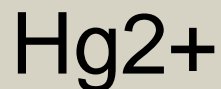
Two major forms of mercury

THE CHEMISTRY IS AFFECTED BY COAL TYPE, ASH CONTENT, CHLORINE CONTENT ETC – IT IS COMPLEX!



Oxidised mercury

- Soluble and sticky
- Easy to capture in solutions, ash or sorbents

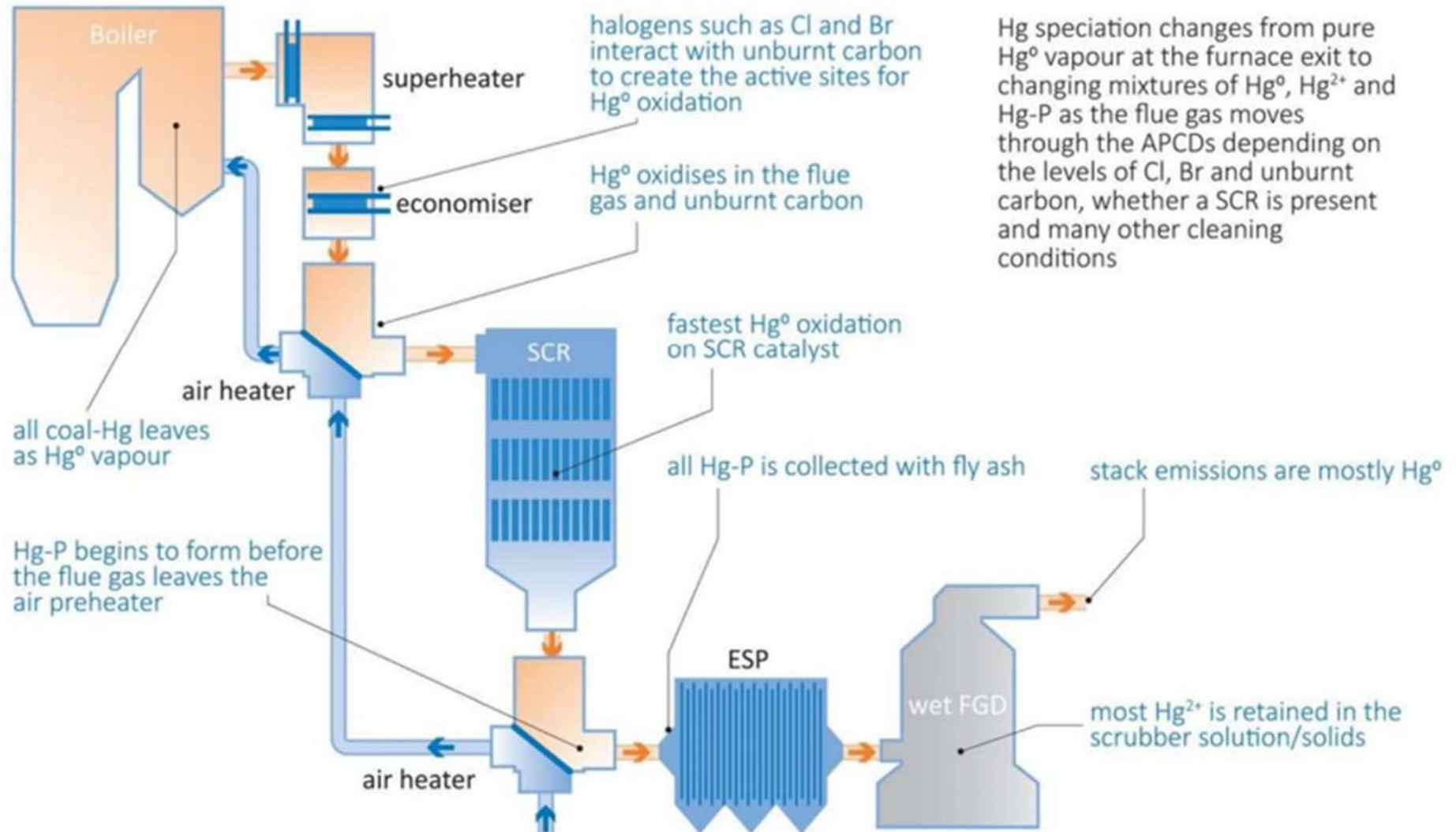


Elemental mercury

- Not soluble and not sticky
- Hard to capture
- Can be oxidised by chemicals such as chlorine and bromine



Mercury flow through a coal plant



“Co-benefit effects”

MAXIMISING “FREE” MERCURY CONTROL

If you can control mercury, you can also control acid gases and particulates

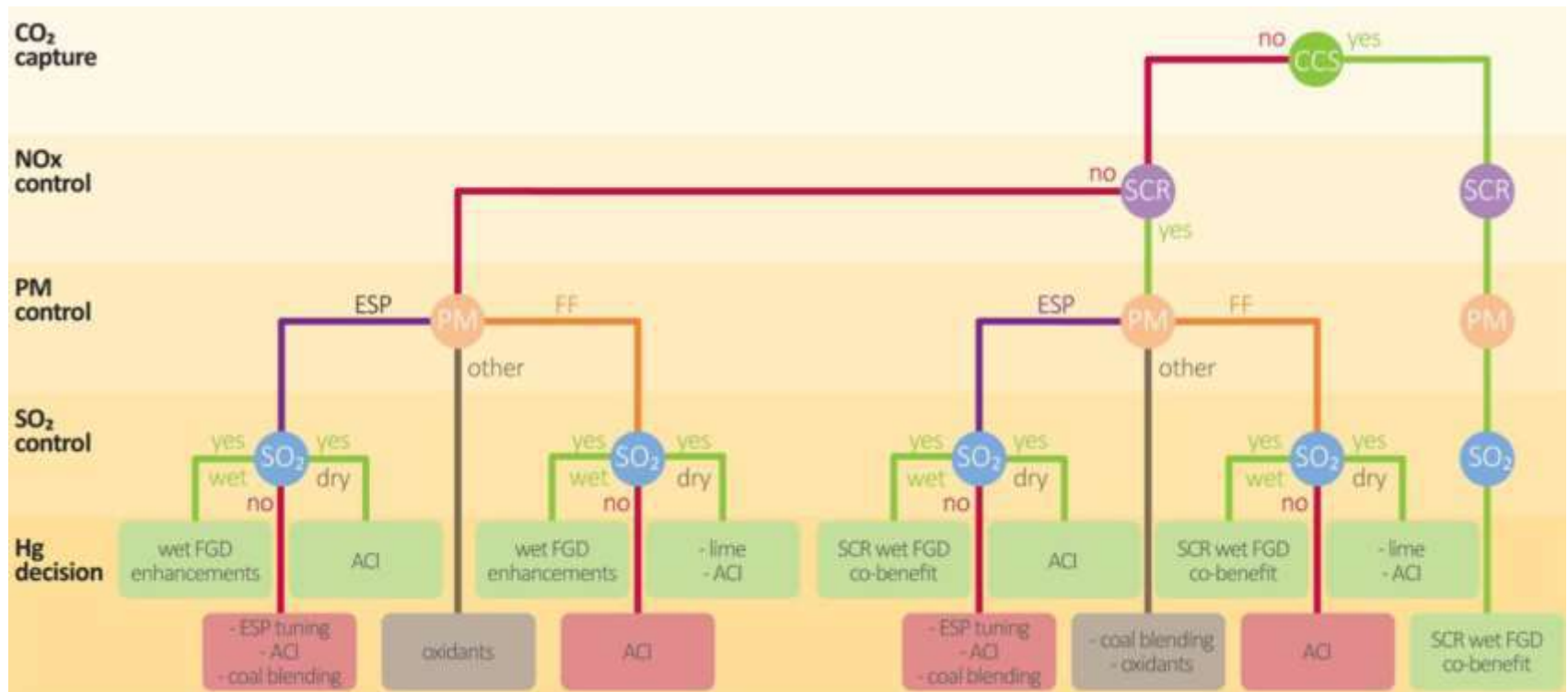
and

if you control acid gases and particulates, you also control mercury



Flow chart for technology selection

INCLUDED IN THE UNEP BAT/BEP GUIDANCE FOR COAL



Coal cleaning* and blending

*CHEMICAL COAL CLEANING HAS YET TO PROVE COST-EFFECTIVE FOR MERCURY CONTROL

Selecting coal type

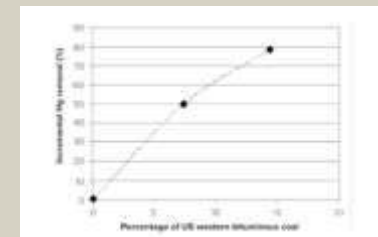
- US sub-bituminous coals tend to contain less chlorine and can be high in calcium
- Many US plants firing sub-bituminous coals found mercury reduction a challenge as most mercury is produced in the elemental form
- Oxidation with halogen addition was proven to work, but so was coal-blending

Blending

- Coal plants blend coals to maintain the characteristics required for efficient combustion
- Low-quality coals can be mixed with higher-quality coals to keep costs down
- Low sulphur coals can be mixed with high sulphur coals to keep emissions down
- Coal blending for emission control of anything other than sulphur is not a common strategy but theoretically it is possible

Strategic blending

- Study in a US plant firing sub-bituminous coal – mercury emissions remained high, even though the plant was fitted with a flue gas desulphurisation system
- Blending with bituminous coal helped to oxidise the mercury
- By blending in 15% bituminous coal in with the sub-bituminous coal, mercury emissions could be reduced by up to 80%



Co-firing biomass

- Most vegetation for co-firing will be low in mercury content. Reducing the mercury input in the total fuel will reduce the mercury input to the plant and thus reduce overall emissions
- The chlorine and ash contents of biomass can be higher than coal. This can help mercury oxidation and capture.

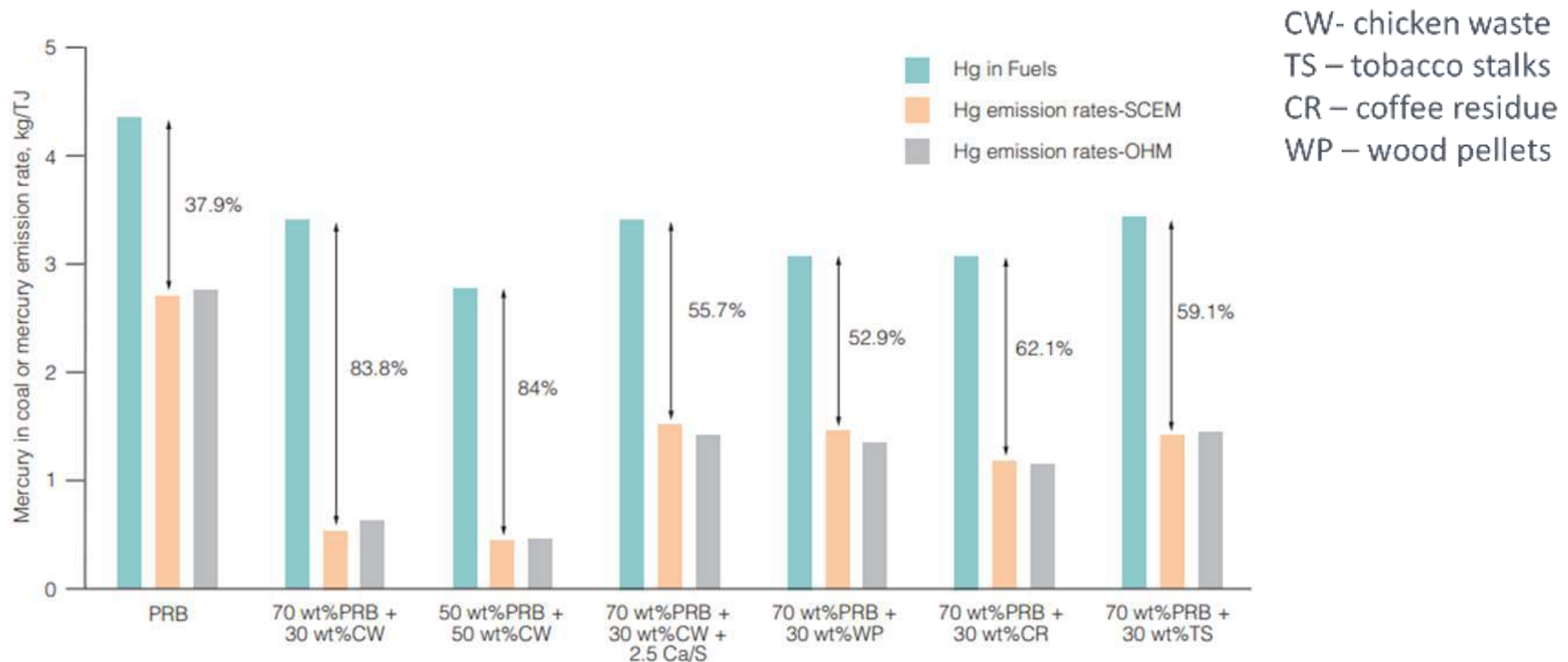


Figure 15 Variation of mercury emission during cofiring of subbituminous coal and biomass (Cao and others, 2008)

Particulate controls and mercury

VARIES WITH COAL AND PLAN TYPE

Particulate control systems can reduce PM emissions by >99.99%

Particulate control systems can capture mercury – oxidised mercury will stick to unburnt fly ash (sorbents can be added)

Mercury capture in ESP is generally lower (10-30%) than in fabric filters/baghouses (40-70%)

Emission values must be established for each site, due to potential variations in coal chemistry



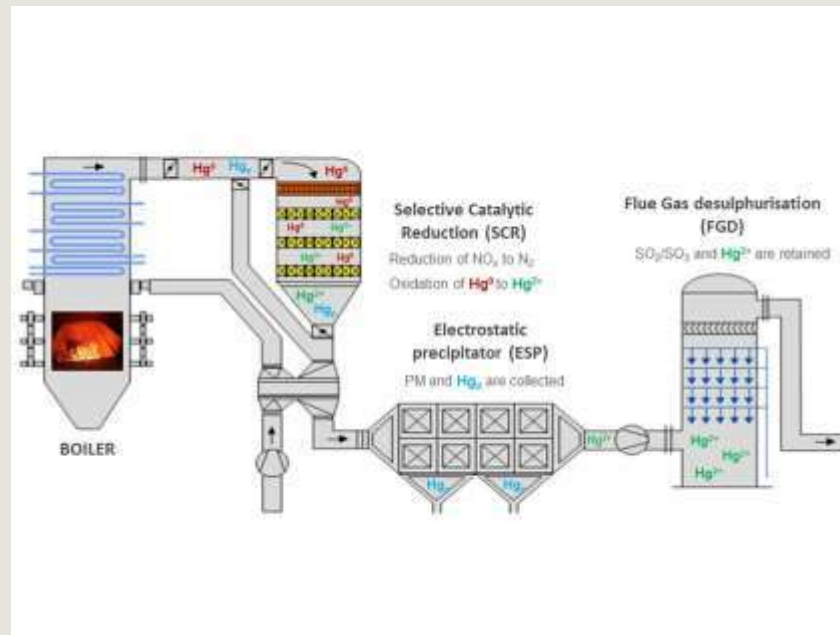
NOx controls and mercury

VARIES WITH COAL AND PLANT TYPE

NOx burners do not have a significant effect on mercury emissions

Selective catalytic reduction technologies fitted upstream of particulate controls can oxidise mercury and lead to increased mercury capture in the ash

BUT: Mercury can contaminate and shorten the life of catalysts.



Sulphur controls and mercury

VARIES WITH COAL AND PLANT TYPE

IF mercury is in the oxidised form, it will be trapped in most FGD systems:

- Wet FGD systems will dissolve oxidised mercury
- Dry FGD systems will capture oxidised mercury in the dry sorbent
- Seawater FGD systems will dissolve oxidised mercury but may release it into the local water body

Mercury capture in any FGD system can be enhanced by converting elemental mercury to oxidised mercury by adding an oxidant such as bromine

Mercury-specific control options



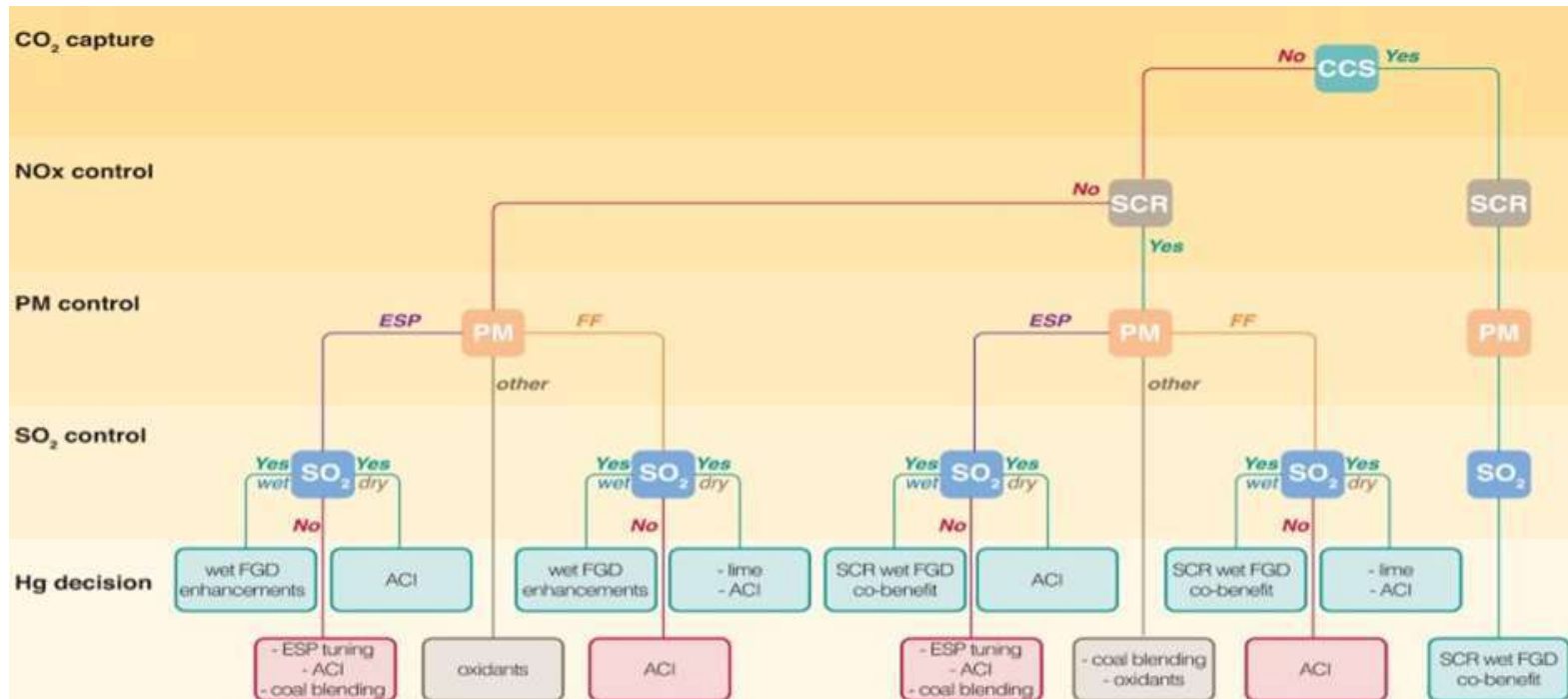
Many mercury-specific control systems have been developed and some are commercialised

Most plants see these systems as a “last-resort” to reducing emissions due to the cost

Table 7 Multi-pollutant control technologies

System	Format	Demonstration status	Marketed by
WESP	Wet ESP	Full scale at many plants	Various
COHPAC™	ESP plus fabric filter or pulse-jet fabric filter	1700 MW installed on coal plant and waste to energy incinerators	EPRI, via Babcock and Wilcox, Hamon Research-Cottrell
TOXECON™	Sorbent, and pulsed-jet fabric filter (COHPAC plus sorbent)	Fitted in 8 plants in USA	EPRI, via Babcock and Wilcox, Hamon Research-Cottrell
EFIC, electrostatic fabric integrated collector	Similar to COPAC with pulse-jet fabric filter	50 units currently in operation	China Fujian Longking
ESFF, ESP-FF hybrid system	Split level filters either integrated or separated	3 plants in China and 1 in India	Zhejiang Feida Environmental Science and Technology Co
ECO™ Technology	Dielectric barrier discharge, ammonia based scrubber, and WESP	Slip-stream demonstration	Powerspan
ReACT™	Regenerative activated coke technology	Full scale – Isogo, Japan; Weston, USA; industrial plants in Germany	J-Power, Haldor Topsoe
SNOX™	Dry catalyst/reactors with ammonia addition	Full scale, Nordjyllandsværket, Denmark, plus industrial sites	Haldor Topsoe
SNRB™ (SOX-NOx-Rox-Box)	Alkali sorbent injection and high temperature fabric filter	Demonstration	Babcock and Wilcox
Airborne™ Process	Sodium bicarbonate injection with wet sodium scrubbing and oxidation	Pilot and small scale	Airborne Clean Energy
Neustream™ Technology	Dual-alkali FGD with upstream ozone injection	Pilot scale	Neumann Systems Group
Gore mercury and SO ₂ control modules	Passive, modular, fixed absorption media modules	2100 MW installed in coal-fired power plants in the USA and demonstration pilots in European plants	Gore
Skymine™ Process	Electrochemical sodium hydroxide scrubbing	Pilot scale	Skyonic Corporation
Tri-Mer™	Modular ceramic catalyst and oxidant units	Pilot scale	Tri-Mer

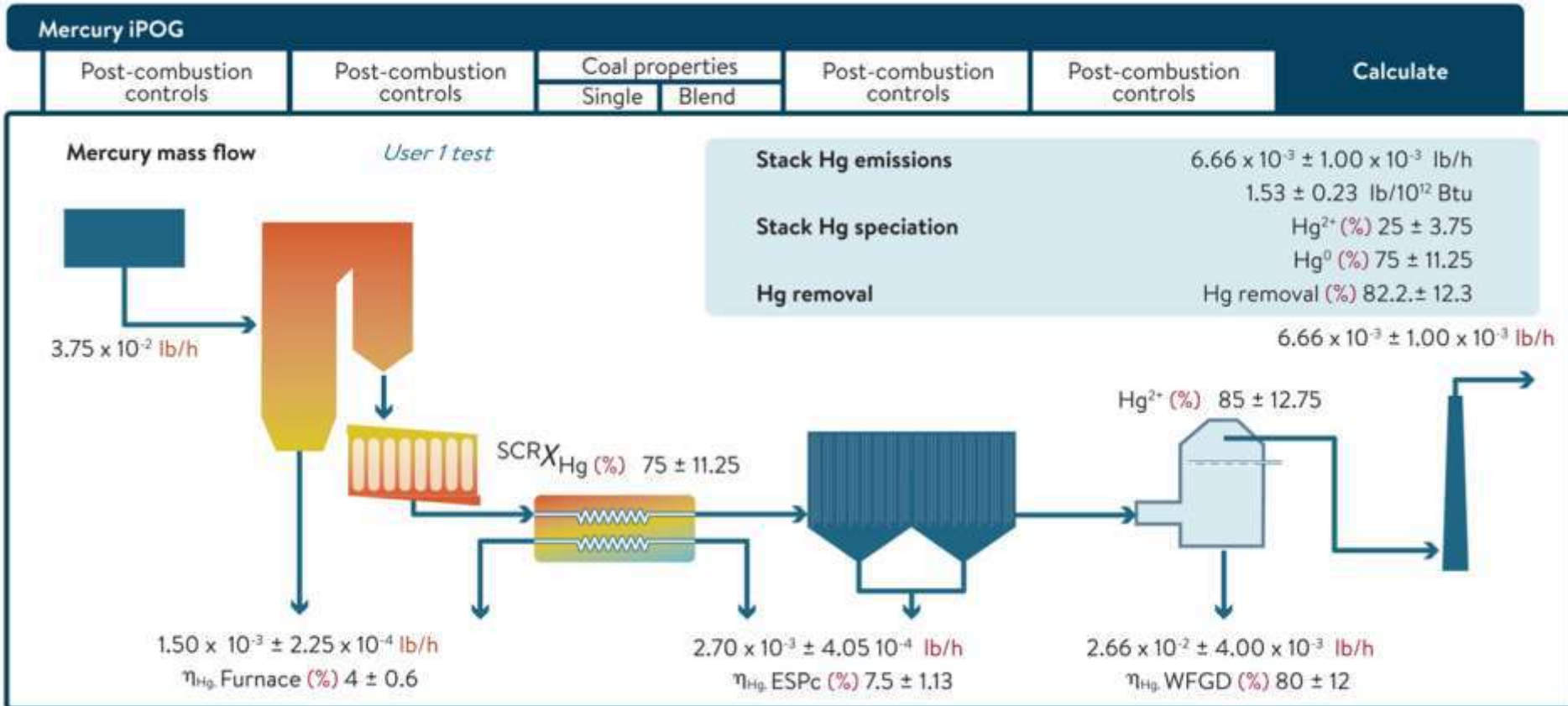
Decision tree



This is a simple flow diagram which allows the user to work through the BAT/BEP (best available technology/best environmental practice) to choose an option which will work best with different plant configurations.

Using the iPOG as a predictor

THE IPOG CAN HELP DETERMINE THE APPROACHES MOST LIKELY TO SUCCEED



Conclusions

Information and data = power

- The Minamata convention only requires a total sectoral inventory. However, an enhanced inventory could inform a significantly more cost-effective reduction strategy
- Creating an enhanced inventory takes time but, once established can simply be updated annually to monitor trends in emissions
- Use a ranking approach, considering plant-specific factors including remaining operating lifetime, to determine where action will achieve the greatest results
- It is possible and even likely that acting on a few plants could achieve faster and more cost-effective emission reduction than a blanket requirement for action across all plants



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MERCURY FROM THE NON-FERROUS SECTOR

Peter Nelson

School of Natural Sciences
Macquarie University
Sydney, Australia

Co-lead UN Environment Mercury in Coal Combustion Partnership

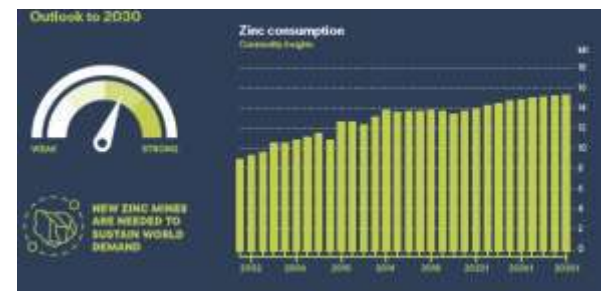
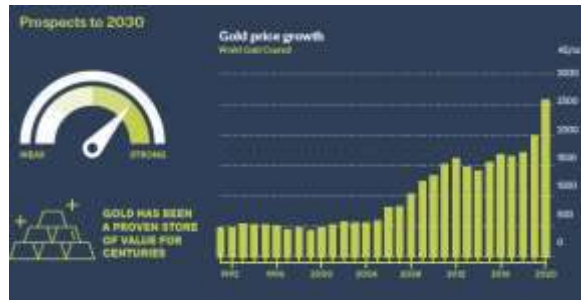
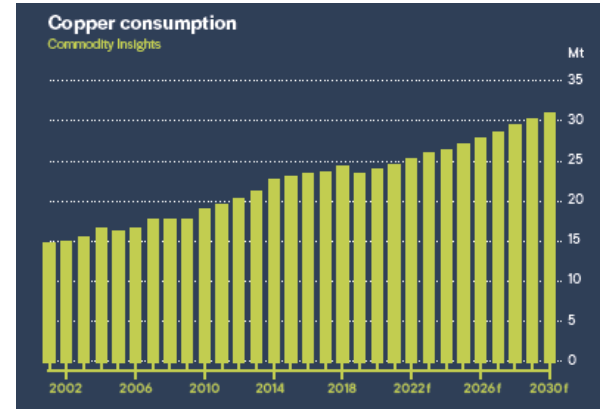
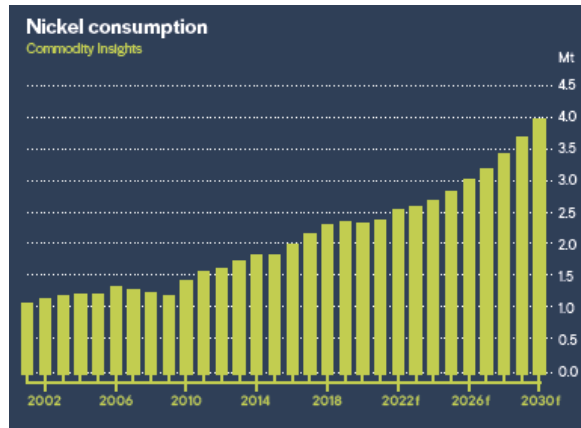
Image: <https://www.mining.com/wp-content/themes/miningdotcom/images/favicon/apple-icon-57x57.png>

Non Ferrous Metals in Article 8 Minamata Convention

- Smelting and Roasting only
- Metals:
 - Copper
 - Lead
 - Zinc
 - Industrial Gold



Non Ferrous Metals – Strong Growth



Emission Estimation

Handwritten mathematical notes and diagrams on a chalkboard background. The notes include:

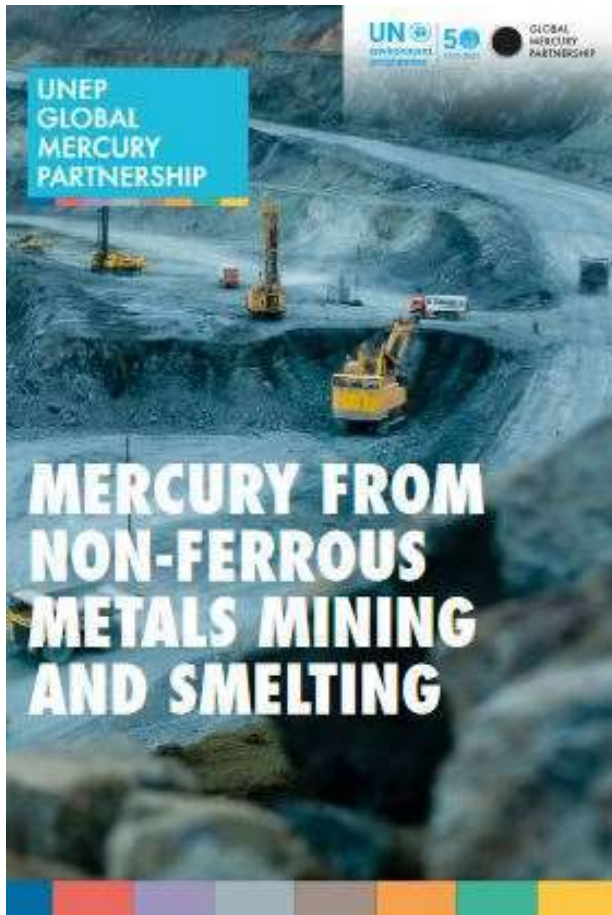
- Equations: $D(x) = 2 + 3 + 4.31447$, $\sqrt{a^2 + b^2} = x^2$, $x^2 + y^2 = ab + 4c$, $c(x, y) = \begin{cases} xy = 2 \\ cx - cy = 25^2 \\ 2\pi = c \end{cases}$, $24 \frac{x}{y} + \frac{a^2 + b^2}{c} + \frac{1}{x} = 9$, $x = 9.22$, $\sum_{x=2}^{u=14} N_{30} \cdot x$, $x \leq 549$, $\frac{1}{2} [984 + x + p + ab]$, $\beta = 9 + x^2 + y^2$.
- Diagrams: A 3D cube, a circle with a shaded sector, a coordinate system with a curve, a graph of a bell-shaped curve, a grid with shaded cells, and a box containing binary code.
- Other markings: Circled numbers (10, 22, 55), a circled 'A', and various arrows and annotations.



Tools for Inventory Development

Inventory Level 2 (IL2)

- a detailed mercury inventory tool
- all factors adjustable to national or local conditions.
- default estimation factors are pre-entered
- requires more reading and experience
- high level of accuracy, provided that the data needed for this are available



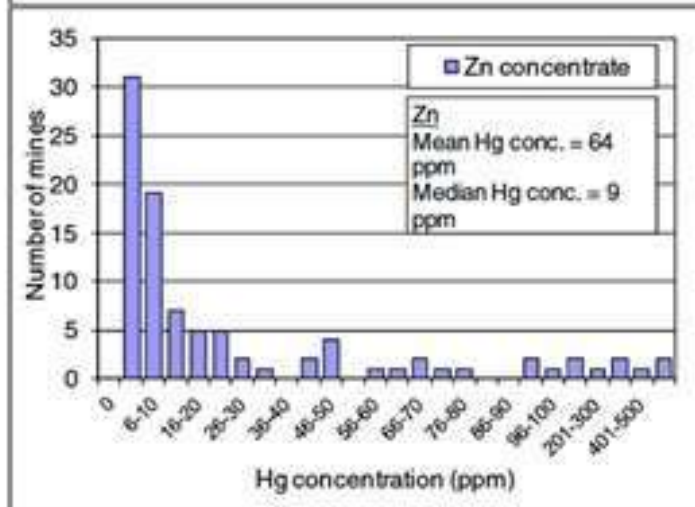
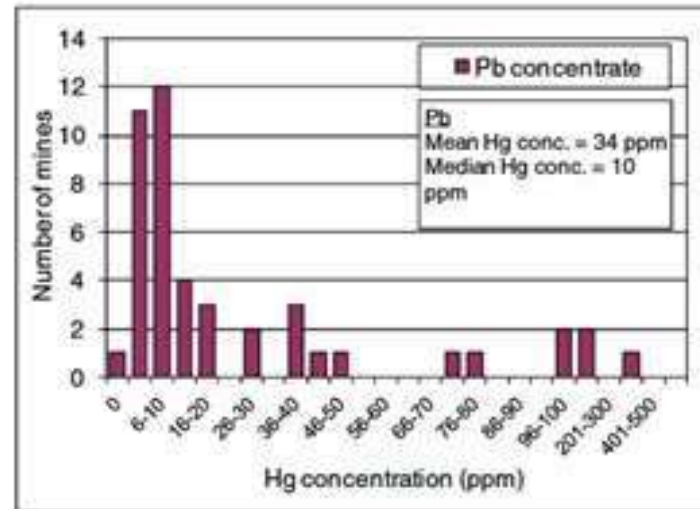
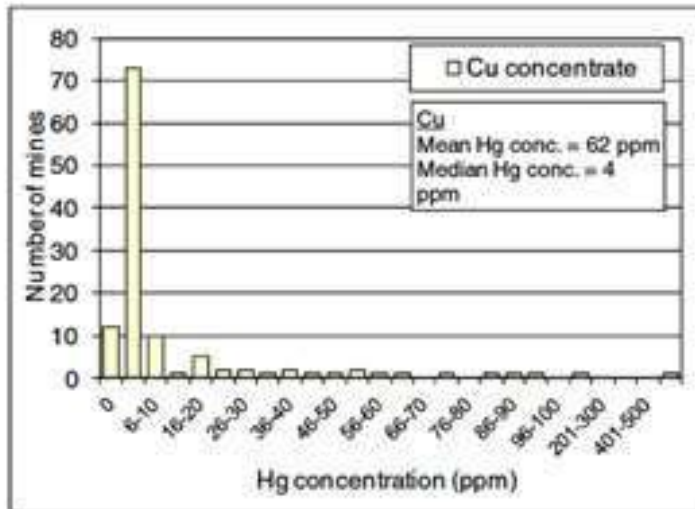
Other resources – Study Report on Non Ferrous Metals

Identified uncertainties and knowledge gaps

- Hg content in ores and concentrates, at plant and country level
- Hg air emissions test data
- Hg concentrations in reject material
- Hg distributions between emissions and other releases
- Activity data (amounts of ores and concentrates processed)
- Effects of pollution control technologies, incl. on distribution of Hg between emissions to air, and capture in solid and liquid waste
- Additional quantitative information on how mercury departs to emissions and releases to air, land, water, waste and by-products

DATA REQUIRED FOR BETTER EMISSION ESTIMATES

Mercury variability in ores



Number of mines and the reported Hg concentrations in

- Cu concentrates
- Pb concentrates
- Zn concentrates

Improving emissions estimations



Improved data (mercury in ore and concentrates, activity data, control technologies and their effectiveness,...)



Individual plant data (but a large task; 70 gold mines are in South Africa, according to GlobalData's mines database)



Harmony Gold Mining, Anglo Gold Ashanti, and Gold Fields made up about 50% production in 2021

Better understanding of ore characteristics

Gold is typically recovered from ores containing only traces of the metal - main challenge is concentrating

- Techniques:
 - Cyanide leaching; gold must be available for leaching
 - Mercury amalgamation – largely now only used in ASGM
 - Refractory ores – hard to leach ultra-fine mercury; requires pre-treatment (roasting, oxidation, ...)



Reducing mercury emissions



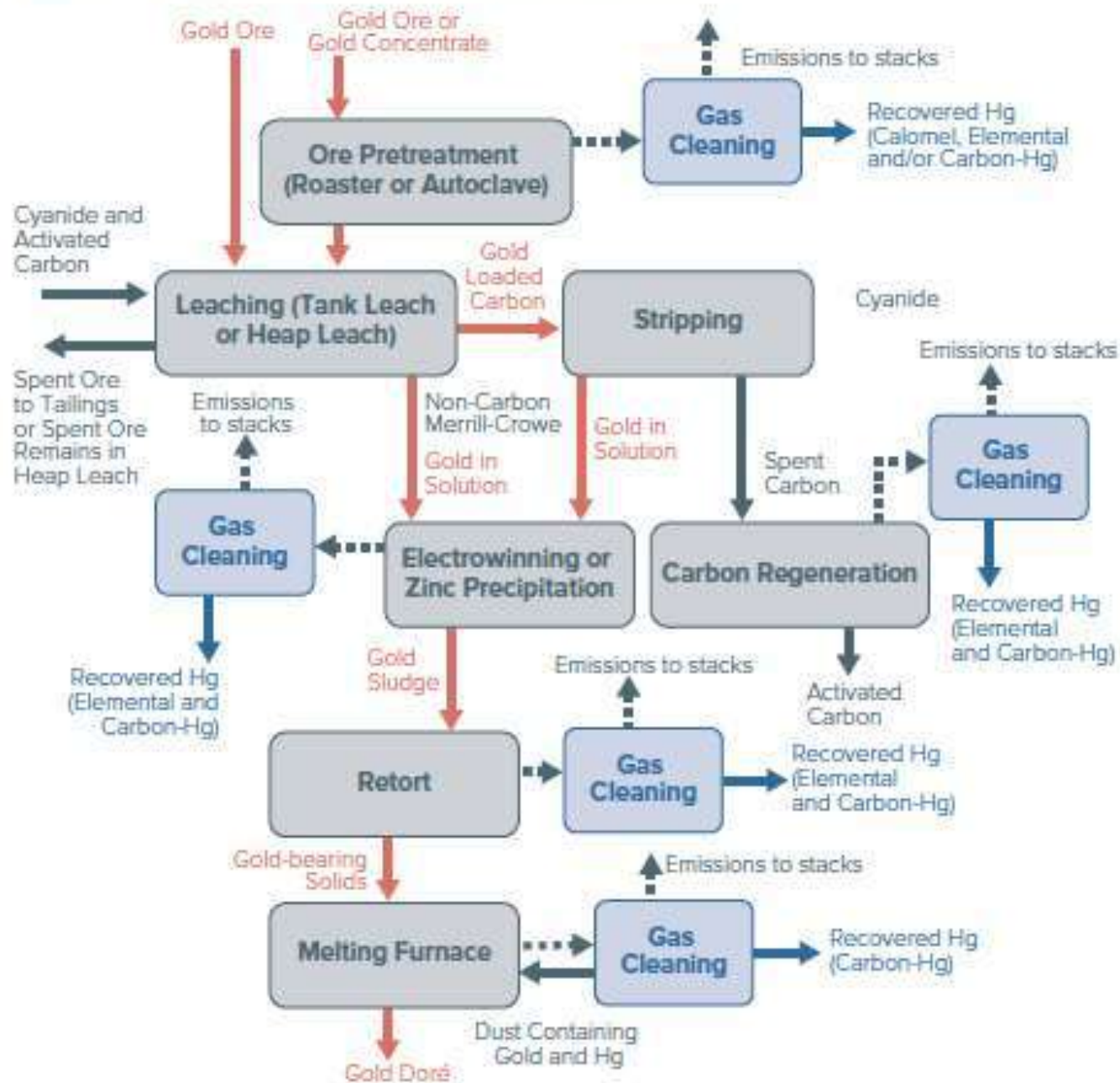
BAT/BEP

Reduction of Hg emissions

- Boliden-Norzink process
 - $\text{Hg} + \text{HgCl}_2 \rightarrow \text{Hg}_2\text{Cl}_2$
(calomel)
- Selenium filter
 - $\text{Se} + \text{Hg} \rightarrow \text{SeHg}$
- Activated carbon
- Co-benefits of air pollution abatement technologies
 - Particulate matter, SO_2 , NO_x

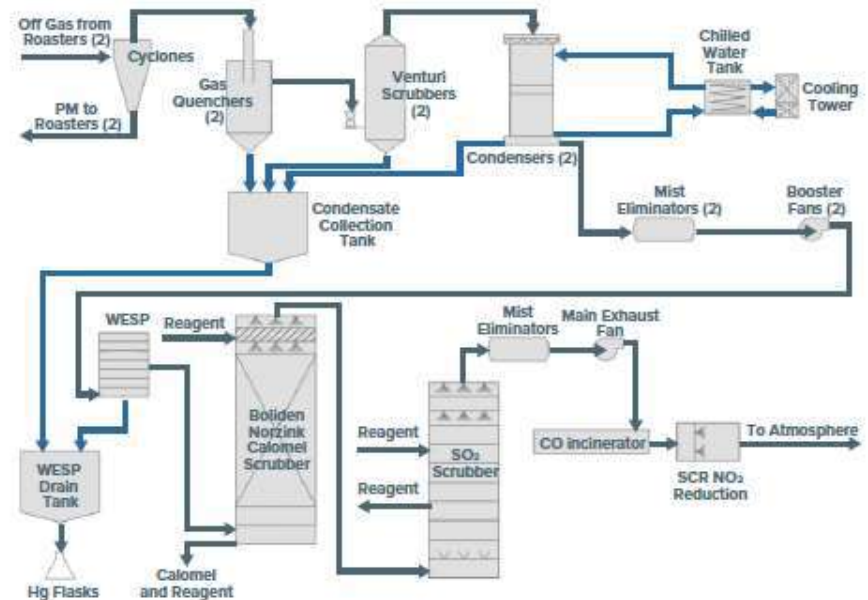


FIGURE 1: EXAMPLE OF GOLD PRODUCTION PROCESSING*



Case Study: Nevada Gold Plant

- Controls employed:
- Cyclone separation
- Gas Quench
- Venturi gas scrubbing
- Gas condenser
- Wet electrostatic precipitator (ESP)
- Calomel scrubber



Mercury Removal Technology	Process Conditions	Advantages	Disadvantages
Carbon Filter beds	Efficiency = 99%	<ul style="list-style-type: none"> Effectively removes mercury chloride 	<ul style="list-style-type: none"> Untreated carbon ineffective in removing elemental mercury
Fixed activated carbon filter beds	Efficiency = 90%	<ul style="list-style-type: none"> Sulfur-impregnated activated carbon is commercially available Removes Hg⁰ and other species Low potential for leaching of mercury from spent carbon 	<ul style="list-style-type: none"> Spent carbon requires disposal in landfill
Activated carbon injection	Efficiency = 90-95%	<ul style="list-style-type: none"> Sulfur-impregnated activated carbon is commercially available Removes Hg⁰ and other species Low potential for leaching of mercury from spent carbon 	<ul style="list-style-type: none"> Spent carbon requires disposal in landfill
Lime/limestone scrubbing	Efficiency = 10-84%	<ul style="list-style-type: none"> Effective for water soluble species 	<ul style="list-style-type: none"> Ineffective for elemental mercury Wastewater requires treatment prior to disposal
Selenium filters	Efficiency = 99.6% Max Hg _{IN} = 9 mg/m ³ Max Hg _{OUT} = 40 µg/m ³	<ul style="list-style-type: none"> Successful installation at metallurgical plants 	<ul style="list-style-type: none"> Limited inlet mercury concentration Ineffective for species other than elemental mercury Spent filter requires disposal in landfill
Boliden-Norzink process	Efficiency = 99% Max Hg _{IN} = 5-80 mg/m ³ Max Hg _{OUT} = 20-50 µg/m ³	<ul style="list-style-type: none"> Widely demonstrated Mercury removed as marketable product 	<ul style="list-style-type: none"> Removes only elemental mercury Complicated flowsheet Chlorine gas handling



Introduction: Minamata Convention on Mercury, Article 8, emissions inventories

Workshop to enhance inventories and strategies under Article 8 of
the Minamata Convention in South Africa, 31 May 2024

Alexander Romanov, UNEP-GEF Chemicals and Waste (alexander.romanov@un.org)
on behalf of the Secretariat of the Minamata Convention on Mercury



Minamata Convention of Mercury



- Objective: to **protect the human health and the environment from anthropogenic emissions and releases of mercury and mercury compounds.**
- Adopted in October 2013, entered into force in August 2017.
- Mercury is a chemical of global concern owing to its:
 - Long-range atmospheric transport,
 - Persistence in the environment once anthropogenically introduced,
 - Ability to bioaccumulate in ecosystems, and
 - Significant negative effects and human health and the environment.
- Recognizes the lessons of Minamata Disease, in particular the serious health and environmental effects from mercury pollution.

See [Minamata Convention at a Glance](#)

UN environment
Global Action Programme

MINAMATA CONVENTION ON MERCURY
FACT SHEET
www.mercuryconvention.org

AT A GLANCE:
MINAMATA CONVENTION ON MERCURY

Why develop an international treaty on mercury?

The Minamata Convention on Mercury was the first new global Convention on environment and health adopted for close to a decade. It is named after the place in Japan where, in the mid-20th century, mercury-tainted industrial wastewater poisoned thousands of people, leading to crippling symptoms that became known as the "Minamata disease".

Mercury is a highly toxic heavy metal that poses a global threat to human health and the environment. Together with its various compounds, it has a range of severe health impacts, including damage to the central nervous system, thyroid, kidneys, lungs, immune system, eyes, gums and skin. Victims may suffer memory loss or language impairment, and the damage to the brain cannot be reversed. There is no known safe exposure level for elemental mercury in humans, and effects can be seen even at very low levels. Fetuses, newborn babies and children are amongst the most vulnerable and sensitive to the adverse effects of mercury. Mercury is transported around the globe through the environment, so its emissions and releases can affect human health and environment even in remote locations.

No country can control transboundary effects of mercury alone. It can be effectively tackled only through international cooperation. With the adoption of the Minamata Convention, Governments from around the world have taken a major step in dealing with worldwide emissions and releases of mercury, which threaten the environment, and the health of millions.

Why is mercury present in our environment and how are we exposed to it?

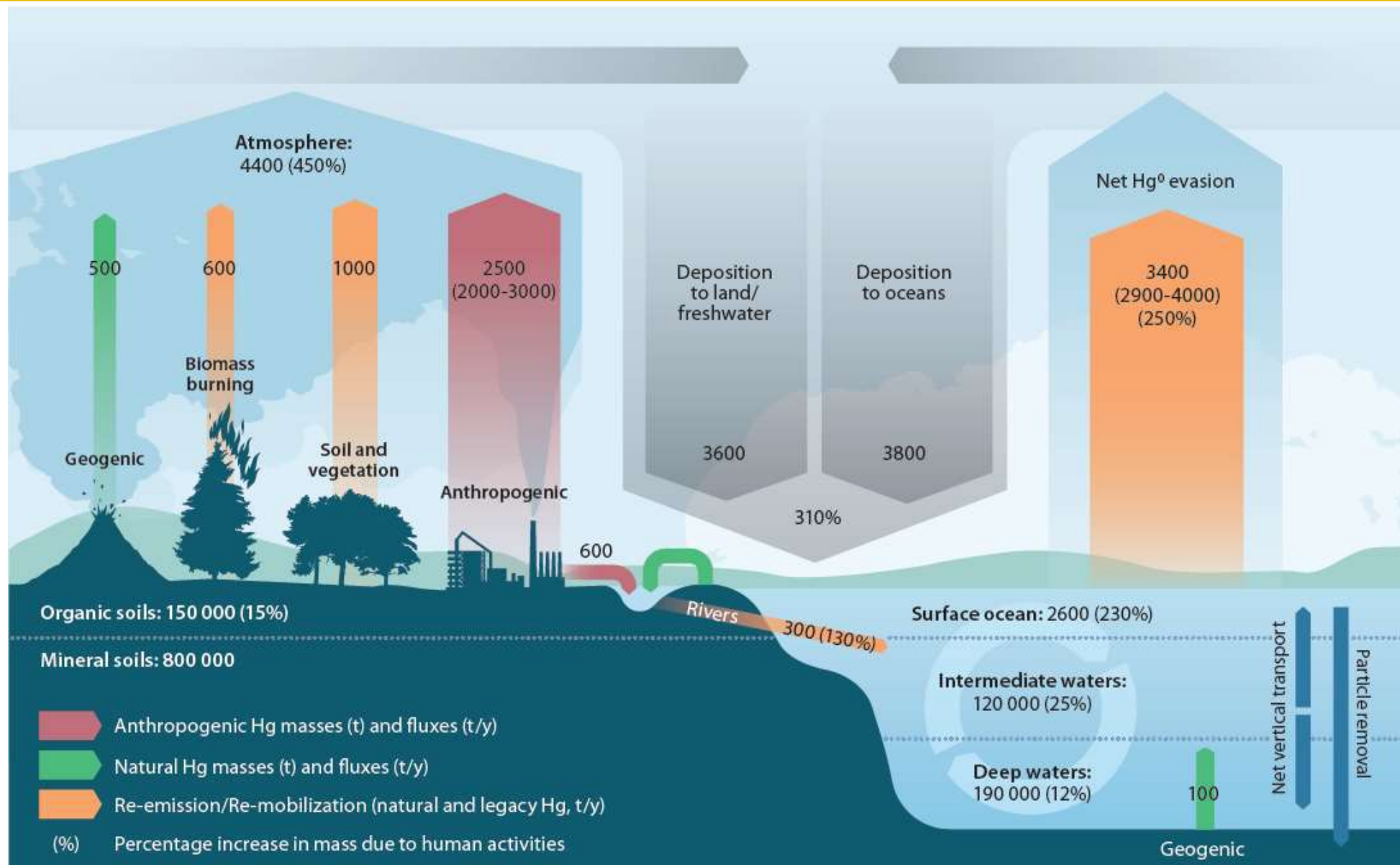
Mercury is a naturally occurring element. It can be released to the environment from natural sources – such as weathering of mercury-containing rocks, forest fires, volcanic eruptions or geothermal activities – but also from human activities. Of the estimated 5500-8000 tons of mercury currently emitted and re-emitted each year to the atmosphere, only about 10 per cent is accounted to be from natural sources¹.

Due to its unique properties, mercury has been used in various products and processes for hundreds of years. Currently, it is mostly utilised in industrial processes that produce chlorine and sodium hydroxide (mercury chlor-alkali plants) or vinyl chloride monomer for polyvinyl chloride (PVC) production, and polyurethane elastomers. It is extensively used to extract gold from ore in artisanal and small-scale gold mining. It is contained in products such as electrical switches (including thermostats), relays, measuring and control equipment, energy-efficient fluorescent light bulbs, batteries and dental amalgam. It is also used in laboratories, cosmetics, pharmaceuticals, including in vaccines as a preservative, paints, and jewellery.

¹ UNEP, Global Mercury Assessment 2013: Sources, Emissions, Releases, and Environmental Transport

DISCLAIMER: The information contained in this document is presented for information purposes only and does not constitute an interpretation of the text of the Minamata Convention on Mercury by UNEP or the Secretariat of the Minamata Convention. It does not substitute the original text of the Convention, as updated with the Secretary General of the UN Factoring as the Depositary, available at: <http://www.unep.org/mercuryconvention>

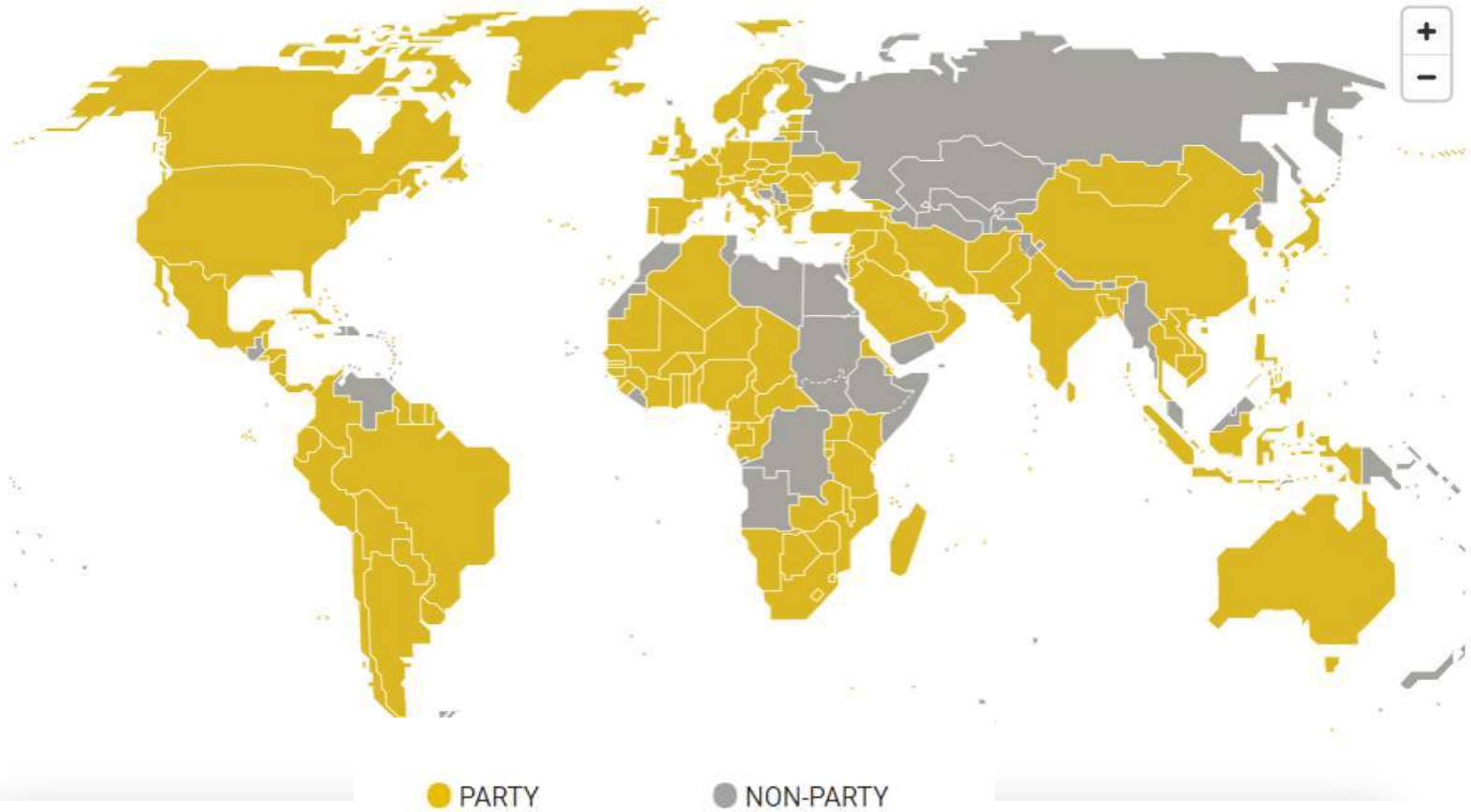
GMA 2018 - Update on global Hg pools and cycles



Parties to the Minamata Convention



▶ 148 parties as of May 2024



United Nations Treaty Collection

Dispository

17. Minamata Convention on Mercury

Kyoto, 10 October 2013

There has been 18 August 2013, in accordance with article 31(1), the Convention shall enter into force on the seventh day after the date of deposit of the fifth instrument of ratification, acceptance, approval or accession. For each State or regional economic integration organization that ratifies, accepts, approves or accedes to the Convention on a date later than the date of the fifth instrument of ratification, acceptance, approval or accession, the Convention shall enter into force on the seventh day after the date of deposit by such State or regional economic integration organization of its instrument of ratification, acceptance, approval or accession. Any instrument deposited by a regional economic integration organization that is not a party to the Convention shall be deemed to have been deposited by member States of the organization.

Registration: 18 August 2013, No. 32888

Name: Minamata Convention


Text: [CMBTreaty.docx](#)


Note: The Convention was adopted on 10 October 2013 at Minamata (based on the outcome of the Conference of Plenipotentiaries on the Minamata Convention on Mercury held from 7 to 11 October 2013). The Convention was opened for signature by States and regional economic integration organizations at Minamata, Japan, on 20 and 21 October 2013, and, following the United Nations Headquarters in New York, on 6 October 2014.

Party/State	Signature	Accession/Approval/Accession
Algeria	01 Oct 2013	27 May 2017 *
Algeria	01 Oct 2013	05 May 2020
Algeria	21 Oct 2013	
Antigua and Barbuda		01 Nov 2018 *
Armenia	01 Oct 2013	01 Sep 2017
Australia	02 Oct 2013	11 Dec 2017
Austria	01 Oct 2013	7 Dec 2010
Austria	01 Oct 2013	11 Jun 2007
Azerbaijan		22 Feb 2022 *
Bahrain		06 Jul 2021 *
Bangladesh	01 Oct 2013	
Belarus	01 Oct 2013	
Belgium	01 Oct 2013	05 Feb 2018
Belize	01 Oct 2013	7 Nov 2020
Belize (Participating State of)	01 Oct 2013	08 Jan 2024
Bhutan	01 Oct 2013	05 Jul 2018 *
Bolivia	01 Oct 2013	08 Aug 2017
Bolivia	01 Oct 2013	05 May 2021
Bosnia and Herzegovina	01 Oct 2013	01 Apr 2017
Brazil	24 Feb 2014	05 Mar 2021
Canada	01 Oct 2013	04 Apr 2011
Canada	04 Sep 2010	01 Mar 2020

For most recent list of parties, see [UN Treaties Section website](#)

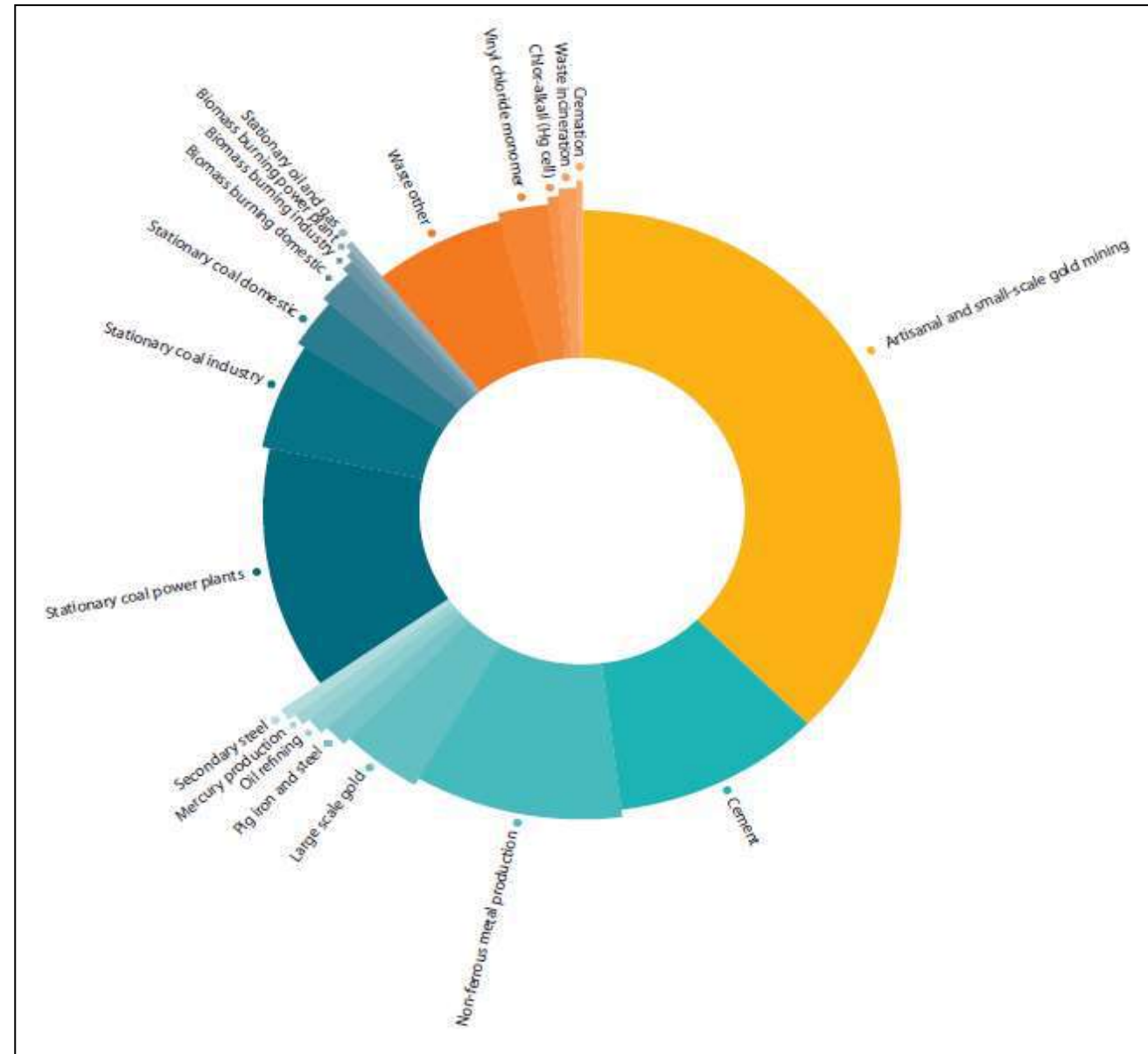
Control measures and support measures

 Control Measures		Reduce mercury to the environment
	Reduce the use and presence of mercury in the economy, industry and society	
Keep mercury underground	Art. 3.5 (a): <u>Stocks</u>	Art. 7: <u>ASGM</u>
Art. 3.3: No new primary <u>mines</u>	Art. 3.5 (b): <u>Excess mercury</u> from decommissioned chlor-alkali facilities	
Art. 3.4: Existing <u>mines</u> - 15 years	Art. 3.6 – 3.10: <u>Trade of mercury</u>	Art. 8: <u>Emissions</u>
	Art. 4: Mercury-added <u>Products</u>	
	Art. 5: Manufacturing <u>Processes</u>	Art. 9: <u>Releases</u>
	Art. 7: <u>ASGM</u>	
	Art. 10: <u>Interim Storage</u>	Art. 9: <u>Releases</u>
	Art. 11: <u>Mercury wastes</u>	
	Art. 12: <u>Contaminated sites</u>	

 Enabling / Supportive Context
Art. 13: Financial Resources and Mechanism
Art. 14: Capacity-building, technical assistance and technical transfer
Art. 15: Implementation and Compliance Committee
Art. 16: Health aspects
Art. 17: Information Exchange
Art. 18: Public information, awareness and education
Art. 19: Research, development and monitoring
Art. 20: Implementation plans
Art. 21: Reporting
Art. 22: Effectiveness evaluation
Art. 23: Conference of the Parties
Art. 24: Secretariat
Arts. 25-35: Various procedural articles

Global Mercury Assessment 2018

- The predominant source sector is artisanal and small-scale gold mining (about 38%).
- It is followed by stationary combustion of coal (about 21%), non-ferrous metal production (about 15%) and cement production (about 11%).



Article 8 of the Minamata Convention

- ▶ Controls the emissions of total mercury to air from the following sources listed in Annex D:
 - Coal-fired power plants
 - Coal-fired industrial boilers
 - Smelting and roasting processes used in the production of non-ferrous metals (lead, zinc, copper and industrial gold)
 - Waste incineration facilities
 - Cement clinker production facilities.
- ▶ Parties with relevant sources shall take measures to control emissions and may prepare a national plan, which is to be submitted **within 4 years after the entry into force** if prepared.
- ▶ For **new sources**, each Party shall require the use of BAT/BEP to control and reduce emissions, as soon as practicable but **no later than 5 years after the date of entry into force**.
- ▶ For **existing sources**, each Party shall include in any national plan, and shall implement, one or more of the following measures, as soon as practicable but **no more than 10 years after the date of entry into force**:
 - A quantified goal
 - Emission limit values
 - The use of BAT/BEP
 - A multi-pollutant control strategy that would deliver co-benefits
 - Alternative measures to reduce emissions from relevant sources
- ▶ Each Party shall establish, as soon as practicable and no later than **5 years after the date of entry into force** of the Convention for it, and maintain thereafter, an **inventory of emissions from relevant sources**.

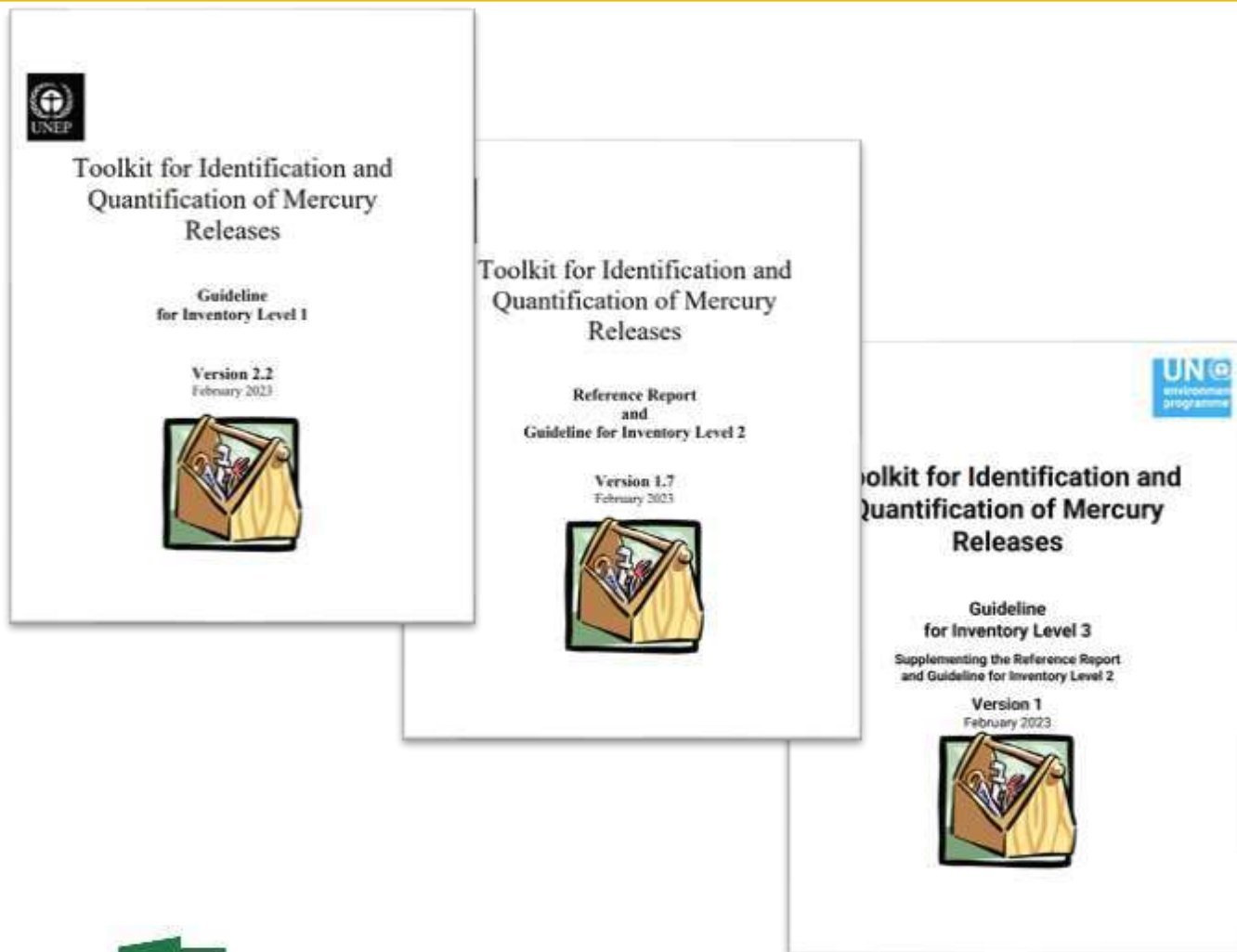
Guidance under Article 8 of the Minamata Convention

- ▶ Decision MC-1/4
 - Adopted the [guidance](#) on BAT/BEP and on support for parties in implementing the measures
 - Recognized that some of the measures described in the guidance may not be available to all parties for technical or economic reasons,
 - Requested parties with experience in using such guidance to provide the secretariat with information on that experience, and the secretariat to compile such information and to update the guidance as necessary.
- ▶ Decision MC-1/16
 - Adopted the guidance on criteria that parties may develop to identify emission sources, and on the methodology for emission inventories.



- ▶ UNEP's *Toolkit for identification and quantification of mercury releases* –aka UNEP Mercury *Toolkit* – is intended to assist countries to identify and quantify the sources of mercury emissions and releases, set priorities and reduction targets, enhance international co-operation, knowledge sharing, and **enable targeted technical assistance**.
- ▶ Inventories from countries contribute to the Global Mercury Assessment, the hub of the scientific knowledge of worldwide mercury emissions and releases.
- ▶ The Toolkit provides clear guidance on different stages of inventory development: identifying mercury sources, quantifying the consumption and calculating the final emissions and releases.
- ▶ The Toolkit includes detailed manual, calculation spreadsheet and a standard template for reporting.
- ▶ The Toolkit is one of the methods recommended in guidance from the Minamata Convention on preparing inventories of emissions pursuant to Article 8.

UNEP's Toolkit for identification and quantification of mercury releases



Inventory Level 1 (IL1) – simplified model based on default factors, requires national sectoral activity rate data; useful for first-time inventories, yet less accuracy of emission/release estimates should be expected

Inventory Level 2 (IL2) – detailed mercury inventory tool, all emission/release factors can be adjusted to national/local conditions (default factors are included), requires detailed national sectoral data to fully reflect mercury cycles

Inventory Level 3 (IL3) - integrates all mercury sources into their entire mass flow through and out of society to the environment linking different mercury sources and provides increased accuracy in estimations; most data- and expertise-intensive

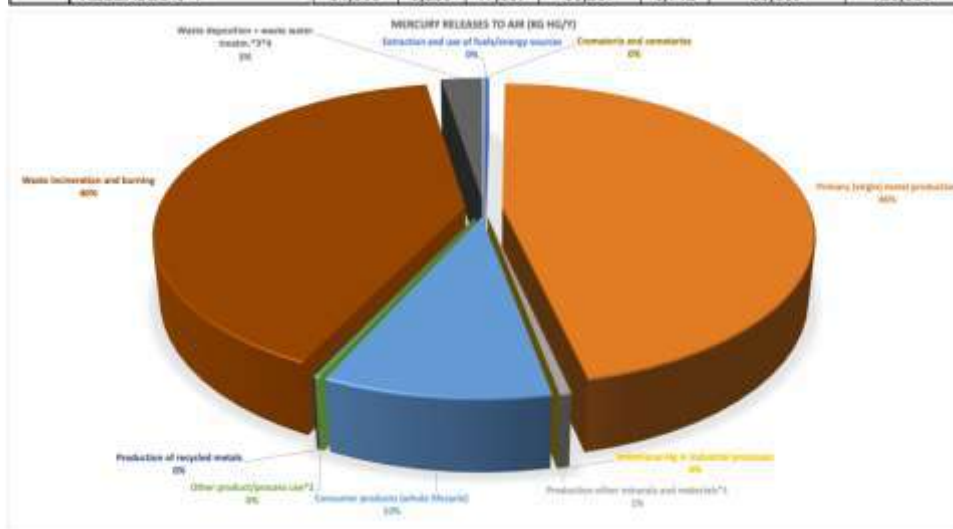


+ Excel calculations sheets for Level 1, Level 2, and Level 3 inventories

UNEP's Toolkit for identification and quantification of mercury releases

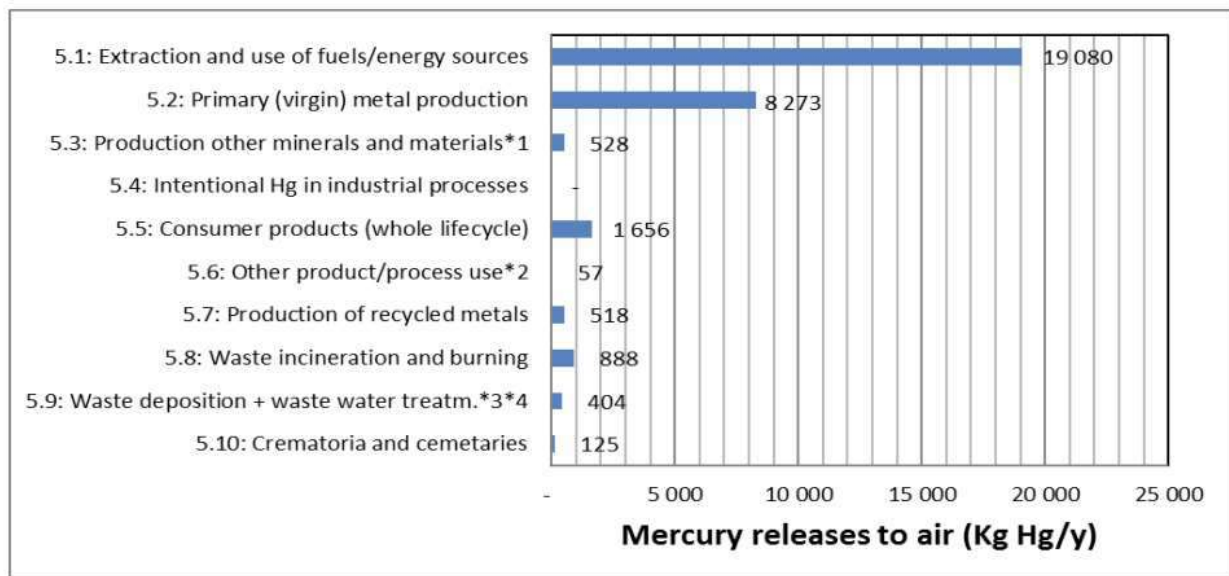
Minamata Initial Assessment Report for Zambia (2017)

Source No.	Source category	Calculated Hg output, Kg/y						Total releases by source category
		Air	Water	Land	By-products and impurities	General waste	Sector specific treatment/disposal	
1	Extraction and use of fuels/energy sources	99	0	-	-	1	5	105
2	Primary (virgin) metal production	14,705	1,719	13,976	30,196	-	33,174	93,770
3	Production other minerals and materials*1	166	-	-	71	-	-	237
4	Intentional Hg in industrial processes	-	-	-	-	-	-	-
5	Consumer products (whole lifecycle)	3,164	288	3,219	-	6,190	0	12,861
6	Other product/process use*2	68	459	3	-	394	389	1,314
7	Production of recycled metals	36	-	37	-	36	-	109
8	Waste incineration and burning	12,815	-	-	-	-	-	12,815
9	Waste deposition + waste water treatm.*3*4	813	1,165	6,258	-	121	121	8,477
10	Crematoria and cemeteries	0	-	141	-	-	-	141
SUM OF QUANTIFIED RELEASES*3*4		31,865	3,229	17,537	30,267	6,742	33,689	123,330



Minamata Initial Assessment Report for South Africa (2021)

Category	Source category	Calculated. Hg input to society(Kg/y)	Percentages of Total
5.1	Extraction and use of fuels/energy sources	38080	53.8
5.2	Primary (virgin) metal production	12894	18.2
5.3	Production of other minerals and materials with mercury impurities	803	1.1
5.4	Intentional use of mercury in industrial processes	0	0.0
5.5	Consumer products with intentional use of mercury	11726	16.6
5.6	Other intentional product/process use	4346	6.1
5.7	Production of recycled metals ("secondary" metal production)	1594	2.3
5.8	Waste incineration*3	905	1.3
5.9	Waste deposition/landfilling and wastewater treatment	408	0.6
5.10	Crematoria and cemeteries	1250	1.8



MercuryLearn Training

- In response to the increasing interest of countries to develop mercury emissions inventories and the subsequent high demand of guidance and training on this topic, UNEP and UNITAR decided to collaborate on developing an online training platform: MercuryLearn. The main component is the UNEP Toolkit for Identification and Quantification of Mercury Releases.
- This initiative has been funded by the European Commission and the government of Switzerland.



► <https://mercurylearn.unitar.org/>

► Online training modules on the UNEP's Toolkit for identification and quantification of mercury releases Inventory Level 1 and 2

► Self-paced, available in English and Spanish



Inventory Level 1

- Self-paced
- 10-15 hours
- Introductory video
- How to access
- English
- Free course



Inventory Level 2

- Self-paced
- 10-15 hours
- Introductory video
- How to access
- English
- Free course



Nivel 1 del inventario

- A su propio ritmo
- 10-15 horas
- Vídeo de introducción
- Cómo acceder
- Español
- Curso gratis



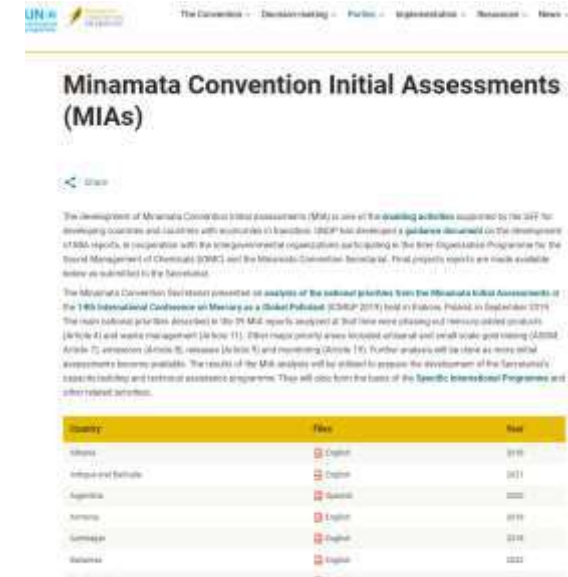
Nivel 2 del inventario

- A su propio ritmo
- 10-15 horas
- Vídeo de introducción
- Cómo acceder
- Español
- Curso gratis

Minamata Convention Initial Assessments






- GEF enabling activities include the development of Minamata Convention Initial Assessments (MIA), which support countries to prepare to implement the obligations of the Minamata Convention as soon as possible.
- MIA may include:
 - National Mercury Profile, including identification of significant sources of emissions and releases
 - Overview of structures, institutions, and legislation already available to implement the Convention;
 - Challenges to implementation, including identification of legal and/or regulatory gaps to be addressed prior to ratification
 - Capacity building, technical assistance as well as other needs required for the implementation of the Convention.
- MIA reports are available on [website](#).

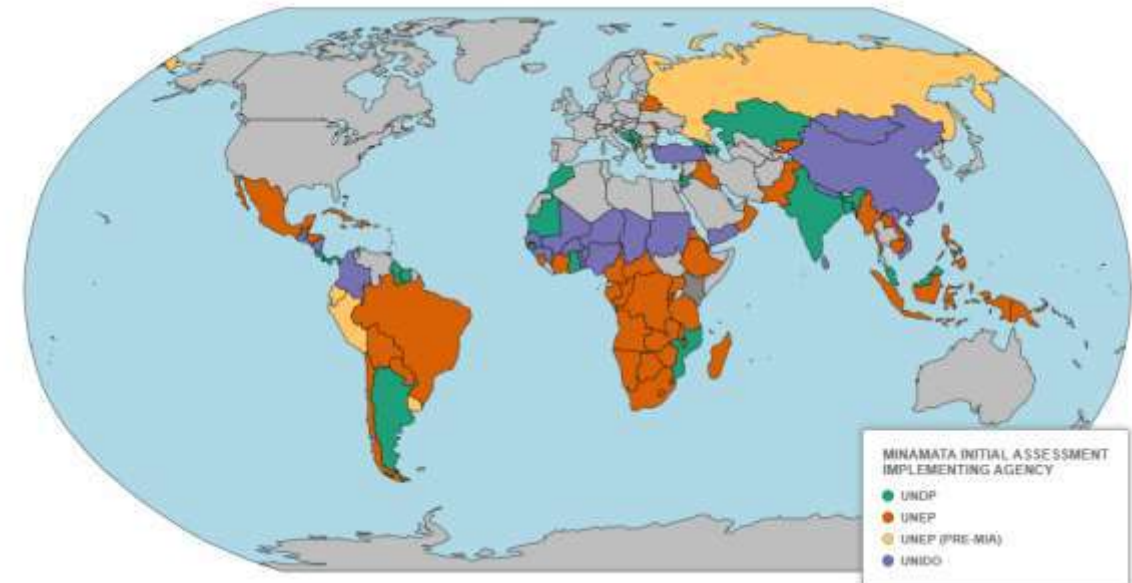
See [Convention website](#)



The development of Minamata Convention Initial Assessments (MIAs) is one of the enabling activities supported by the GEF for developing countries and countries with economies in transition. UNEP has developed a guidance document on the development of MIA-Reports, in cooperation with the intergovernmental organizations participating in the Inter-Departmental Programme for the Sound Management of Chemicals (IDPC) and the Minamata Convention Technical Secretariat. Final reports are to be made available online as submitted to the Secretariat.

The Minamata Convention Secretariat presented an analysis of the national priorities from the Minamata Initial Assessments of the 19th International Conference of Ministers of Health on a Global Pollution Strategy (2018) held in Finland in September 2019. The main national priorities identified in the 29 MIA reports included: air pollution, hazardous waste management (Article 11), other major priority areas included chemical and small scale gold mining (Article 7), emissions of toxic substances (Article 9) and monitoring (Article 10). Further analysis will be done as more initial assessments become available. The results of the MIA analysis will be utilized to prepare the development of the Secretariat's capacity building and technical assistance programme. This will also form the basis of the Specific International Programmes and other related activities.

Country	File	Year
Indonesia	 English	2019
Indonesia and Bahrain	 English	2021
Argentina	 Spanish	2022
Kenya	 English	2019
Guatemala	 English	2019
Maldives	 English	2022



Minamata Convention Initial Assessments

MIA Mercury Inventory Dashboard by [Mark Burton](#)



UN environment programme **gef**

This dashboard allows you to explore data from national mercury inventories made with the UNEP Toolkit as part of Minamata Initial Assessments (MIA). Data are aggregated by region and country names are hidden for confidentiality.
[Click for more information on UNEP Toolkit](#)
[Click for more information on MIAs](#)

Global

78 Countries

4,016,788 kg Hg yr⁻¹

Primary metal production
Consumer products
Waste deposition/landfill/wastewater

28 Level 1 Toolkits
50 Level 2 Toolkits

For more information on how mercury inputs vary, please explore the following metrics:

[Sub-category](#) [Income & Development](#)

[Pathways](#)

Minamata Initial Assessments Mercury Inventories

Summary of results from UNEP's *Toolkit for identification and quantification of mercury releases*

✓ Select a region using the regional drop down menu to update the Hg Inputs in each figure and statistic.

Global



Hg input by country (kg Hg yr⁻¹)

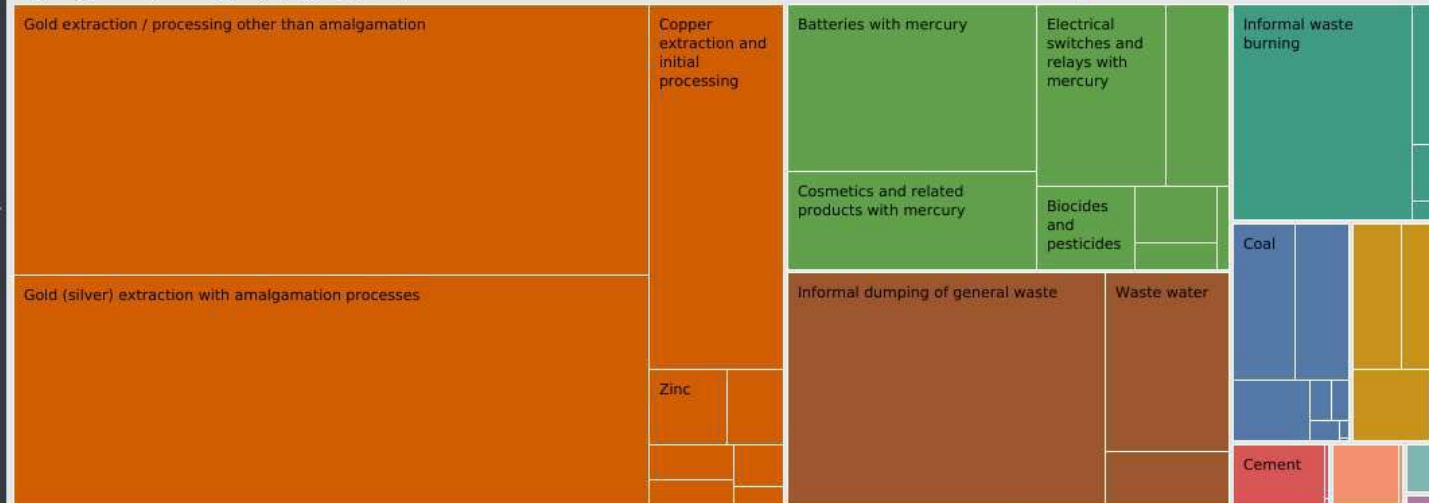


per capita Hg input by country (kg Hg yr⁻¹ 100,000 people⁻¹)



© 2024 Mapbox © OpenStreetMap

Hg Inputs by Category (kg Hg yr⁻¹)



- 5.1-Fuels/energy sources
- 5.2-Primary metal production
- 5.3-Other minerals and materials
- 5.4-Industrial processes
- 5.5-Consumer products
- 5.6-Other product/process use
- 5.7-Production of recycled metals
- 5.8-Waste incineration/burning
- 5.9-Waste deposition/landfill/wastewater
- 5.10-Crematoria and cemeteries

[View on Tableau Public](#)



Minamata (training) Tools



The screenshot shows the website for the United Nations System Staff College. At the top left is the college's logo. The navigation menu includes "Courses", "Customized Services", "Campuses", "Latest", "In Focus", and "About". On the right, there are search, shopping cart, and "Log in" buttons. The main banner is blue and white, featuring the text "ONLINE", "13 MAY 2024 - 31 DEC 2024", and "Minamata Tools". Below this, there are two tags: "CLIMATE CHANGE" and "SUSTAINABLE DEVELOPMENT AND THE SDGS", and a "Sign up" button. The background of the banner shows a large, dark, metallic-looking sphere and several smaller, lighter spheres. Below the banner is a white bar with eight icons and their corresponding details: LANGUAGE (English), DURATION (8 hours), ENROLL BY (31 Dec 2024), PRICE (Free), LOCATION (ONLINE), TARGET (Everyone), CONTACT US (by email), and FAQ (Read more).



Developed with the generous support of the European Union as part of project "Support to the capacity-building and technical assistance programme of the Secretariat of the Minamata Convention on Mercury"



UNITED NATIONS
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Minamata Convention on Mercury

Learning Path for National Focal Points

Self-paced

► <https://www.unssc.org/courses/minamata-tools-0>


Minamata Online series of virtual webinars on various topics related to the Minamata Convention on mercury since 2020

Support to Parties through the Minamata Convention Financial Mechanism

< Share

MINAMATA ONLINE | VIRTUAL | 9 - 9 MAY 2024

This session of Minamata Online is designed to give Parties and other interested stakeholders an understanding of the support currently available to eligible Parties through the Minamata Convention's financial mechanism, including the just-launched Fourth Round of applications to the Specific International Programme, as well as new materials available through training platforms.



SUPPORT TO PARTIES THROUGH THE MINAMATA CONVENTION FINANCIAL MECHANISM

MINAMATA ONLINE SEASON 3 / 2023

16 OCTOBER 2023
14:00 - 15:00 CEST
REGISTER NOW
BIT.LY/MO16OCT23

List of speakers:

- Marianne Bailey, Minamata Convention on Mercury Secretariat
- Maria Irene Rizzo, Minamata Convention on Mercury Secretariat
- Kevin Helps, UN Environment Programme
- Talita De Melo Pinotti, United Nations System Staff College


Time and place

Takes place in
Virtual

Starts on
9 MAY 2024
2:00 PM CET

Ends on
9 MAY 2024
3:00 PM CET

https://minamataconvention.org/en/meetings/upcoming-list-view?field_event_type_target_id=287



BRIEFING FOR NEW PARTIES

MINAMATA CONVENTION ON MERCURY

16 OCTOBER 2023
14:00 - 15:00 CEST
REGISTER NOW
BIT.LY/MO16OCT23

MINAMATA ONLINE SEASON 3 / 2023

Time and place

Takes place in
Virtual

Starts on
16 OCT 2023
2:30 PM CEST

Ends on
16 OCT 2023
3:30 PM CEST

[Add to Calendar](#)

Links

- » Register now to the event
- » Flyer of the event

Resources:

- Full presentations of the session
- Video recording of the event below



INFORMATION SESSION ON NATIONAL REPORTING

MINAMATA CONVENTION ON MERCURY

8 SEPTEMBER, 2023
14:00 - 14:45
REGISTER NOW
BIT.LY/MO08SEP23

MINAMATA ONLINE SEASON 3 / 2023

Time and place

Takes place in
Virtual

Starts on
8 SEP 2023
2:00 PM CEST

Ends on
8 SEP 2023
2:45 PM CEST

[Add to Calendar](#)

Links

- » Full presentations
- » Video recording of the event
- » Save the date for upcoming events



MINAMATA
CONVENTION
ON MERCURY

Thank you for your attention

Secretariat of the Minamata Convention on Mercury
United Nations Environment Programme
11-13, Chemin des Anémones - 1219 Châtelaine, Switzerland

WEB: <https://minamataconvention.org/>
MAIL: MEA-MinamataSecretariat@un.org
TWITTER: [@minamataMEA](https://twitter.com/minamataMEA)
[#MakeMercuryHistory](https://twitter.com/minamataMEA)

Using Inventory Data and Planned Policies to Inform Future Emission Scenarios in South Africa

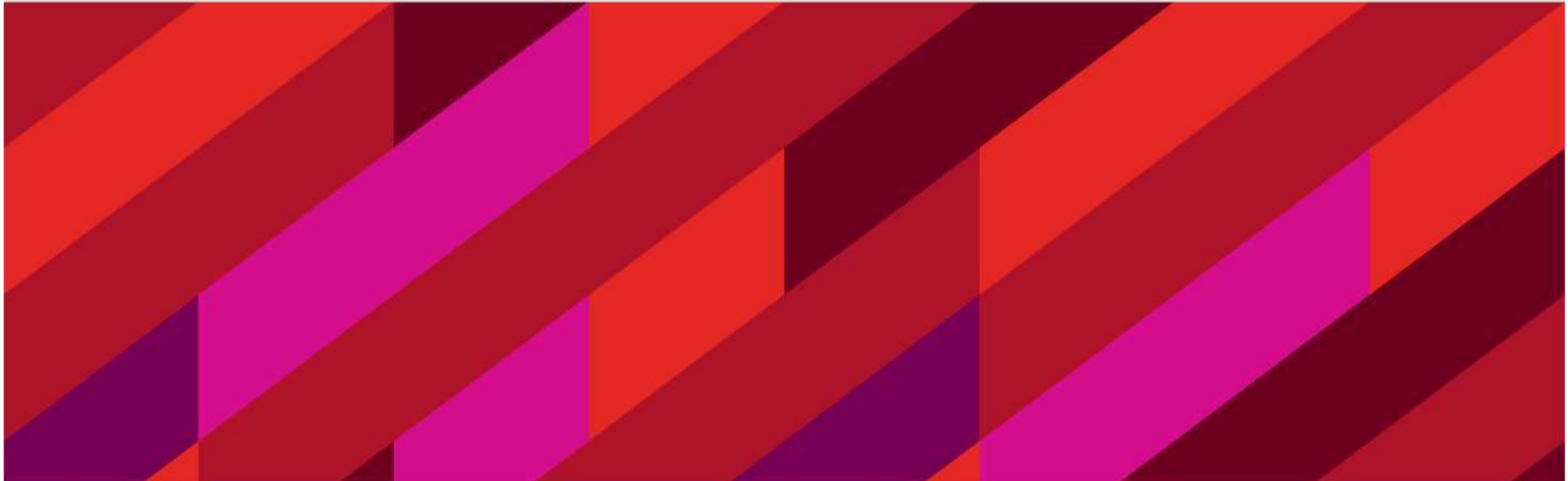


MACQUARIE
University
SYDNEY · AUSTRALIA

EDWARD ARCHER

One-day working event on inventory production and compliance strategies for the South African
Coal fleet under the Minamata Convention

30 May 2024



Project Outcomes



OUTCOME 1: Comprehensive coal sectoral analysis

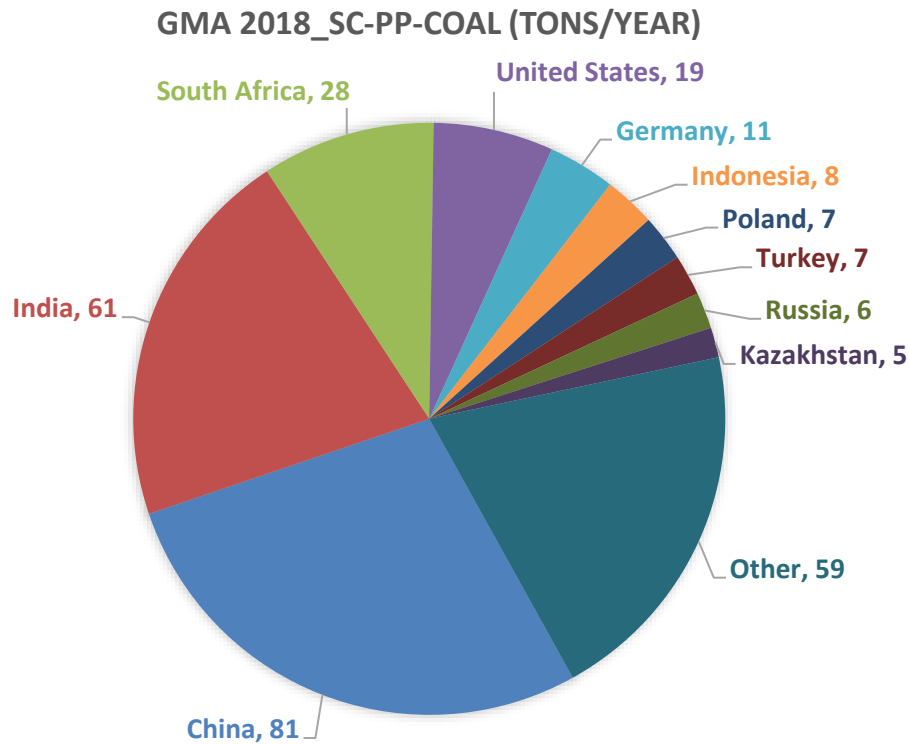
Activities

- Review scientific data on mercury emissions from CFPPs
- Evaluate the impact of commitments and targets by UN Conventions on Hg/GHG/POP emissions from the coal sector
- Potential mercury reduction figures & scenarios from CFPPs produced
 - Expand to Coal-Fired Industrial Boilers (CFIB)

Global Mercury Assessment 2018

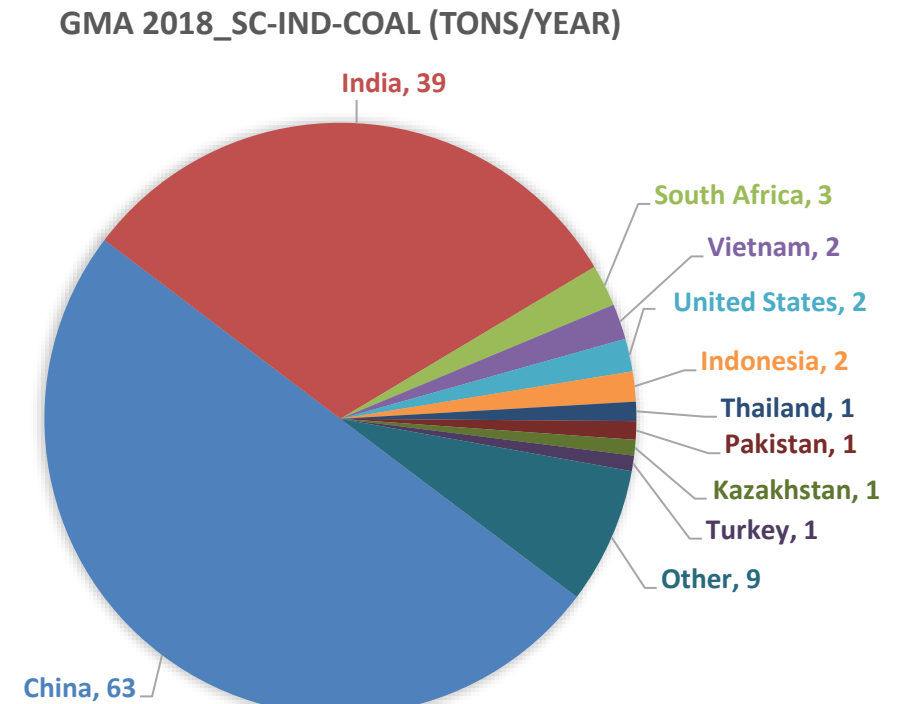
SOUTH AFRICA IN THE GLOBAL CONTEXT

Stationary Combustion of Coal at Power Plants 292 tons/year



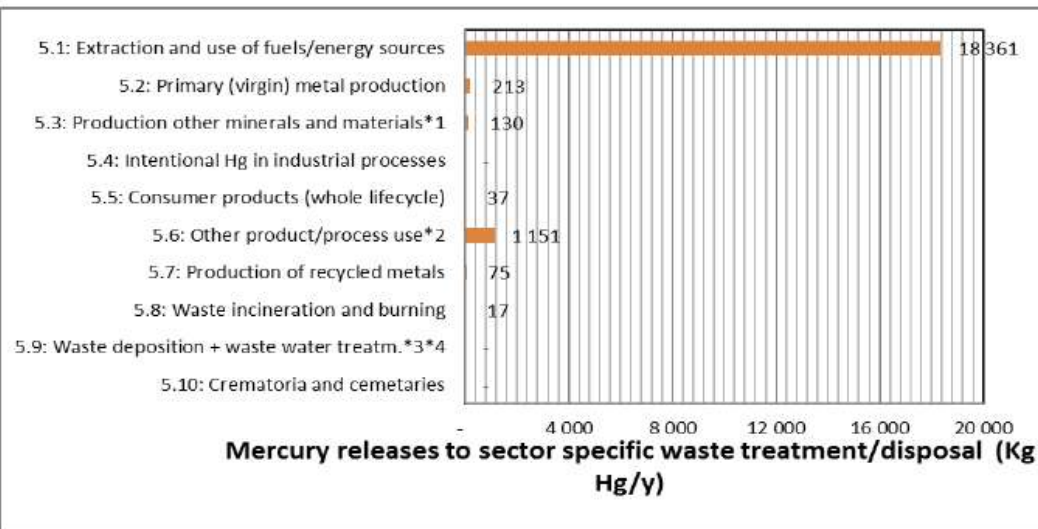
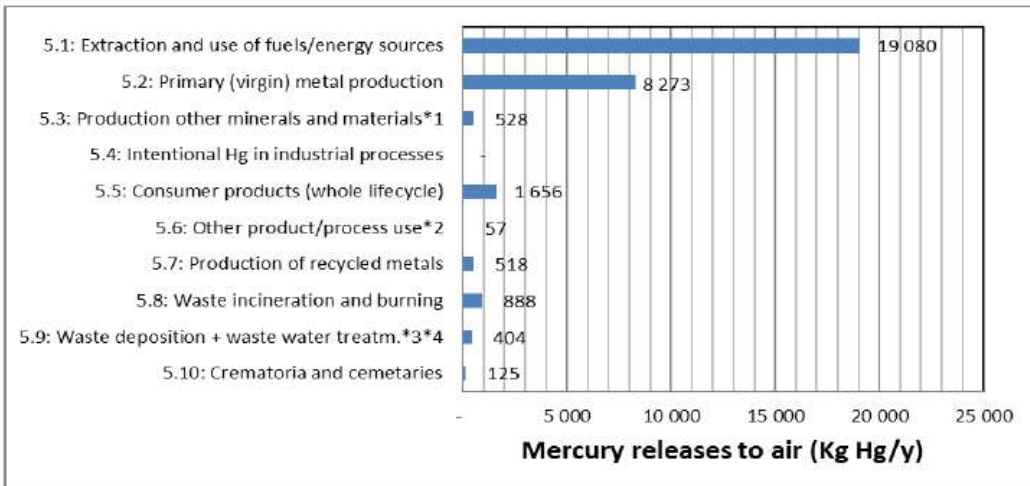
China, India & South Africa = 47% - 59% global coverage

Stationary Combustion of Coal at Industrial Boilers 126 tons/year



China & India = 73% - 83% global coverage

Minamata Initial Assessment 2019



CFPPs:

Coal consumption:

- 77 million tonnes (washed/unwashed)

Input factor:

- 0,13-0,24 mg/kg (bituminous)
- 0,15 mg/kg (washed anthracite)
- 0,105 mg/kg (washed coal)

Emission to air:

- 18.096 tonnes / year

Important consideration - The mass balance of mercury in CFPPs

(input-retention-emission-release)

Methodology – CFPP emissions



BASELINE DATA FROM THE GLOBAL ENERGY MONITOR

<https://globalenergymonitor.org/projects/global-coal-plant-tracker/>

<https://globalenergymonitor.org/projects/global-coal-plant-tracker/methodology/>

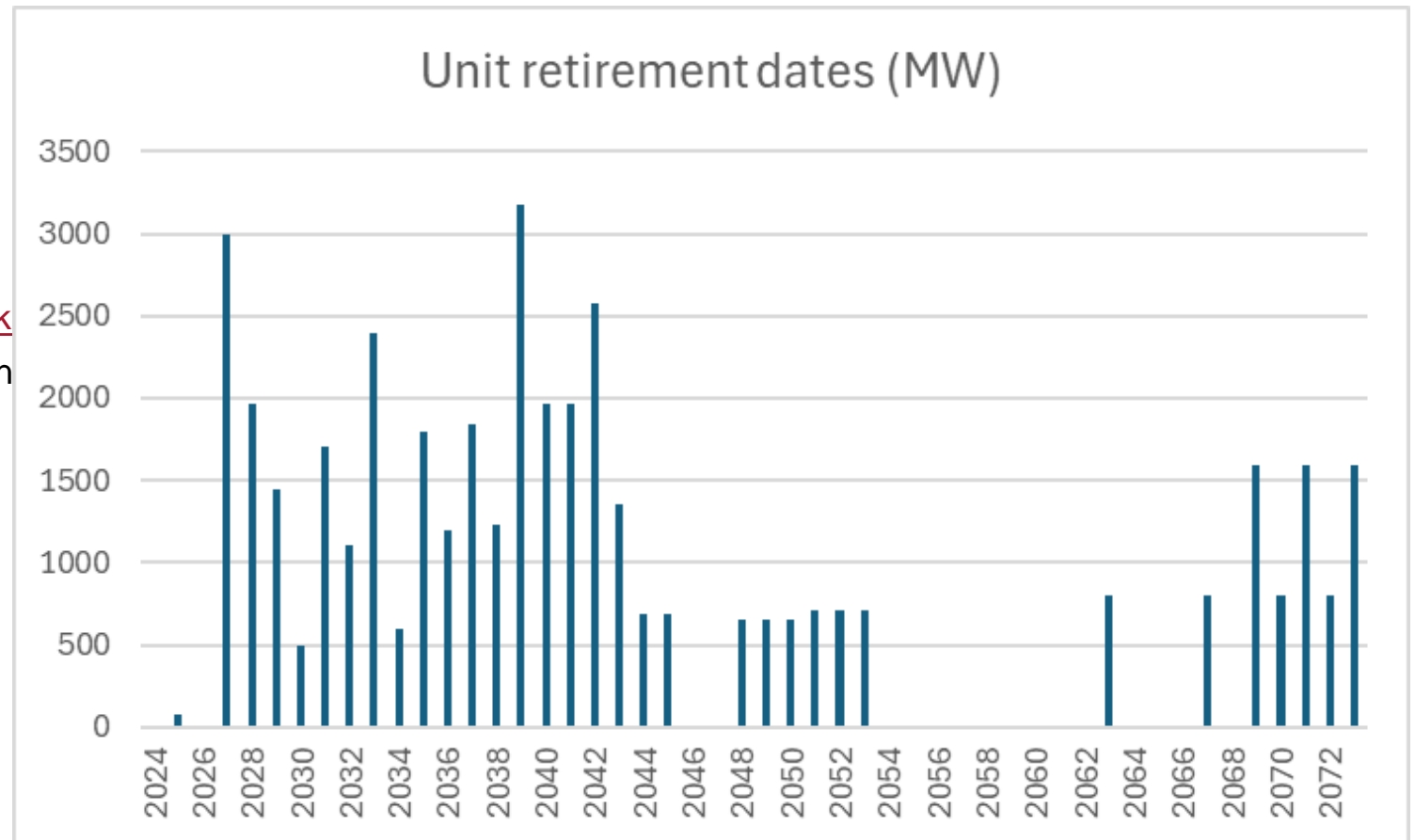
Country- & Unit-level information

- Capacity (MW)
- Start/Planned retirement year
- Combustion technology
- Coal type
- Heat rate (Btu/kWh) - <https://www.gem.wik>
- Capacity factor - Global average from Intern
- Remaining plant lifetime

E.g., Heat Rate (Btu per kWh) –
South Africa units

Low – 8,409

High – 12,618



Methodology – CFPP emissions



BASELINE DATA FROM THE GLOBAL ENERGY MONITOR

Assumptions/uncertainties

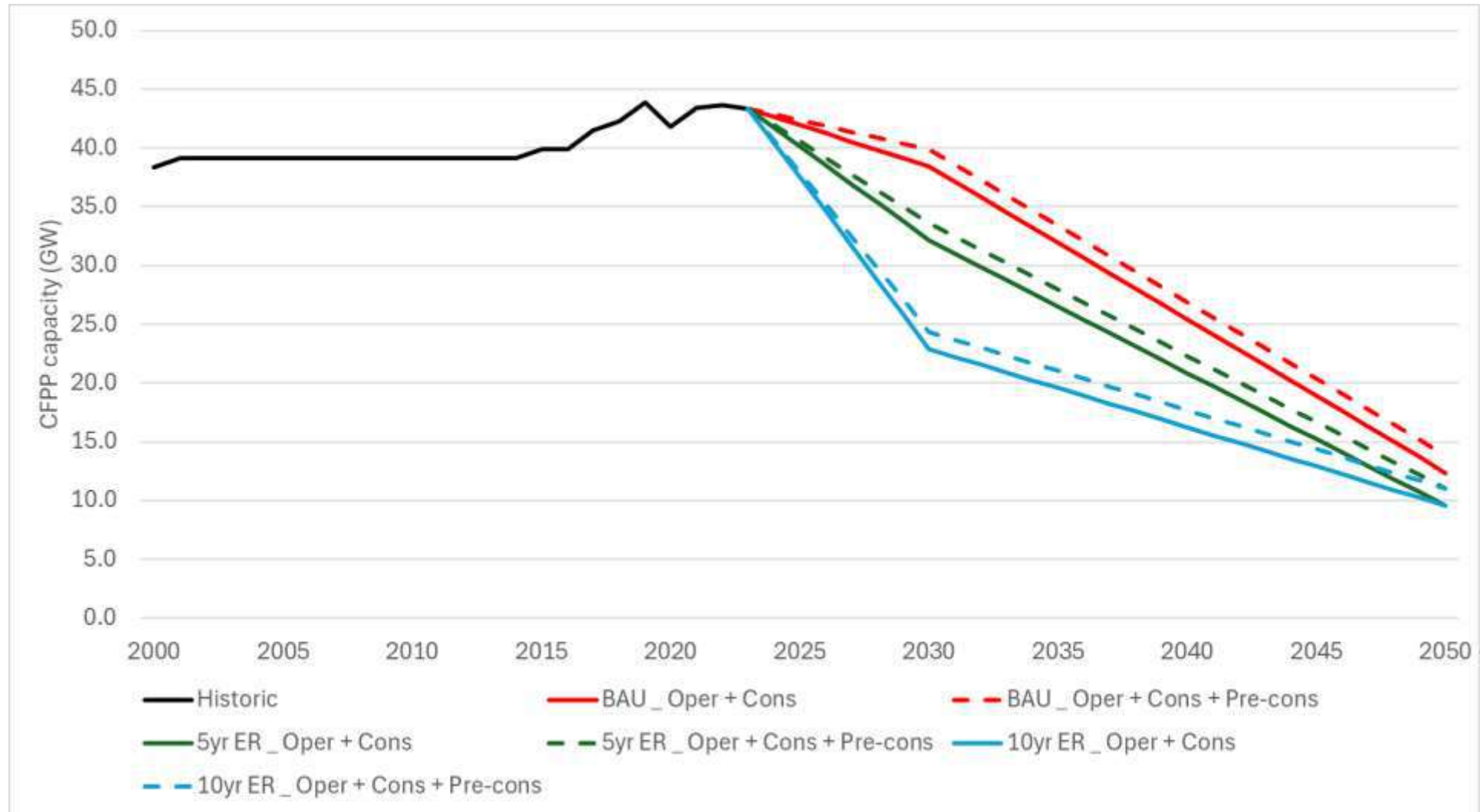
- Default 40-year plant life expectancy (SA plants operating for >40yrs)
- New project start year (where not indicated) – operational by 2030
- Mercury emissions
 - Defined APCD configurations (Garnham & Langerman, 2016, Clean Coal Journal, Vol 26, No 2)
 - Unit-level capacity factors
 - Unit-level GCV (kJ/kg coal) – average levels per coal type based on Annex 28 of the Stockholm Convention Toolkit

GCV (kJ/kg coal)	Av
Bituminous	29300
Subbituminous	14500
Anthracite	30667
Lignite	8583
Unknown	25000
Waste coal	25000

Stockholm Convention
Annex 28 averages

- Mercury coal input factor – 0,23 mg/kg - Wagner and Hlatshwayo 2005, Int J Coal Geol 63:228–246; Tewalt et al. 2010, Open-File Report 2010–1196. United States Geological Survey, Reston; Bergh et al. 2011, Fuel Process Technol 92:812–816.
- Limitations: Additions of mercury-specific controls, Br additions, coal washing, Hg speciation, Cl content, coal blending/co-firing

CFPP capacity outlook



Methodology – Mercury Emissions Estimate (UNEP toolkit)



$$\text{Mercury emission (kg/year)} = \text{Coal consumption} * \text{IF} * ((100 - \text{RF}) / 100)$$

HRV / GCV * CAP * CF * 9.24E03
 South African CFPPs = 76,740,000 tonnes / year

Mercury input factor by country (mg/kg) - USGS default		
China	0,17	Liu et al., 2019
India	0,22	India country profile
Indonesia	0,06	BCRC-SEA, 2017
Vietnam	0,28	UNEP, 2017
Philippines	0,08	USGS
Thailand	0,14	USGS
Malaysia	0,08	USGS
South Africa	0,21	https://link.springer.com
REMAINING WORLD	0,15	USGS
Australia	0,08	USGS
United States	0,13	https://pubs.usgs.gov

Table 5-11 Mercury retention rates and application profile developed by UNEP/AMAP (2012).

	Intermediate mercury retention rates, %, by coal type		Degree of application (%) by country group *1				
	Hard coal (anthracite, bituminous)	Brown coal (sub-bituminous, lignite)	1	2	3	4	5
Air pollution controls							
Industrial use (combustion):							
Level 0: None	0.0	0.0			25	50	75
Level 1: Particulate matter simple APC: ESP/PS/CYC	25.0	5.0	25	25	50	50	25
Level 2: Particulate matter (FF)	50.0	50.0	25	50	25		
Level 3: Efficient APC: PM+SDA/wFGD	50.0	30.0	25	25			
Level 4: Very efficient APC: PM+FGD+SCR	90.0	20.0	25				
Level 5: Mercury specific	97.0	75.0					
Other coal combustion:							
Level 0: None	0.0	0.0	50	50	100	100	100
Level 1: Particulate matter simple APC: ESP/PS/CYC	25.0	5.0	50	50			

Energy Action Plan (2022)

Facilitated by the National Energy Crisis Committee (NECOM)

Actions:

1. *Improve availability of existing supply*

- Reduce unplanned outages & increased generation from renewables
- Debt relief package from the National Treasury - investment in necessary maintenance, diesel supplies, OCGT load factor increase & expand transmission networks
- Eskom Generation Recovery Plan – independent technical review
- Return Kusile & Medupi units to service
- Distribution Demand Management Programme – energy savings incentives

2. *Accelerate private investment in generation capacity*

- E.g., Resource Mobilisation Fund (RMF) – technical support
- E.g., Energy Council of South Africa – engineering support
- Electricity Regulation Act, Schedule 2, amendments – remove licensing thresholds for generation facilities
- Reduced time frame for regulatory approvals by energy projects
- Invest SA – applications for renewable energy projects for authorizations
- Eskom-leased land for developers of private energy projects – Phase 1 where transmission infrastructure is already available
- Power purchasing mechanism from private sector - Standard Offer Programme & Emergency Generation Programme

Energy Action Plan (2022)

Facilitated by the National Energy Crisis Committee (NECOM)

Actions:

3. *Fast-track procurement of new generation capacity from non-fossil fuels*

- 14 GW of new wind/solar/battery storage procurement
- Three projects from the Risk Mitigation Programme in construction
- Power Purchase Agreements for 19 projects & additional new capacity construction – 2,300 MW
- Import power from neighboring countries, subject to transmission networks

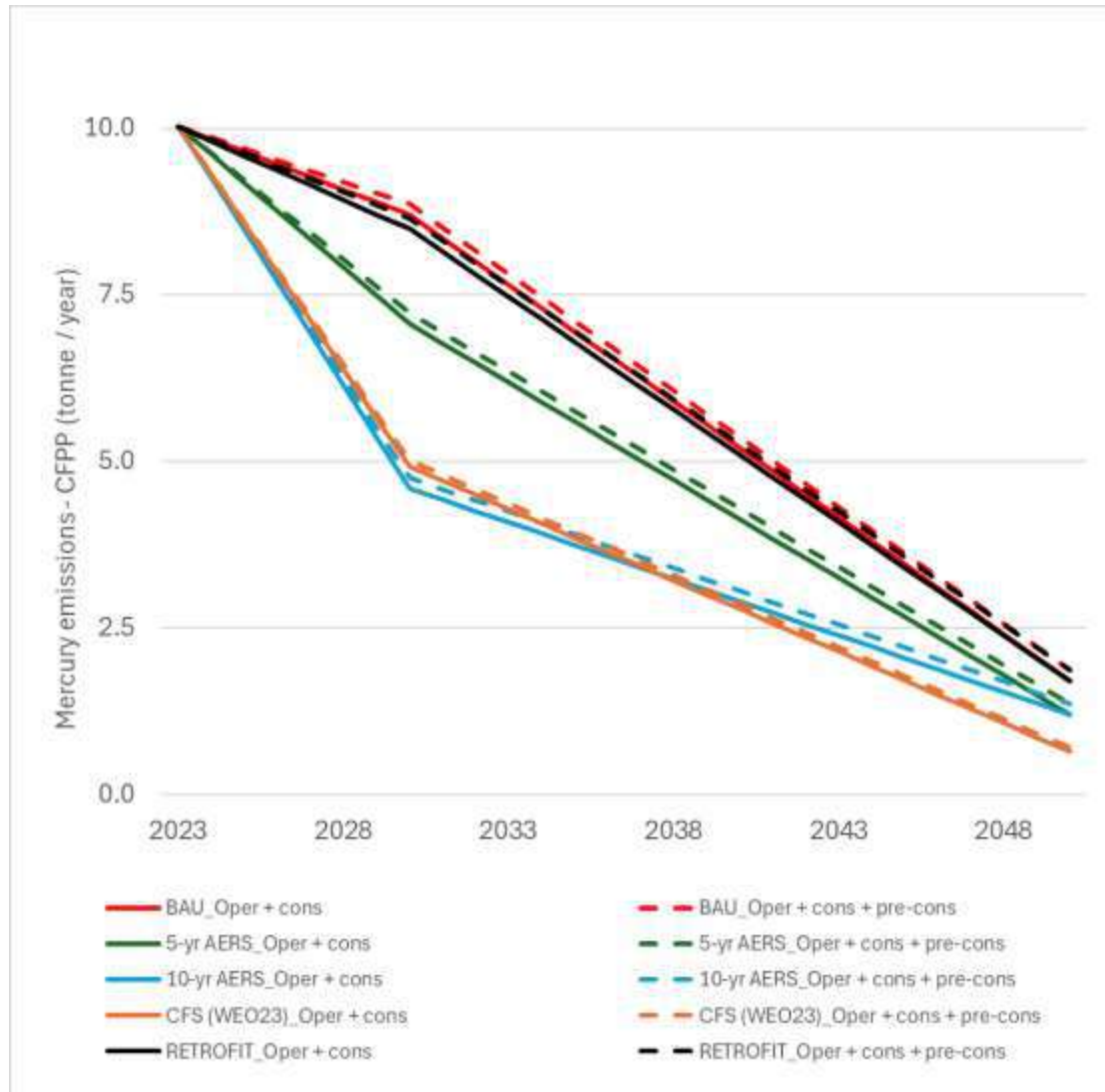
4. *Accelerate investment in rooftop solar (businesses & households)*

- Special tax incentives for businesses & households installing solar
- Bounce-back loan scheme for small businesses going solar
- Progress in rooftop solar installments across the country

5. *Fundamental transformation of electricity sector for long-term energy security*

- National Transmission Company of South Africa as independent entity for managing the national electricity grid (improved private sector participation)
- New legislation for a competitive electricity market (i.e., Electricity Regulation Amendment Bill)

CFPP Mercury Emissions



BAU – Business as Usual

AERS – Early Retirement

- Subcritical CFPPs
- 5-yr/10-yr early retirement

CFS (Capacity factor scenario)

- 2024 – 0.53
- 2030 – 0.3
- 2050 – 0.2

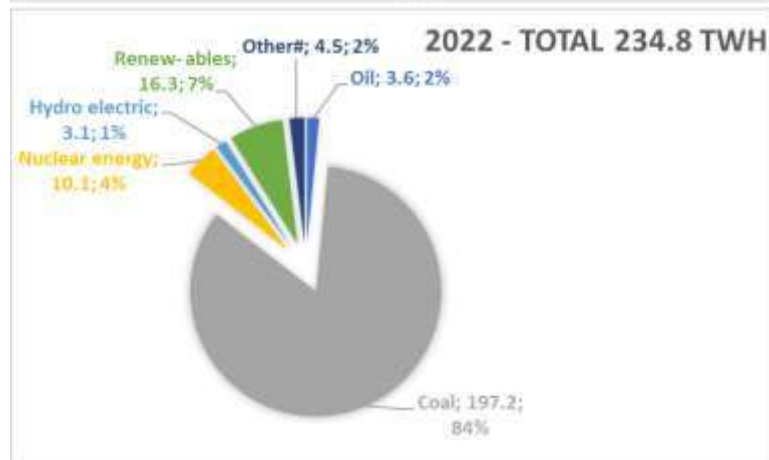
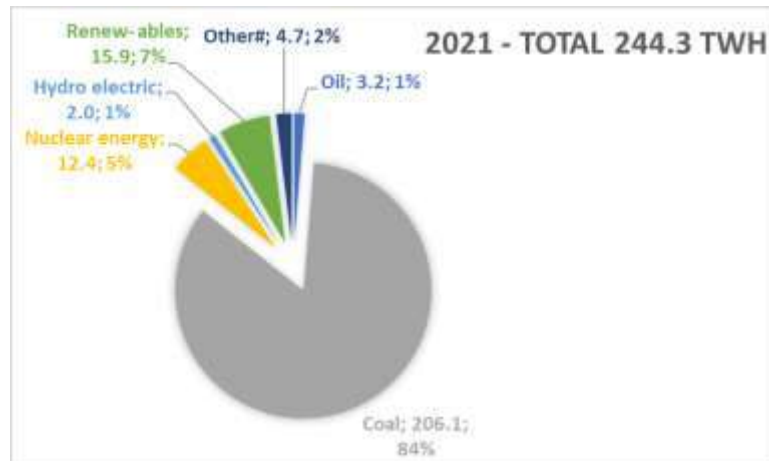
RETROFIT scenario

RETROFIT scenario criteria	
Remaining lifetime	20
Original APCD configuration	FF
New APCD configuration	FF + FGD
Unit status to retrofit	Operating
Retrofit by	2030

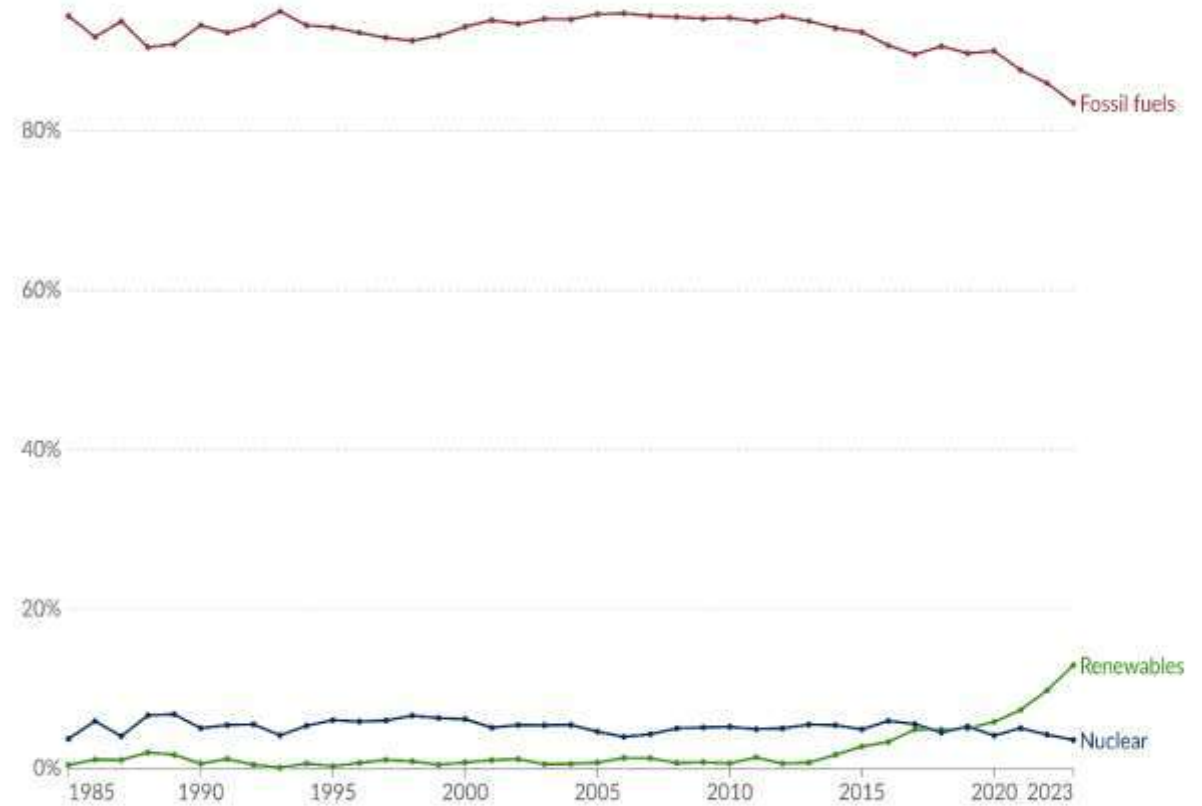
Reduction in CFPP emissions reliant on alternative energy developments



Energy Institute – Statistical Review of World Energy



Share of electricity generation from fossil fuels, renewables and nuclear, South Africa



Data source: Ember (2024); Energy Institute - Statistical Review of World Energy (2023)

OurWorldInData.org/energy | CC BY

South Africa NDC 2020/21



“The key challenge during the implementation periods of this first NDC (2021 to 2025, and 2026 to 2030) will be the transition in the electricity sector, seeking early investment in and preparing for mitigation in harder-to-mitigate sectors, and addressing the **economic and social consequences** resulting from this transition in coal-producing areas.”

“... developing labour and social plans as and when ageing coal-fired power plants and associated coal production infrastructure are decommissioned.”

“Over the next decade, the NDC will require a much greater investment programme, as specified in IPR 2019, of between R860 billion and R920 billion (in 2019 Rands; USD60-64 billion). The shift away from coal that IPR 2019 requires, will require support in the form of transition finance, and associated technology and capacity-building.”

Draft IRP2023

Horizon 1 (2023 – 2030)



Table 2: Emerging Plan from Horizon One Analysis

	Coal	Gas – IPP Programme	Gas - Eskom	Dispatchable Capacity	Nuclear	Hydro	Pumped Storage	CSP	Solar PV	Wind	Hybrid IPP Programme	Distributed Generation*	BESS – IPP Programme	BESS - Eskom	Unserviced Energy (TWh)
Current Base (MW)	38 800	1 005	2 825	-	1 860	1 600	2732	500	2 287	3 443	-	5 000	-	20	
2024	720							100			150	900		199	13.06
2025	720	1 220							2 115	644	476	900	513	141	7.63
2026										140		900			7.66
2027		1 000								684		900	2 000	615	4.55
2028		1 000	3 000						500			900	615		0.22
2029									500	1 500		900			0.25
2030		1 000		1 376					500	1 500		900			0.27
Additional New Capacity (MW)	1 440	4 220	3 000	1 376				100	3 615	4 468	626	6 300	3 743	360	

	Installed Capacity
	Capacity under construction
	Capacity procured
	New Capacity
	Distributed Generation Capacity for own use
	Unserviced Energy, preferred as low as possible

5.2.1. Proposed Interventions

Intervention 1: As already identified and in progress as part of the Energy Action Plan interventions, the improvement of Eskom fleet EAF as per the Generation Recovery Plan is crucial and will make a significant contribution in restoring security of supply.

Intervention 2: In addition to non-dispatchable supply initiatives (business plus the State), the deployment of dispatchable generation options such as gas to power in line with Section 34 Ministerial Determinations must be accelerated as they will address the unserved energy risk and can be adapted to the power system requirements in a relatively short time.

Intervention 3: Where technically and commercially feasible, delay shutting down coal fired power plants to retain dispatchable capacity.

Intervention 4: Support and enable the development of the transmission grid as per the TDP 2023-2032 to enable connection of additional generation capacity initiatives by the public and private sector,

Intervention 5: Manage the following emerging risks:

- Completion of Extension of the design life of Koeberg Power Station
Completion of the planned life extension of the Koeberg nuclear power station should proceed with the necessary speed to mitigate against the loss of dispatchable 1 800 MW.
- Compliance with Minimum Emissions Standards
Resolving the challenges around compliance with the implementation of the Minimum Emissions Standards (MES) on coal fired power stations in terms of the National Environmental Management: Air Quality Act 39 (2004) is critical as it will drastically ensure capacity totalling 16 000 MW immediately and up to 30 000 MW in April 2025 is retained.

Horizon 2 (2031 – 2050)

- ***Renewable & clean energy beneficial for decarbonising the energy system, not for security & supply***
 - Implement dispatchable technologies with high utilization factors.
- Large need for new capacity build programmes, including improved transmission networks
 - Need for technical analysis of power systems & regular adjustments of policies to ensure security of supply

Project Outcomes

Activities

- Synthesis of results from completed & ongoing CFPP projects
- Selection criteria: Future projects based on highest impact potential
 - Guidance on where to support large scale projects – Training/Capacity-Building
- Assist public and private sectors in their decision-making processes



OUTCOME 2:

**STRATEGY FOR THE COAL
SECTOR'S EMISSIONS
REDUCTION CONTRIBUTION
TO STOCKHOLM AND
MINAMATA CONVENTIONS**



MACQUARIE
University
SYDNEY · AUSTRALIA

Thank you

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